

van der Waals forces in density functional theory

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$2n+1$ theorem

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(Many-body) Hamiltonian

$$\hat{H} = -\frac{\hbar^2}{2m_e} \sum_i \nabla_i^2 - \sum_I \frac{\hbar^2}{2M_I} \nabla_I^2 + \sum_I \frac{Z_I e^2}{|\mathbf{r} - \mathbf{R}_I|} + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} + \frac{1}{2} \sum_{I \neq J} \frac{Z_I Z_J e^2}{|\mathbf{R}_I - \mathbf{R}_J|}$$

(Many-body) wave function

$$\Psi = \Psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_{N_e}; \mathbf{R}_1, \mathbf{R}_2, \dots, \mathbf{R}_{N_I})$$

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$$\Psi = \Psi_{\mathbf{R}_1, \mathbf{R}_2, \dots, \mathbf{R}_{N_I}}(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_{N_e})$$

Total energy in the Kohn-Sham density functional theory

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Total energy of the system

$$E_{\text{tot}}[n] = T_s[n] + \int d\mathbf{r} v_{\text{ext}}(\mathbf{r})n(\mathbf{r}) + \frac{e^2}{2} \iint d\mathbf{r} d\mathbf{r}' \frac{n(\mathbf{r})n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} + E_{\text{xc}}[n] + E_{\text{ion}}$$

Electron density

$$n(\mathbf{r}) = N_e \int \cdots \int d\mathbf{r}_2 \cdots d\mathbf{r}_{N_e} |\Psi_{\mathbf{R}, \mathbf{R}_2, \dots, \mathbf{R}_{N_e}}(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_{N_e})|^2$$

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$$E_{\text{xc}}^{\text{hyb}}[n] = aE_{\text{HF}} + (1 - a)E_{\text{x}}^{\text{sl}} + E_{\text{c}}^{\text{sl}}$$

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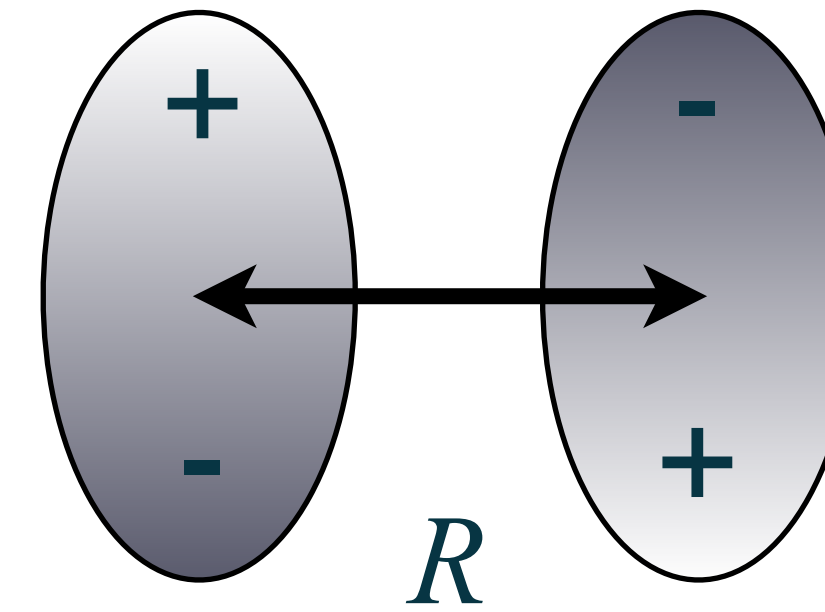
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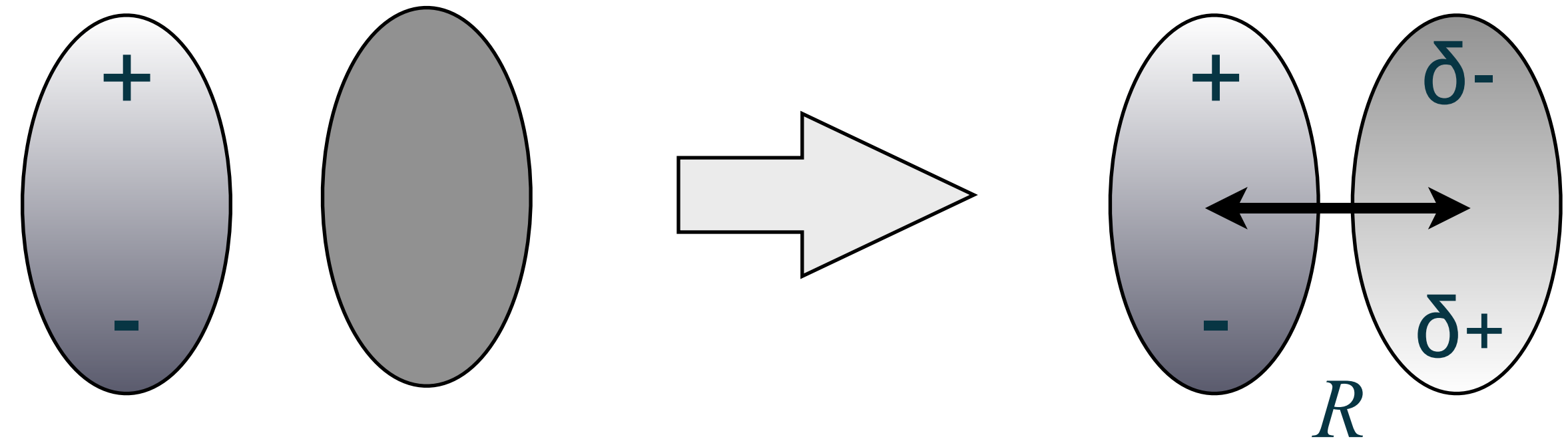
→ Semiempirical correction (DFT-Dx, TS-vdW, XDM, LRD, ...)

Van der Waals forces

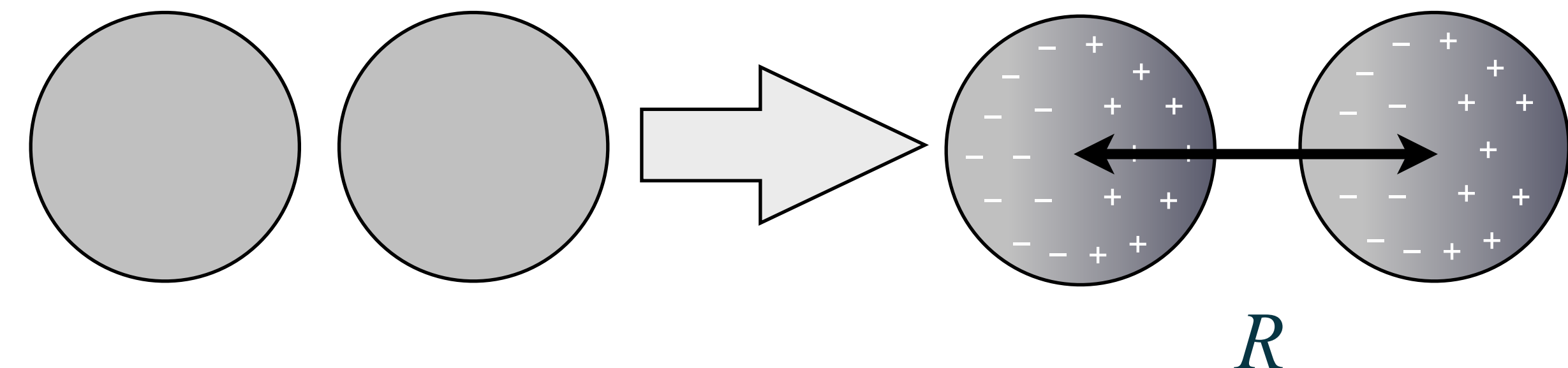
Dipole-dipole interaction
(Keesom force)



Dipole-induced dipole interaction
(Debye force)



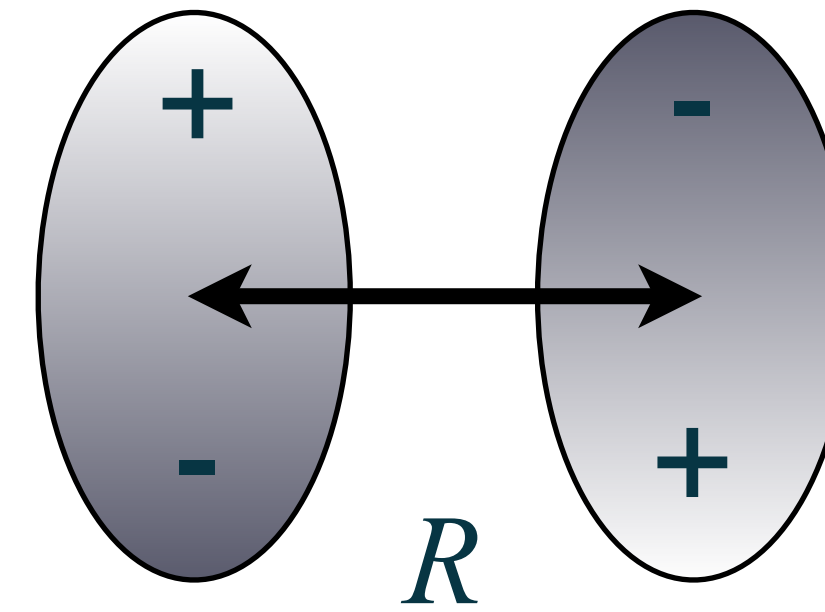
Induced dipole-induced dipole interaction
(London dispersion force)



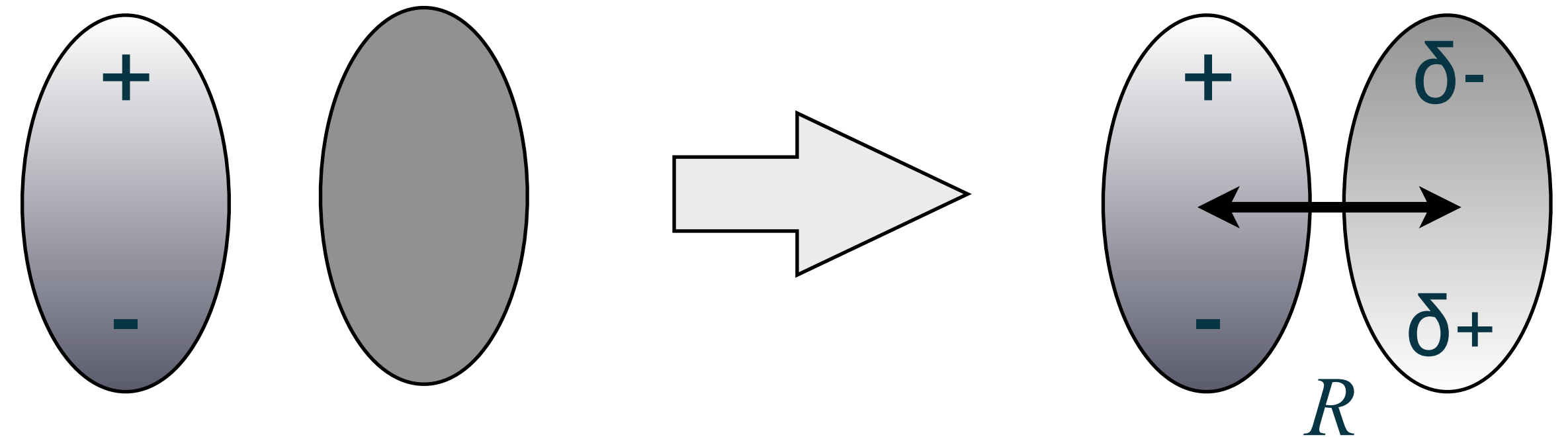
Van der Waals forces

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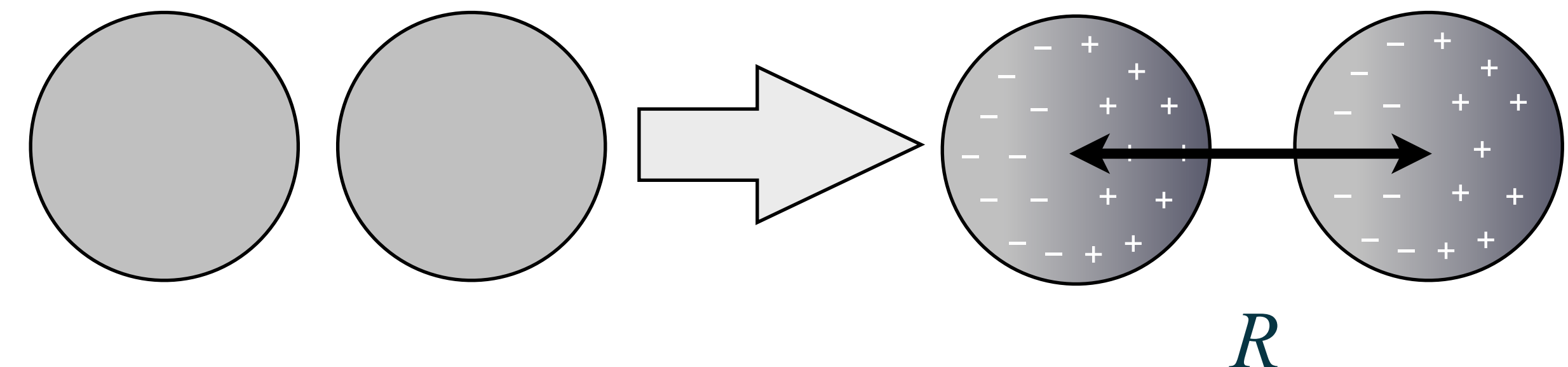
$$1/R^3$$



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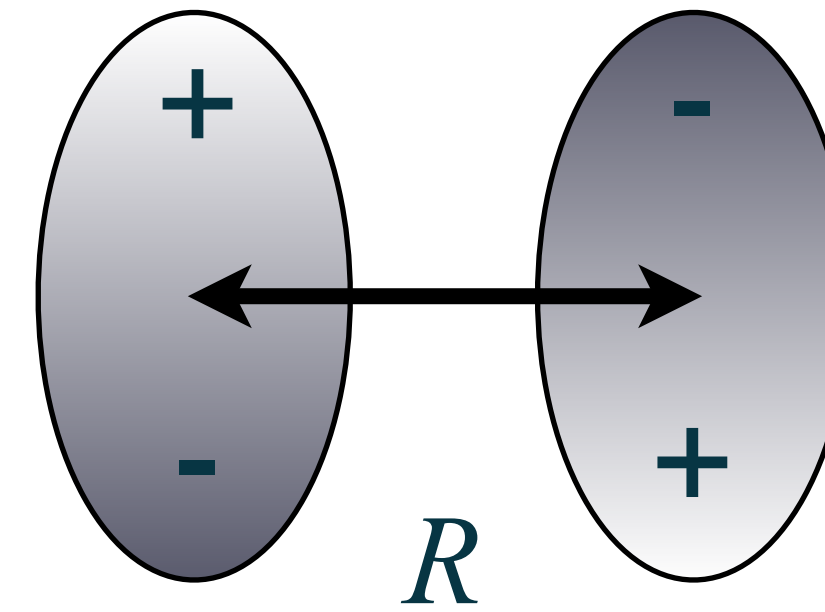
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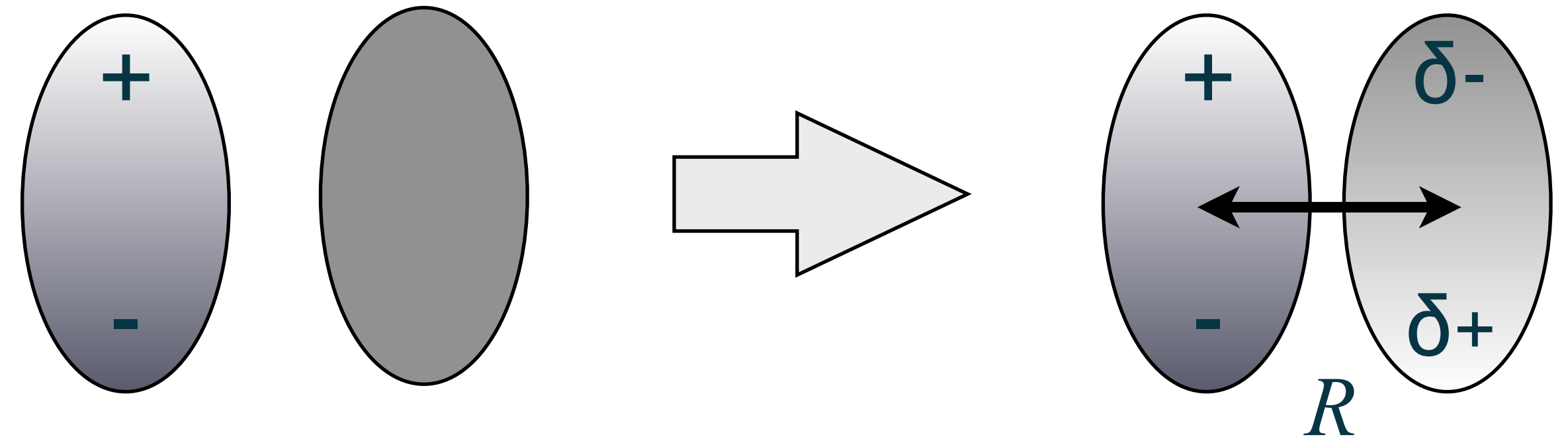
Van der Waals forces

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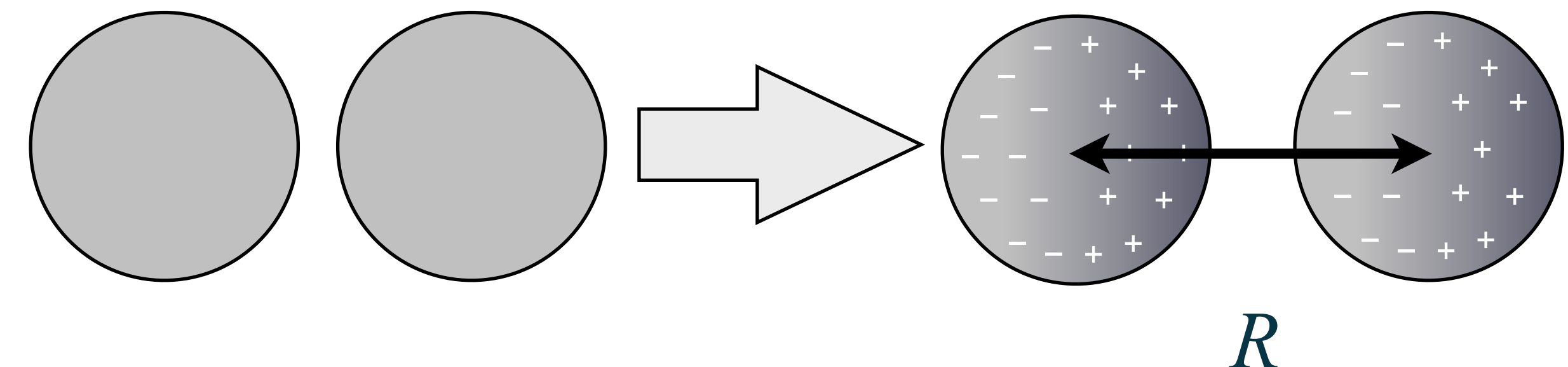
☺ LDA/GGA works



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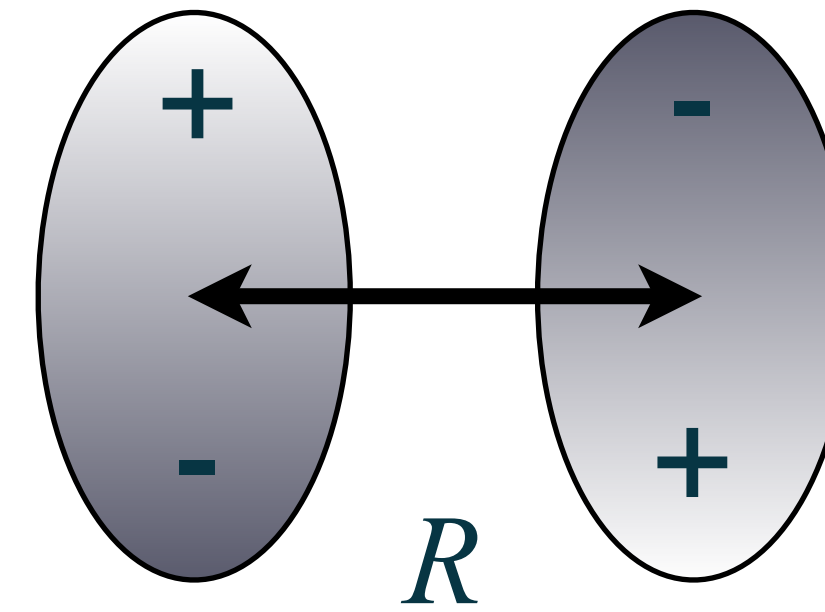
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Van der Waals forces

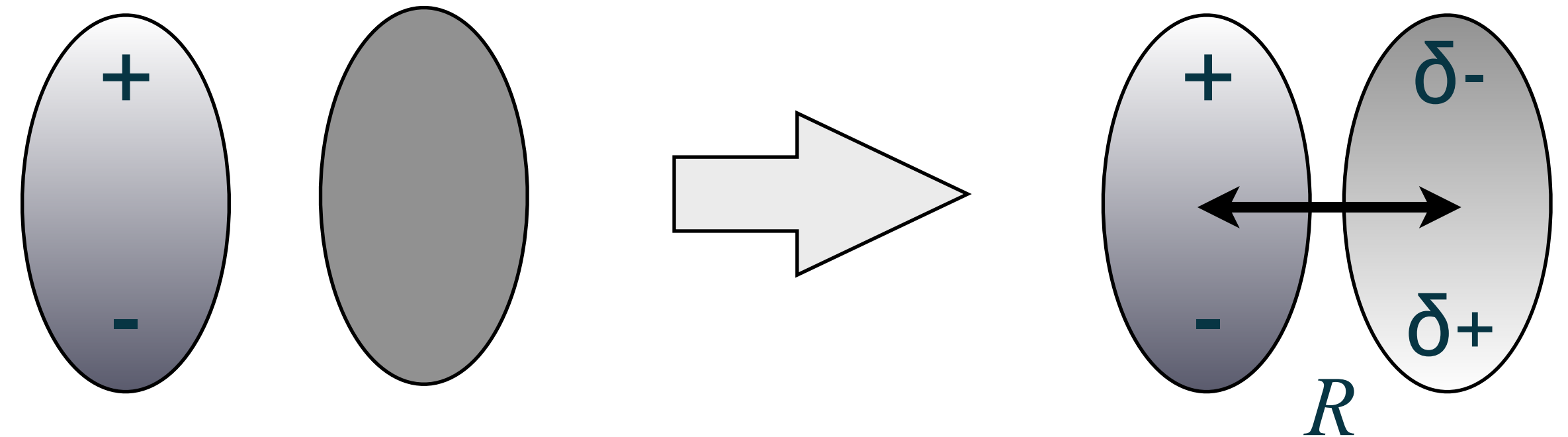
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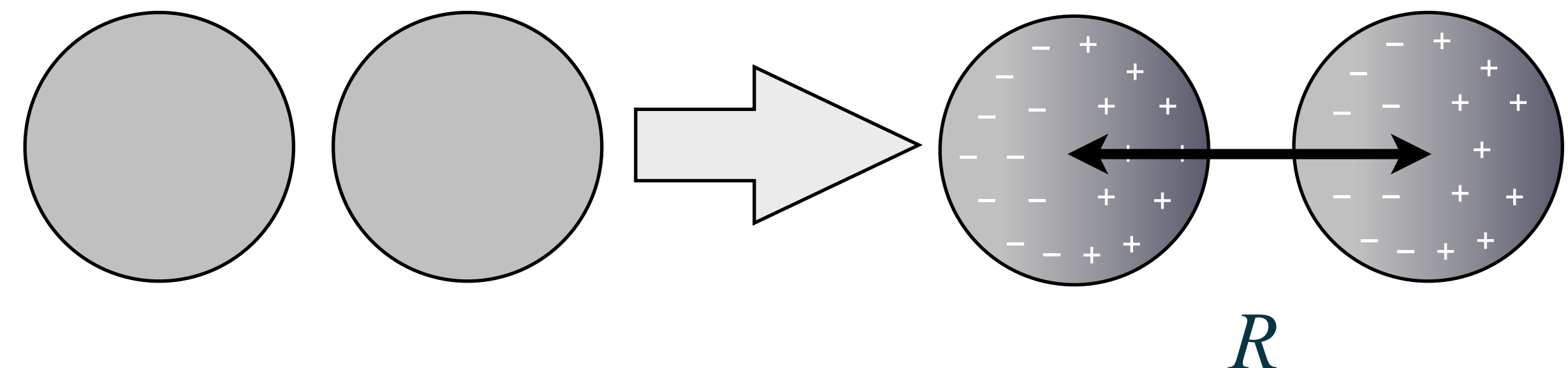


Dipole-induced dipole interaction
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$$1/R^3$$



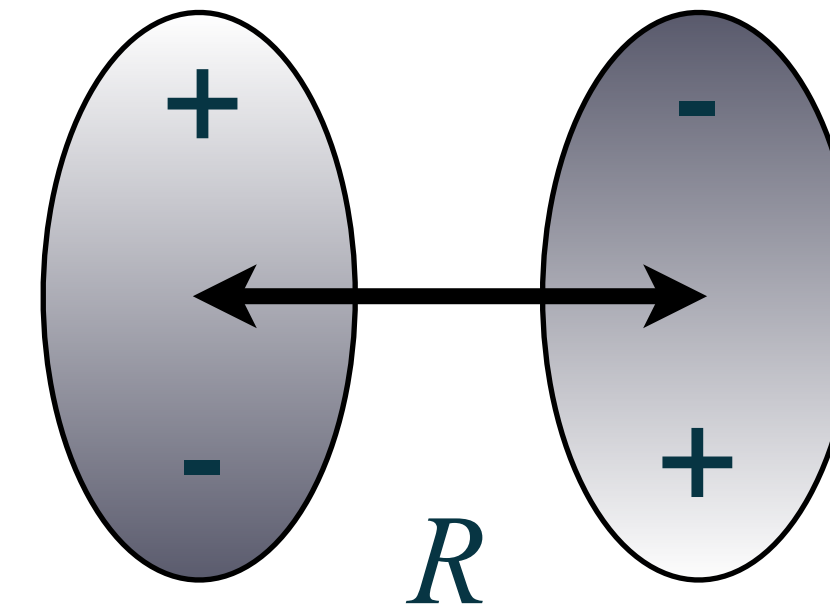
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Van der Waals forces

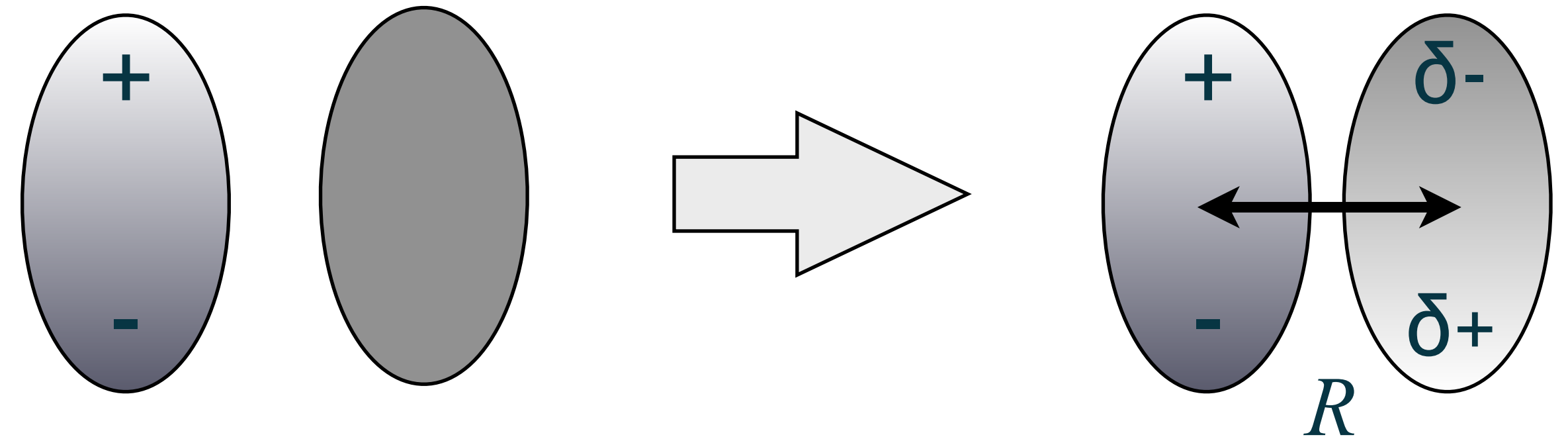
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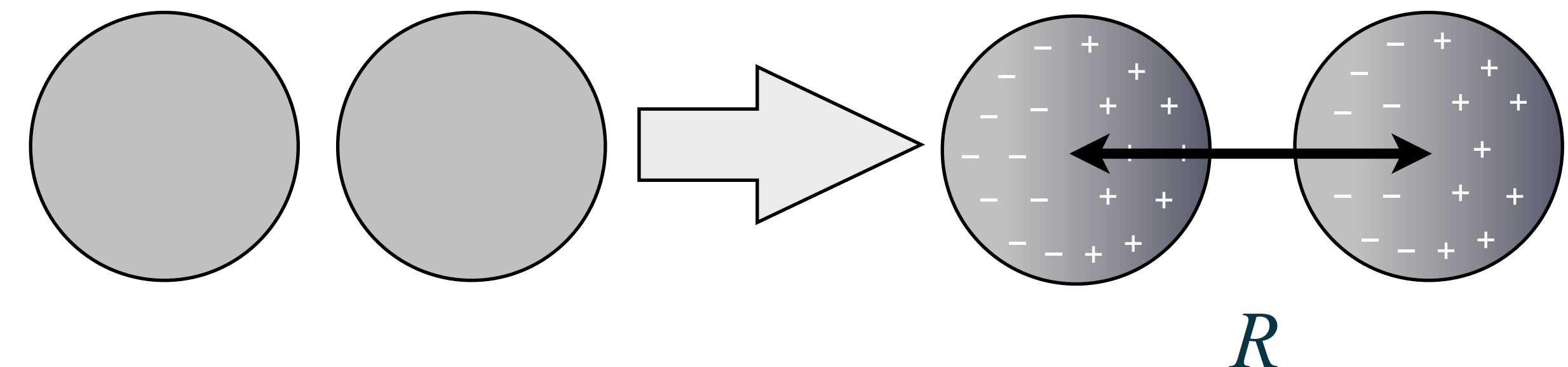


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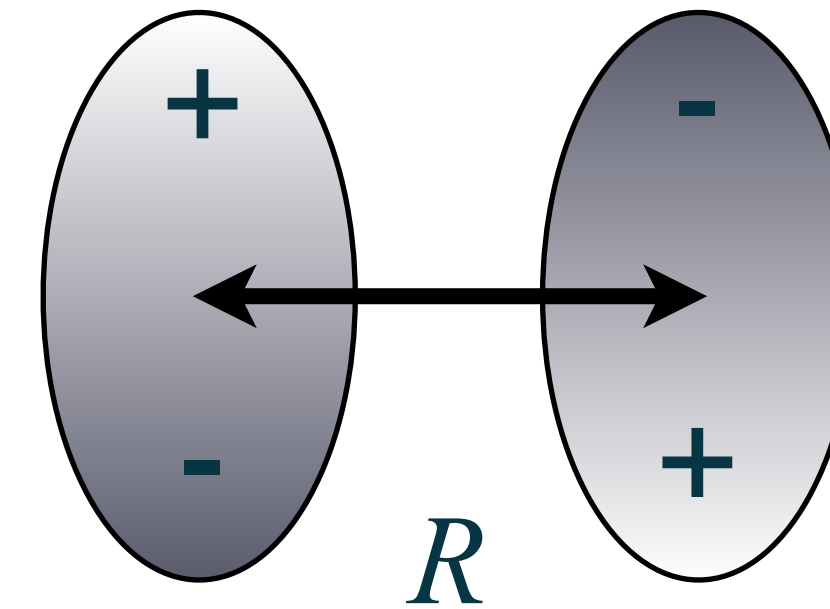
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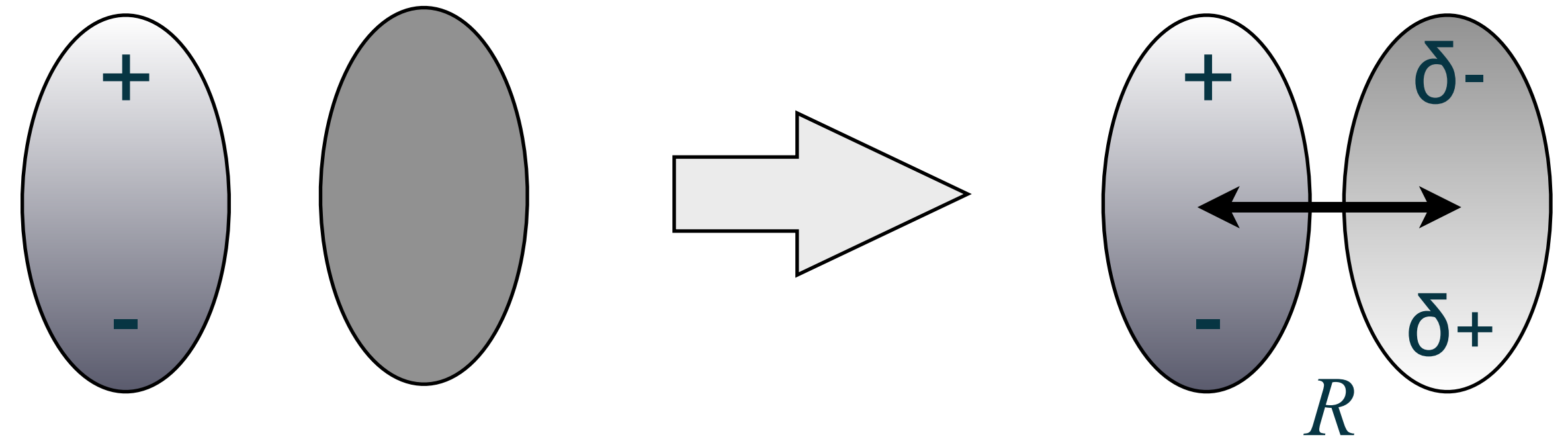
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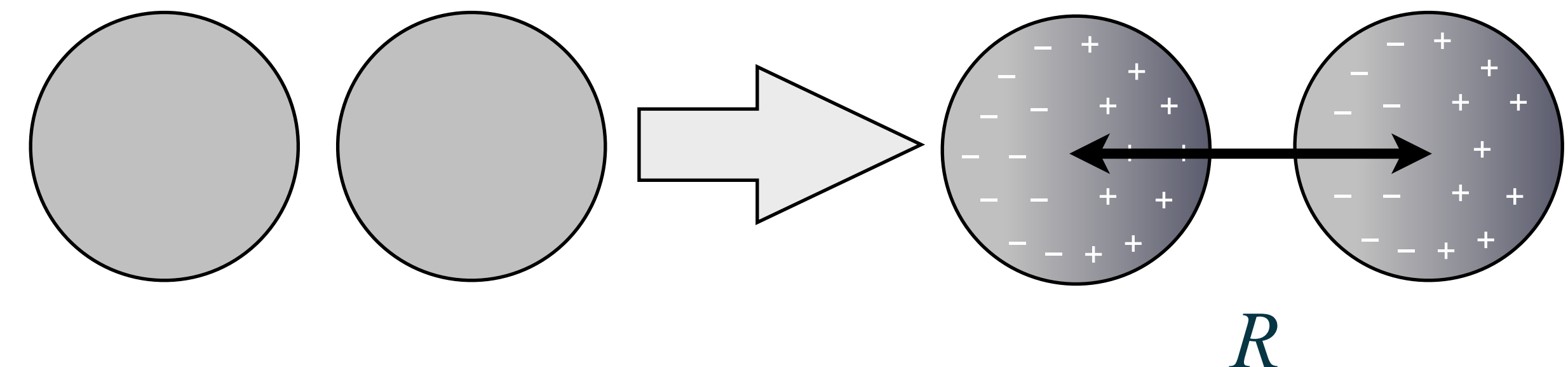


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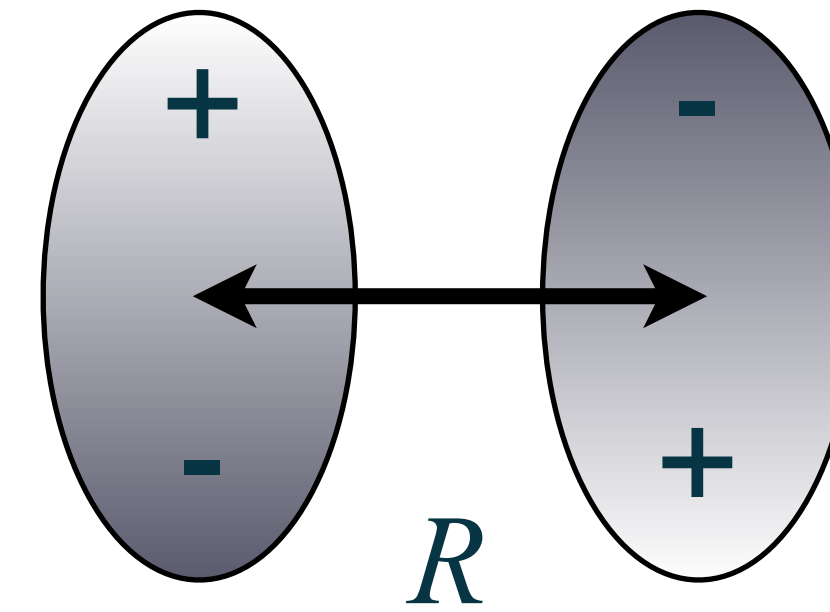


$$1/R^6$$

Van der Waals forces

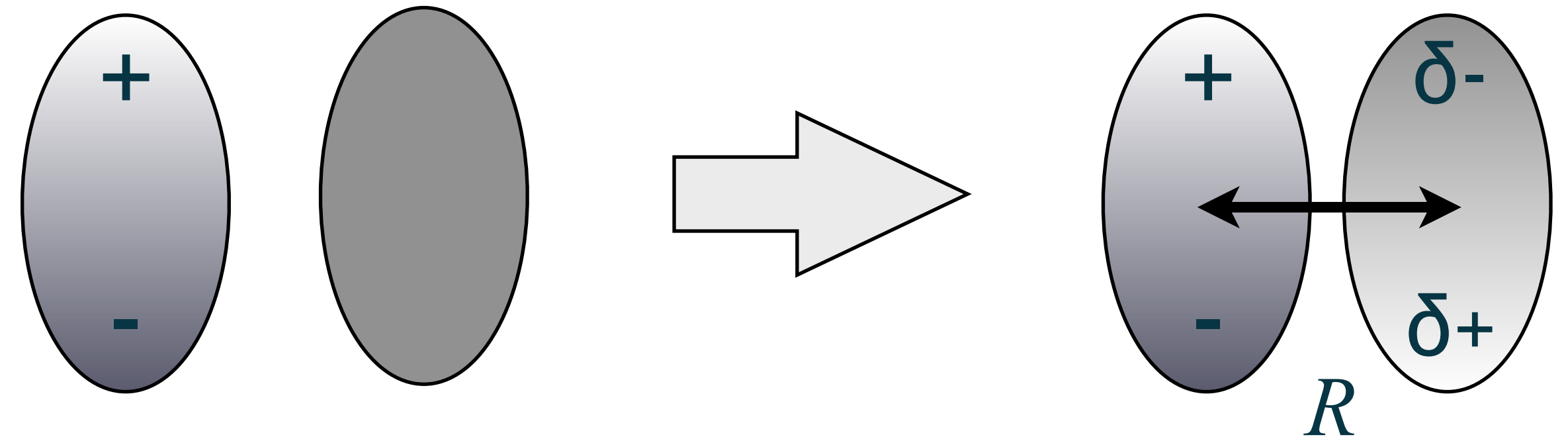
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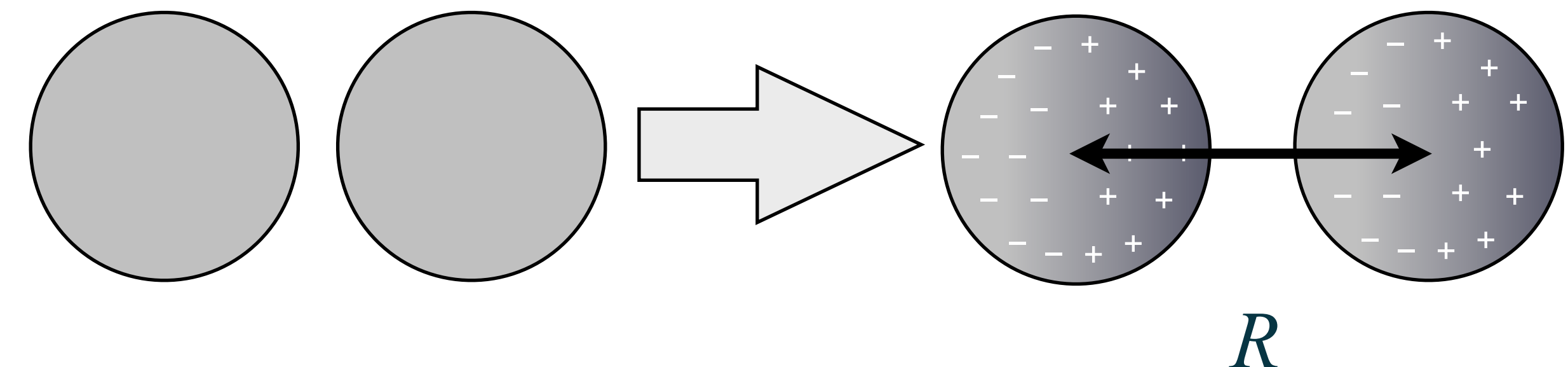
Dipole-induced dipole interaction
(Debye force)

☺ LDA/GGA works



Induced dipole-induced dipole interaction
(London dispersion force)

☹ LDA/GGA fails



Simple derivation of the dispersion interaction

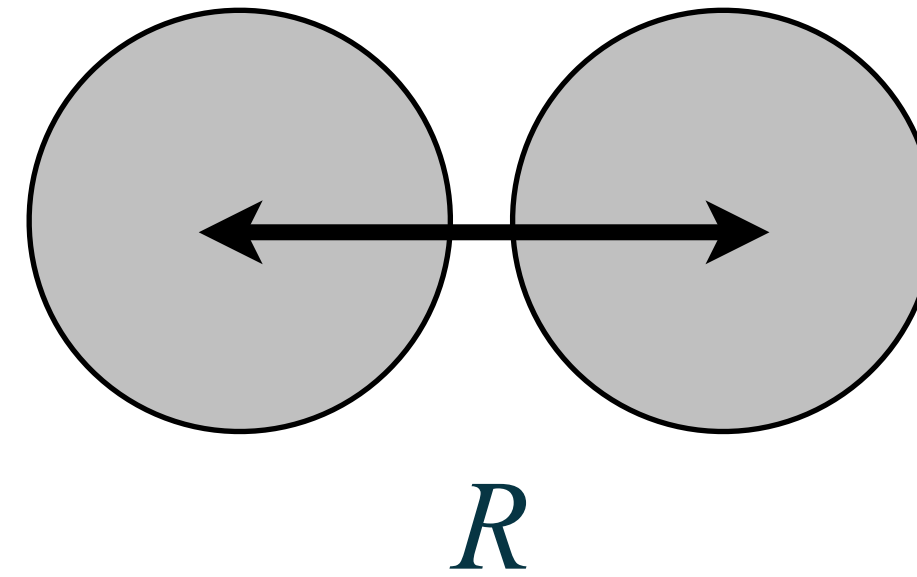
A pair of neutral atoms

$$\hat{H} = \hat{H}^{(0)} + \hat{H}'$$

$$\hat{H}^{(0)} = \hat{H}_1 + \hat{H}_2$$

$$\hat{H}_i = -\frac{\hbar^2}{2m} \nabla_i^2 + V_{\text{eff}}(\hat{\mathbf{r}}_i)$$

$$\hat{H}' = V(\hat{\mathbf{r}}_1, \hat{\mathbf{r}}_2; \hat{\mathbf{R}})$$



Simple derivation of the dispersion interaction

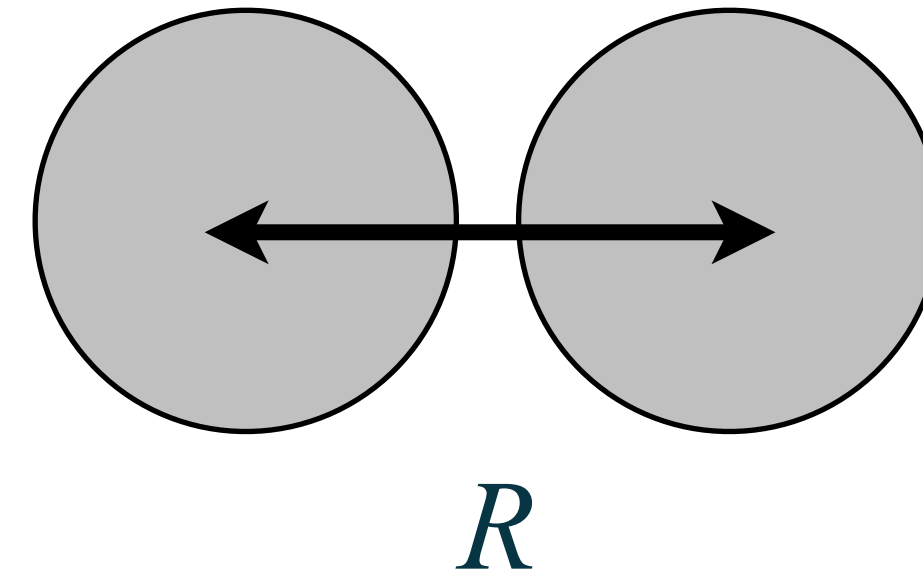
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$$R \gg 1$$

Simple derivation of the dispersion interaction

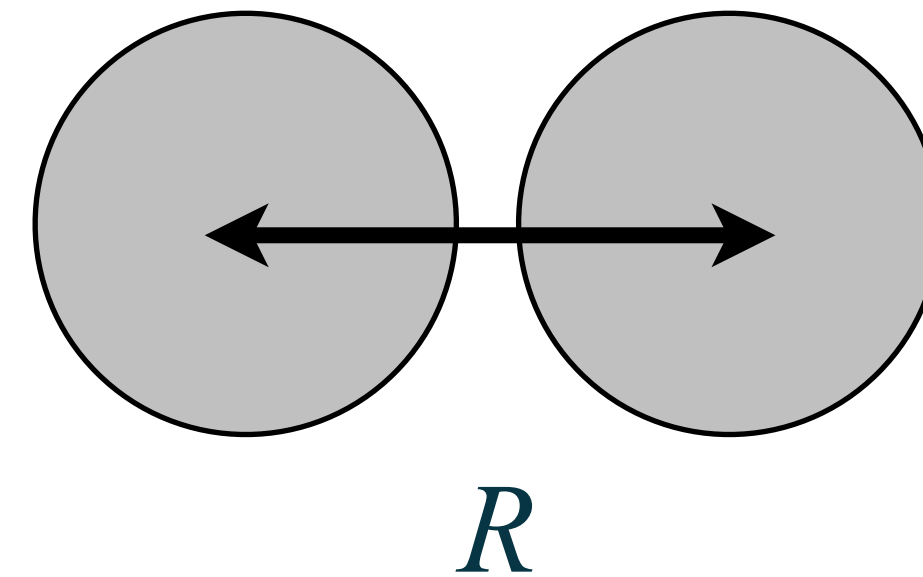
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$$R \gg 1$$

Unperturbed state (non-interacting atoms)

$$E^{(0)} = 2E_0, \hat{H}_i |\psi_0\rangle = E_0 |\psi_0\rangle$$

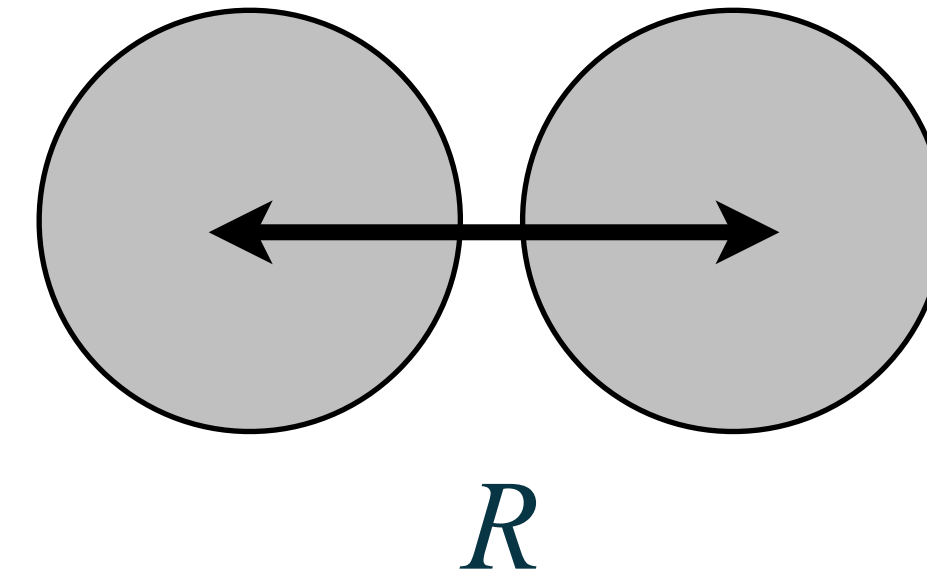
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Unperturbed state (non-interacting atoms)

$$E^{(0)} = 2E_0, \quad \hat{H}_i |\psi_0\rangle = E_0 |\psi_0\rangle$$

First-order energy

$$E^{(1)} = 0$$

Simple derivation of the dispersion interaction

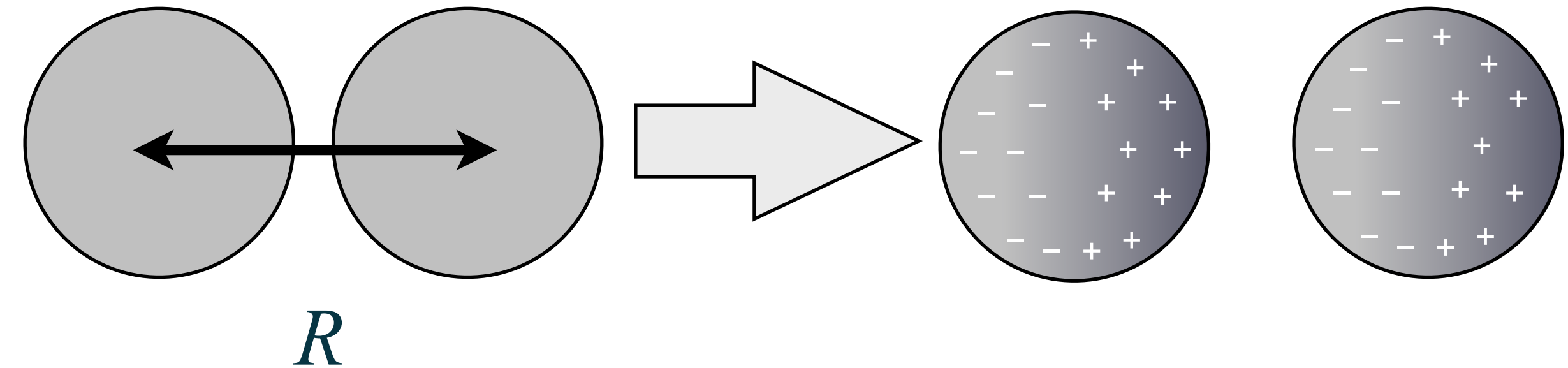
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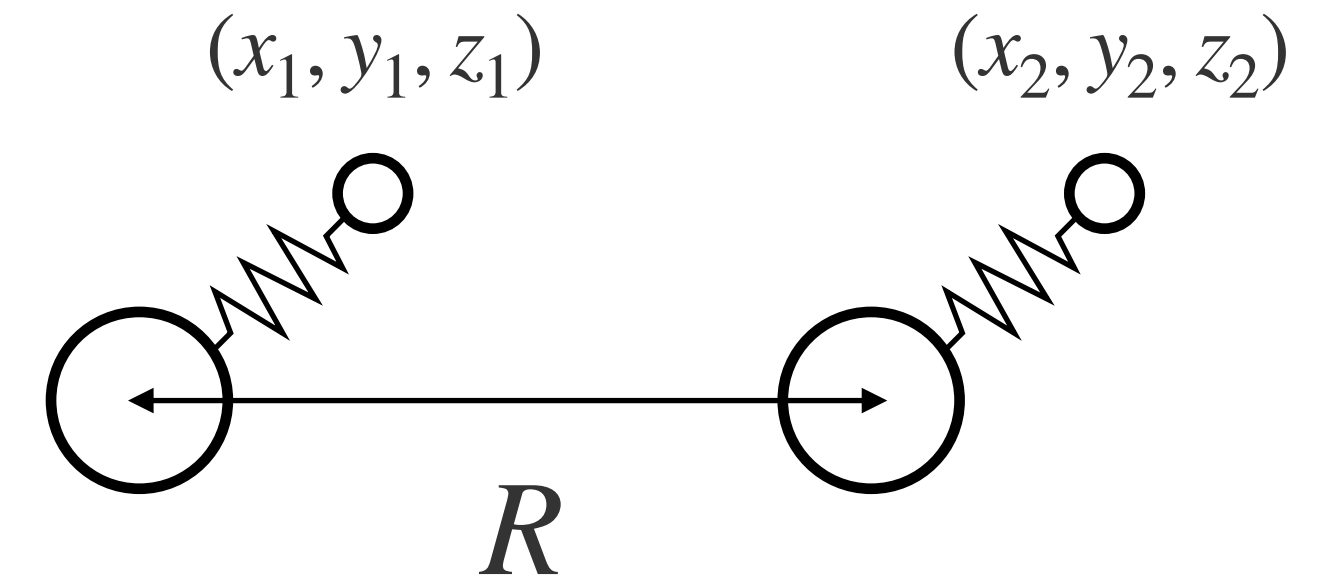
Second-order energy

$$E^{(2)} \propto 1/R^6$$

Simple derivation of the dispersion interaction

Simple derivation of the dispersion interaction

A pair of model hydrogen atoms

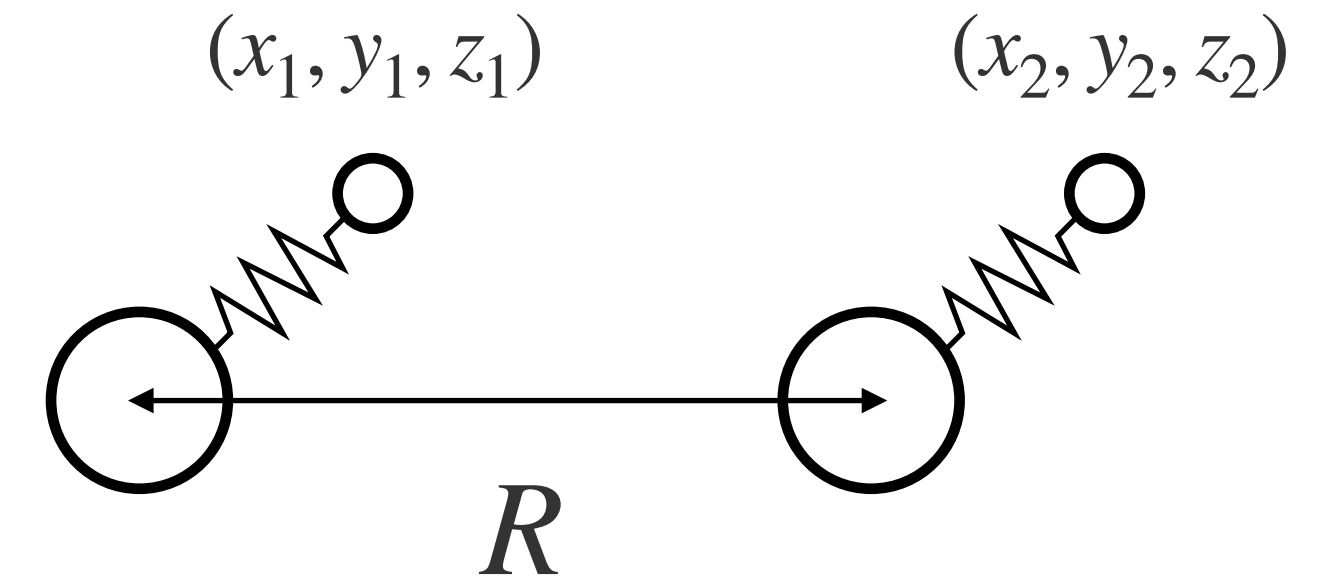


Simple derivation of the dispersion interaction

A pair of model hydrogen atoms

$$\hat{H} = \hat{H}^{(0)} + \hat{H}'$$

$$\hat{H}^{(0)} = \frac{\hat{p}_1^2}{2m} + \frac{1}{2}\hbar\omega\hat{x}_1^2 + \frac{\hat{p}_2^2}{2m} + \frac{1}{2}\hbar\omega\hat{x}_2^2$$



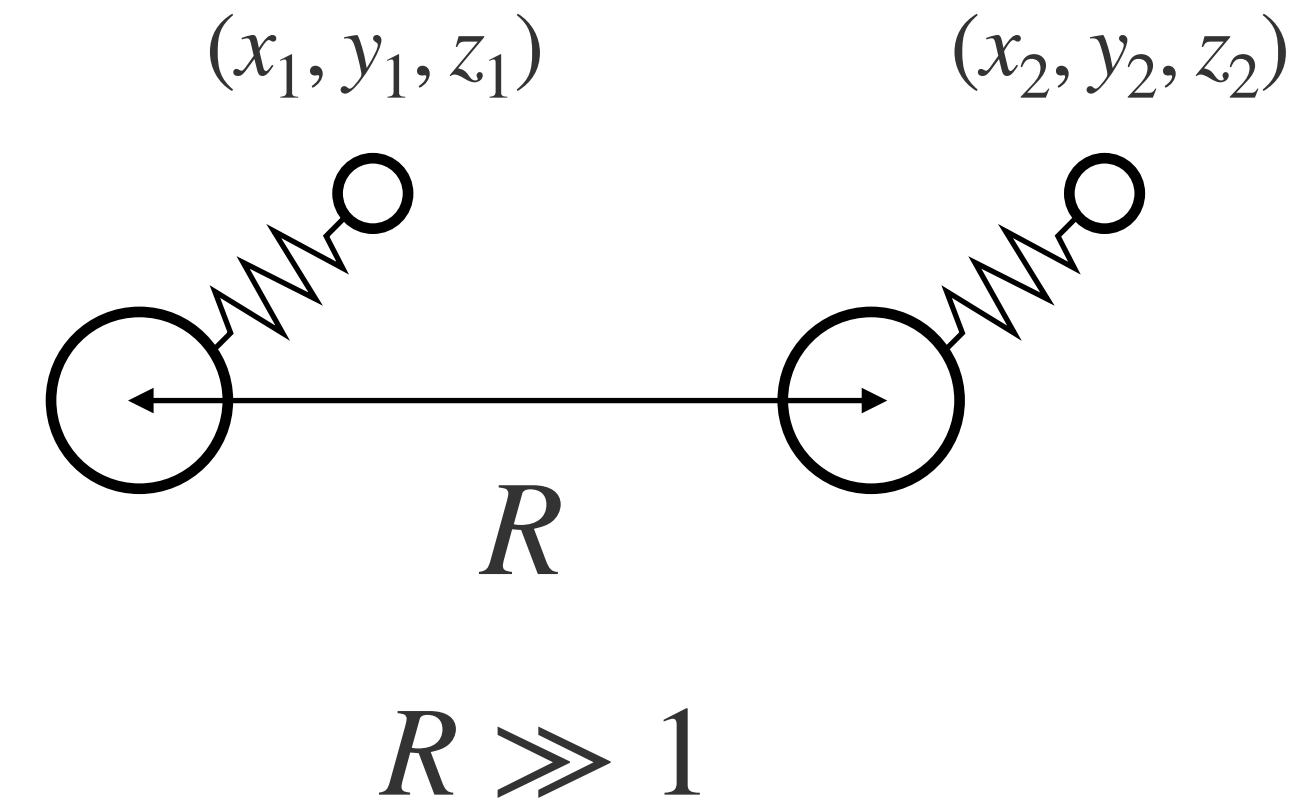
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$$\hat{H}' \approx e^2 \frac{x_1 x_2 + y_1 y_2 - 2z_1 z_2}{R^3}$$



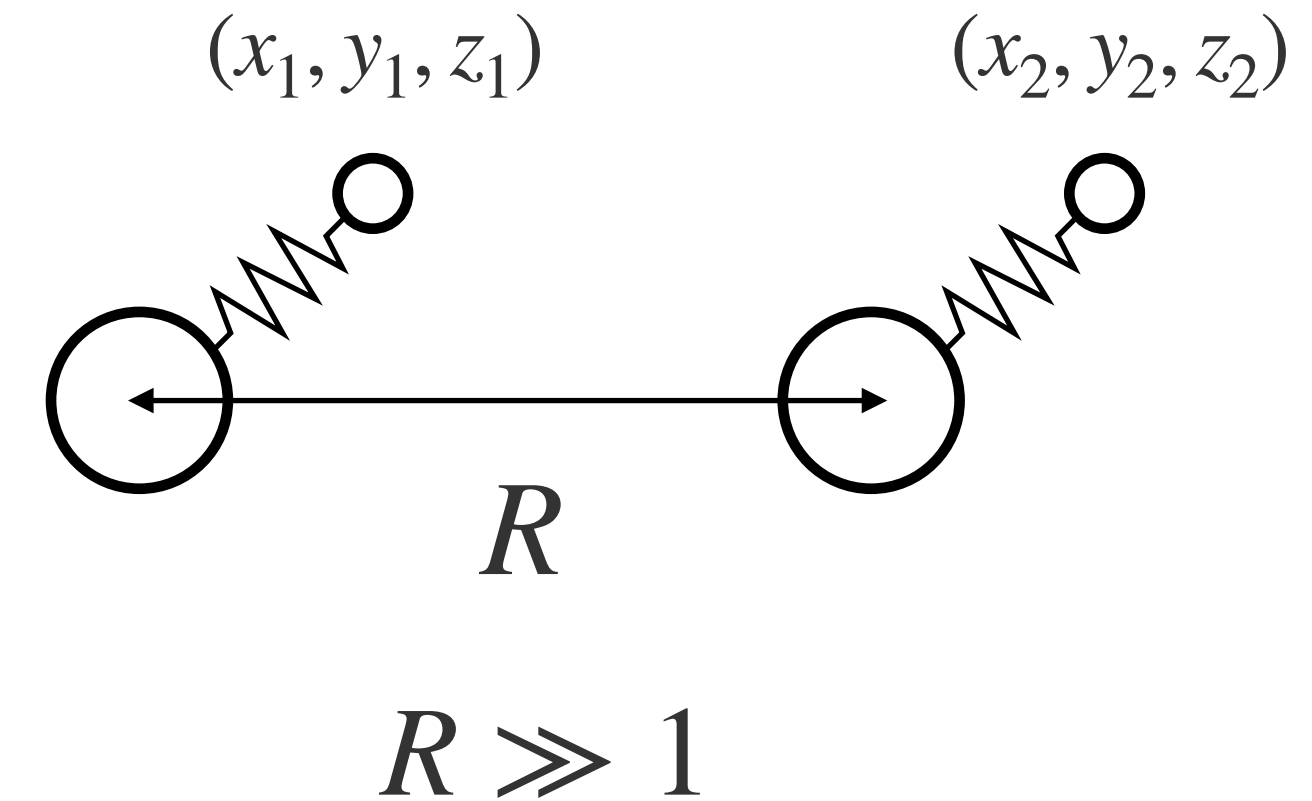
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$$E^{(2)} = -\frac{3\hbar\omega}{4} \left(\frac{e^2}{m\omega^2} \right) \frac{1}{R^6}$$

Dispersion interaction is inverse proportional to R^6

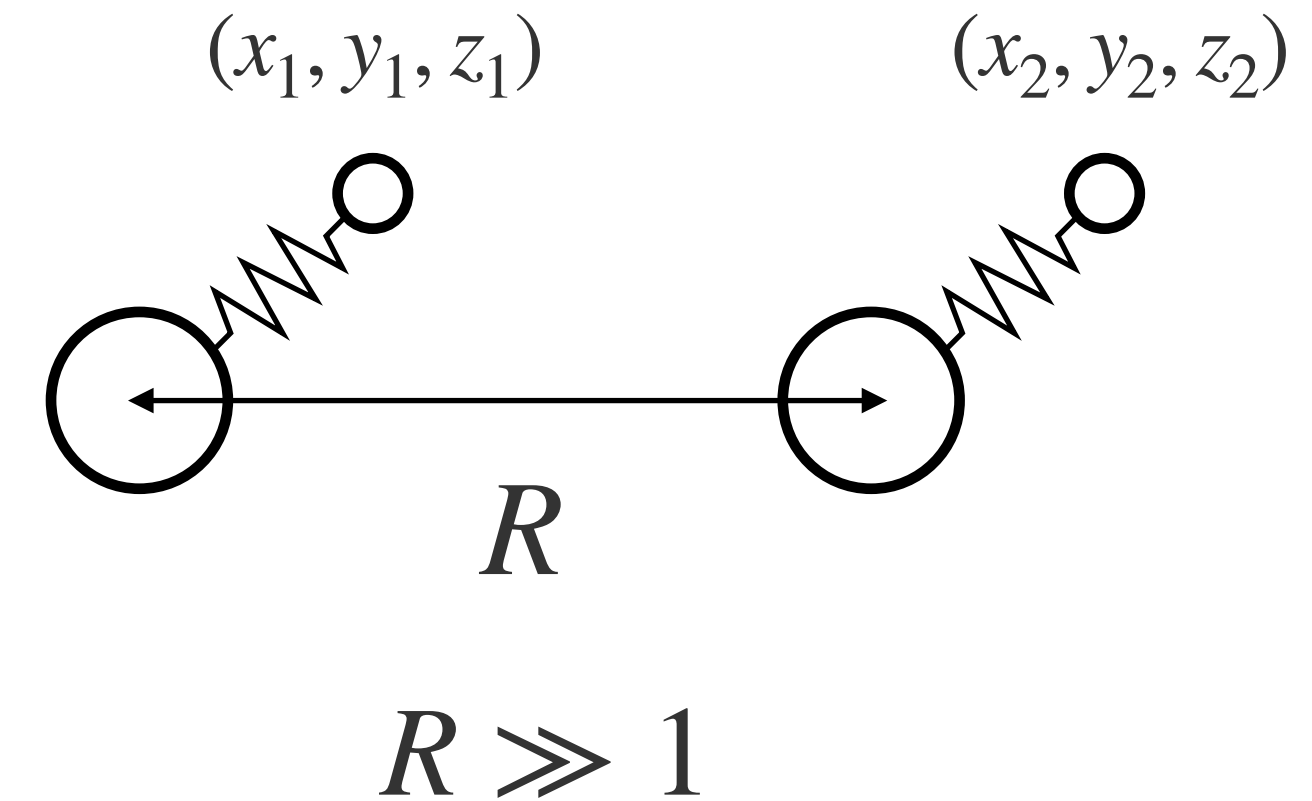
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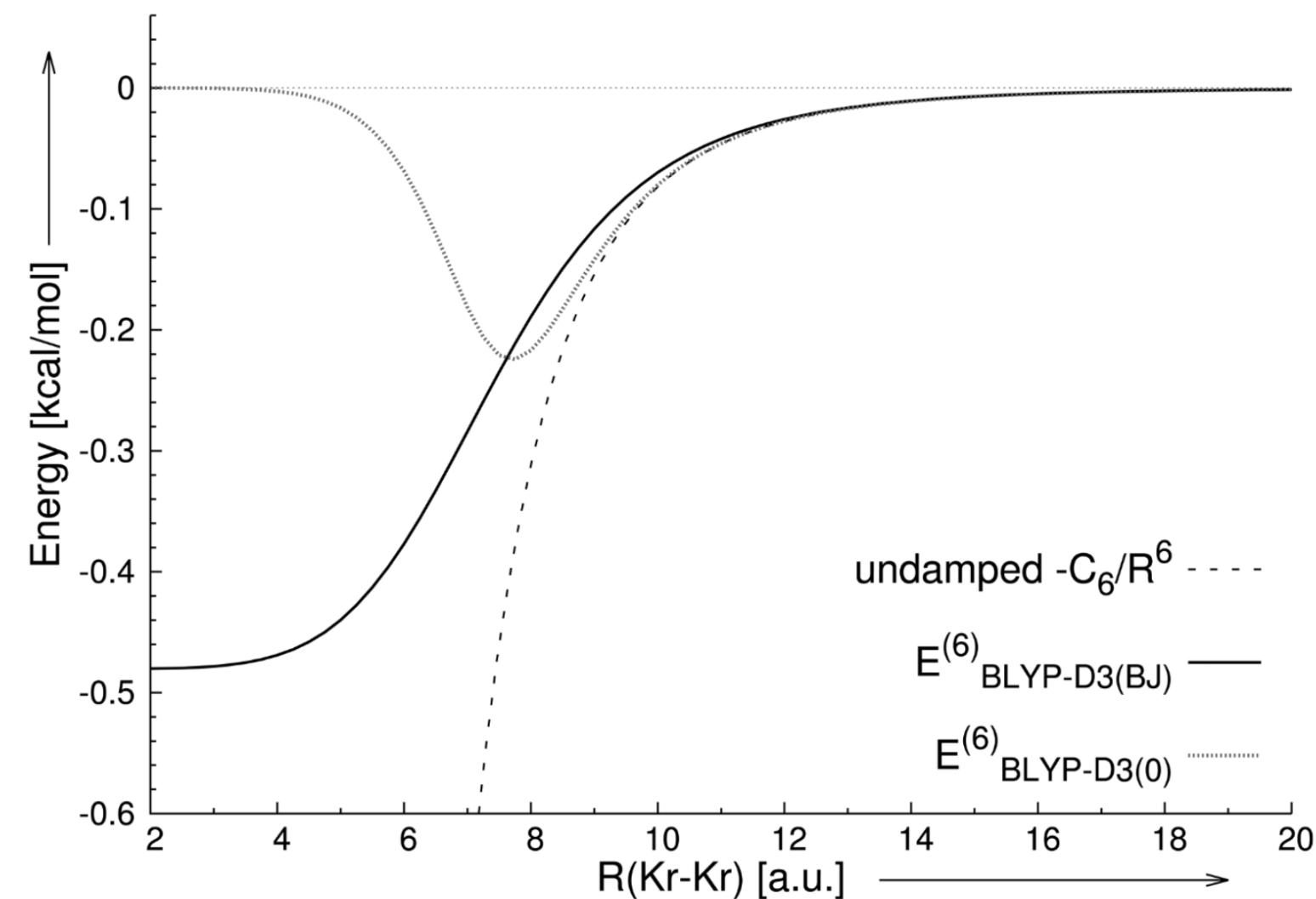
Dispersion interaction originates from nonlocal many-body interaction

Semiempirical dispersion corrected DFT

Grimme's DFT-D2

$$E_{\text{DFT-D}} = E_{\text{LDA/GGA}} - \frac{1}{2} s_6 \sum_{I \neq J} f_{\text{dmp}}(|R_I - R_J|) \frac{C_6^{A(I)B(J)}}{|R_I - R_J|^6}$$

Damping function f_{dmp}



Grimme, *et al.*, *Chem. Rev.* **116**, 5105 (2016).

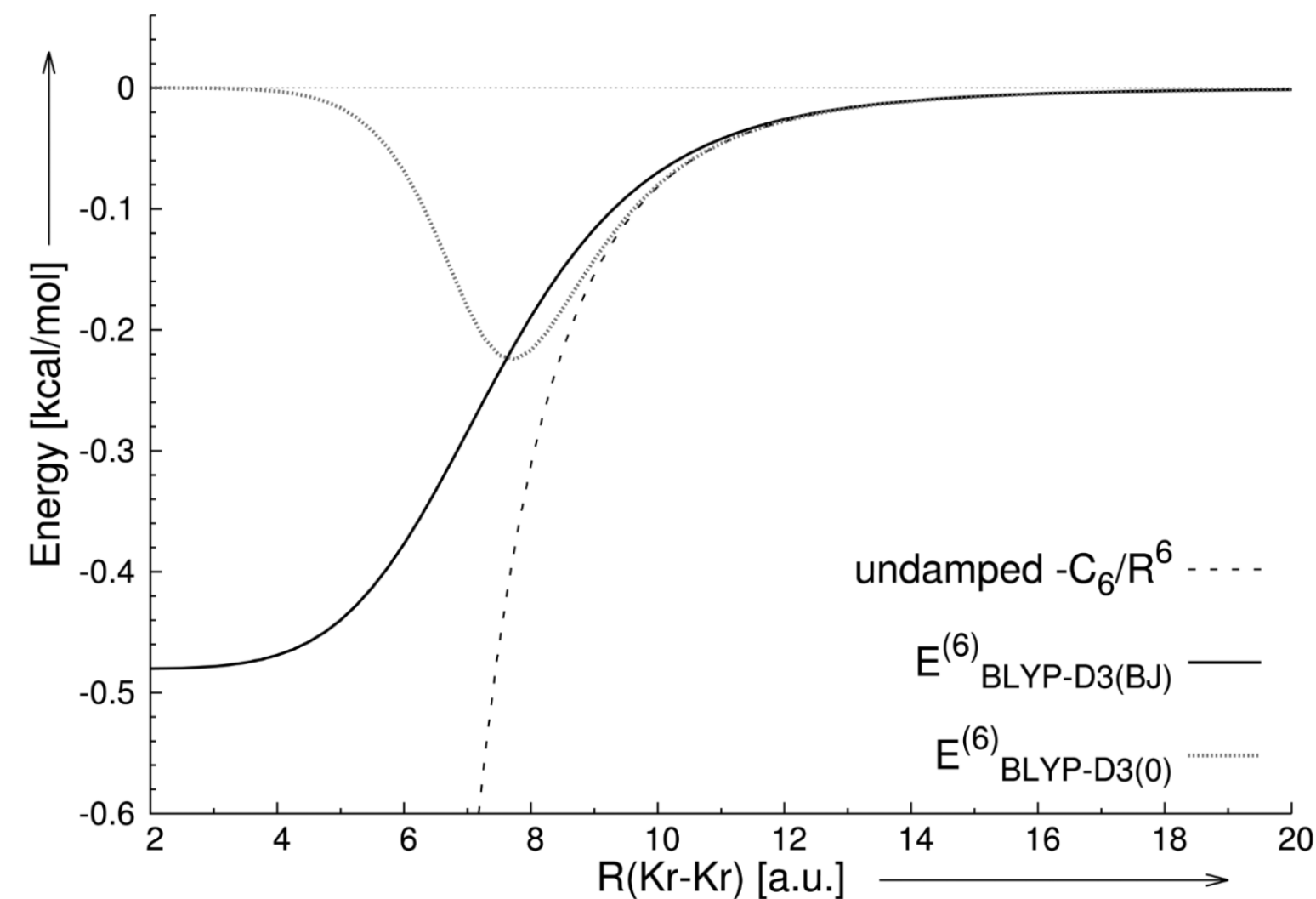
Grimme, Antony, Ehrlich, & Krieg, *J. Chem. Phys.* **132**, 154104 (2010).

Semiempirical dispersion corrected DFT

Grimme's DFT-D2

$$E_{\text{DFT-D}} = E_{\text{LDA/GGA}} - \frac{1}{2} s_6 \sum_{I \neq J} f_{\text{dmp}}(|R_I - R_J|) \frac{C_6^{A(I)B(J)}}{|R_I - R_J|^6} + (\text{higher term})$$

Damping function f_{dmp}



Grimme, *et al.*, *Chem. Rev.* **116**, 5105 (2016).

Grimme, Antony, Ehrlich, & Krieg, *J. Chem. Phys.* **132**, 154104 (2010).

Semiempirical dispersion corrected DFT

Grimme's DFT-D3

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Dispersion coefficient from the Casimir–Polder formula

$$C_6^{AB} = \frac{3}{\pi} \int_0^\infty d\omega \alpha^A(i\omega) \alpha^B(i\omega)$$

Semiempirical dispersion corrected DFT

Grimme's DFT-D3

$$E_{\text{DFT-D}} = E_{\text{LDA/GGA}} - \frac{1}{2} s_6 \sum_{I \neq J} f_{\text{dmp}}(|R_I - R_J|) \frac{C_6^{A(I)B(J)}}{|R_I - R_J|^6} \\ + (\text{higher term})$$

Dispersion coefficient from the modified Casimir–Polder formula

$$C_{6,\text{ref}}^{AB} = \frac{3}{\pi} \int_0^\infty d\omega \frac{1}{m} \left[\alpha^{A_m H_n}(i\omega) - \frac{n}{2} \alpha^{H_2}(i\omega) \right] \times \frac{1}{k} \left[\alpha^{B_k H_l}(i\omega) - \frac{l}{2} \alpha^{H_2}(i\omega) \right]$$

Semiempirical dispersion corrected DFT

Grimme's DFT-D3

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System specific dispersion coefficient via the coordination number (CN)

$$C_6^{AB}(\text{CN}^A, \text{CN}^B) = \frac{\sum_I^{N_A} \sum_J^{N_B} C_{6,\text{ref}}(\text{CN}^A, \text{CN}^B) L_{IJ}}{\sum_I^{N_A} \sum_J^{N_B} L_{IJ}}$$

$$L_{IJ} = \exp \left[-4 \left((\text{CN}^A - \text{CN}_I^A)^2 + (\text{CN}^B - \text{CN}_J^B)^2 \right) \right]$$

Grimme, Antony, Ehrlich, & Krieg, *J. Chem. Phys.* **132**, 154104 (2010).

Semiempirical dispersion corrected DFT

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☹ Accuracy can depend on DFT XC functional
(e.g., PBE+D3 tends to overestimate the binding energy)

Grimme, Antony, Ehrlich, & Krieg, *J. Chem. Phys.* **132**, 154104 (2010).

van der Waals density functional (vdW-DF)

van der Waals density functional (vdW-DF)

Nonempirical exchange-correlation functional

No orbital dependency - pure density functional

van der Waals density functional (vdW-DF)

van der Waals density functional (vdW-DF)

XC functional — sum of semilocal exchange-correlation and nonlocal correlation

$$E_{\text{xc}} = E_{\text{xc}}^{\text{sl}} + E_{\text{c}}^{\text{nl}}$$

van der Waals density functional (vdW-DF)

XC functional — sum of semilocal exchange-correlation and **nonlocal correlation**

$$E_{\text{xc}} = E_{\text{xc}}^{\text{sl}} + E_{\text{c}}^{\text{nl}}$$

van der Waals density functional (vdW-DF)

XC energy from the adiabatic connection fluctuation dissipation theorem

$$E_{\text{xc}} = - \int_0^1 \frac{d\lambda}{\lambda} \int \frac{du}{2\pi} \iint d\mathbf{r} d\mathbf{r}' \chi^\lambda(\mathbf{r}, \mathbf{r}'; iu) \frac{\lambda e^2}{|\mathbf{r} - \mathbf{r}'|} - E_{\text{self}}$$

van der Waals density functional (vdW-DF)

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Density-density response function at λ

Coulomb kernel $V(\mathbf{r}, \mathbf{r}')$ at λ

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Density-density response function at λ

Coulomb kernel $V(\mathbf{r}, \mathbf{r}')$ at λ

- Full potential approximation — Exact at the van der Waals asymptote
- Second-order expansion in the response function
- Plasmon-pole approximation to the response function

van der Waals density functional (vdW-DF)

XC energy from the adiabatic connection fluctuation dissipation theorem

$$E_{\text{xc}} = - \int_0^1 \frac{d\lambda}{\lambda} \int \frac{du}{2\pi} \text{tr} [\tilde{\chi}^\lambda(iu) V^\lambda] - E_{\text{self}}$$

van der Waals density functional (vdW-DF)

XC energy from the adiabatic connection fluctuation dissipation theorem

$$E_{\text{xc}} = - \int_0^1 \frac{d\lambda}{\lambda} \int \frac{du}{2\pi} \text{tr} [\tilde{\chi}^\lambda(iu) V^\lambda] - E_{\text{self}}$$

Full potential approximation (FPA)

$$\tilde{\chi} \equiv \tilde{\chi}^{\lambda=1}$$

$$E_{\text{xc}}^{\text{FPA}} = \int \frac{du}{2\pi} \text{tr} \left[\ln (1 - \tilde{\chi} V) \right] - E_{\text{self}}$$

van der Waals density functional (vdW-DF)

XC energy from the adiabatic connection fluctuation dissipation theorem

$$E_{\text{xc}} = - \int_0^1 \frac{d\lambda}{\lambda} \int \frac{du}{2\pi} \text{tr} [\tilde{\chi}^\lambda(iu) V^\lambda] - E_{\text{self}}$$

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LDA-like approximation to local XC to extract the nonlocal correlation

$$E_{\text{c}}^{\text{nl}} = E_{\text{xc}}^{\text{FPA}} - E_{\text{xc}}^{\text{FPA}}(\text{HEG})$$

van der Waals density functional (vdW-DF)

XC energy from the adiabatic connection fluctuation dissipation theorem

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LDA-like approximation to local XC to extract the nonlocal correlation

$$E_{\text{c}}^{\text{nl}} = \int \frac{du}{2\pi} \text{tr} [\ln (1 - \tilde{\chi} V) - \ln \epsilon]$$

van der Waals density functional (vdW-DF)

Nonlocal correlation in terms of the dielectric function ϵ

$$E_c^{\text{nl}} = \int \frac{du}{2\pi} \text{tr} \left[\ln \left(-\frac{1}{4\pi} \nabla \cdot \epsilon \nabla V \right) - \ln \epsilon \right]$$

van der Waals density functional (vdW-DF)

Nonlocal correlation in terms of the dielectric function ϵ

$$E_c^{\text{nl}} = \int \frac{du}{2\pi} \text{tr} \left[\ln \left(-\frac{1}{4\pi} \nabla \cdot \epsilon \nabla V \right) - \ln \epsilon \right]$$

Second-order expansion wrt S

$$S = 1 - \epsilon^{-1}$$

$$E_c^{\text{nl}} \approx \int \frac{du}{4\pi} \text{tr} \left[S^2 - \left(\frac{\nabla S \cdot \nabla V}{4\pi} \right)^2 \right]$$

van der Waals density functional (vdW-DF)

van der Waals density functional (vdW-DF)

Plasmon-pole approximation to S

$$S(\mathbf{r}, \omega) = \frac{\omega_p^2(\mathbf{r})}{\omega_{\mathbf{q}}^2(\mathbf{r}) - \omega^2}$$
$$\omega_p(\mathbf{r}) = \sqrt{4\pi n(\mathbf{r})}$$

van der Waals density functional (vdW-DF)

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Dispersion function

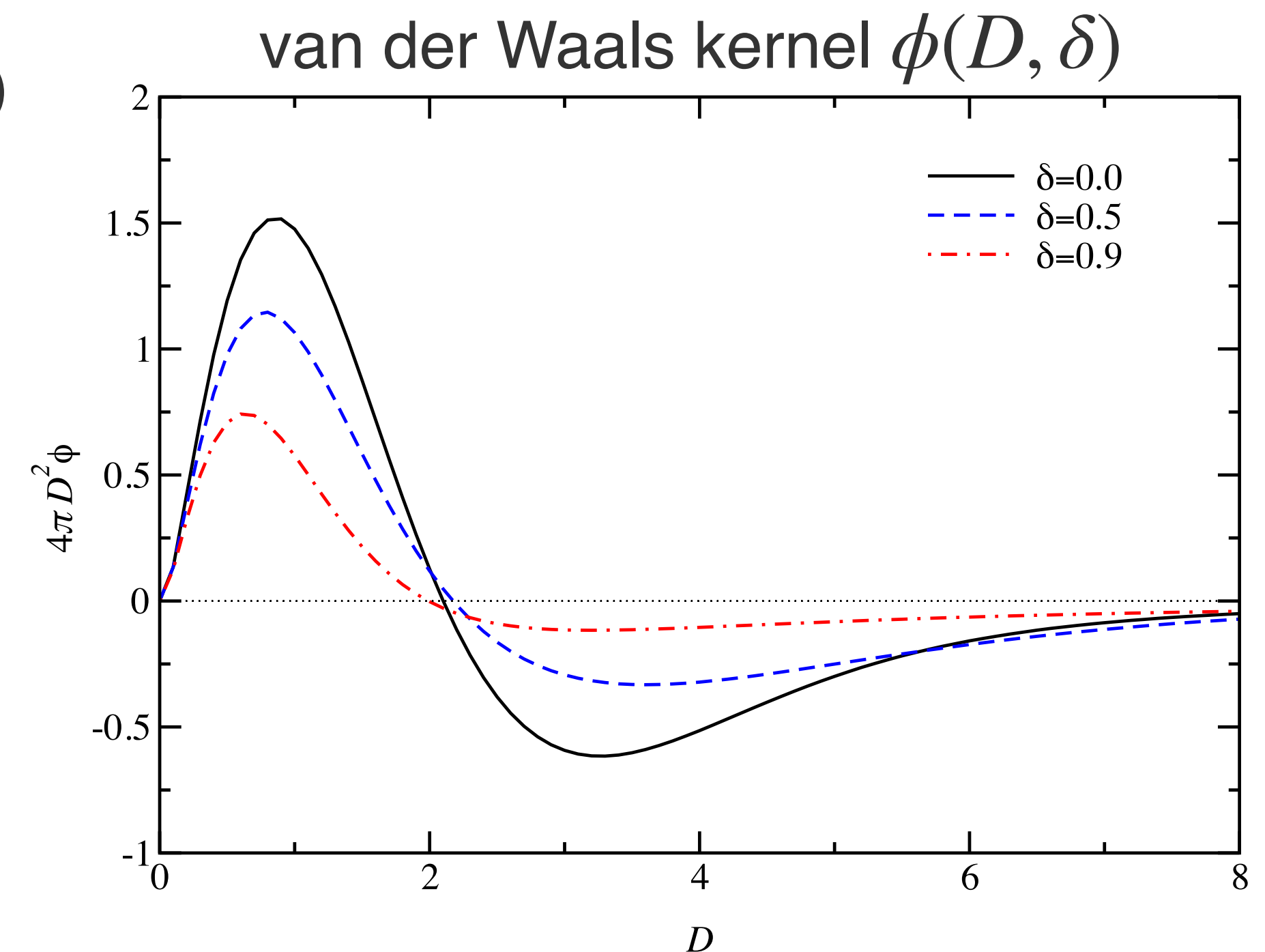
$$\omega_{\mathbf{q}} = \frac{q^2}{2m} \frac{1}{h[q/q_0]}$$
$$h(x) = 1 - e^{-\frac{4\pi}{9}x^2}$$
$$q_0(\mathbf{r}) = -\frac{4\pi}{3}\epsilon_{\text{xc}}^{\text{LDA}}(n(\mathbf{r})) - \frac{Z_{ab}}{9}s^2(\mathbf{r})k_F(\mathbf{r})$$

van der Waals density functional (vdW-DF)

Nonlocal correlation

$$E_c^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$



van der Waals density functional (vdW-DF)

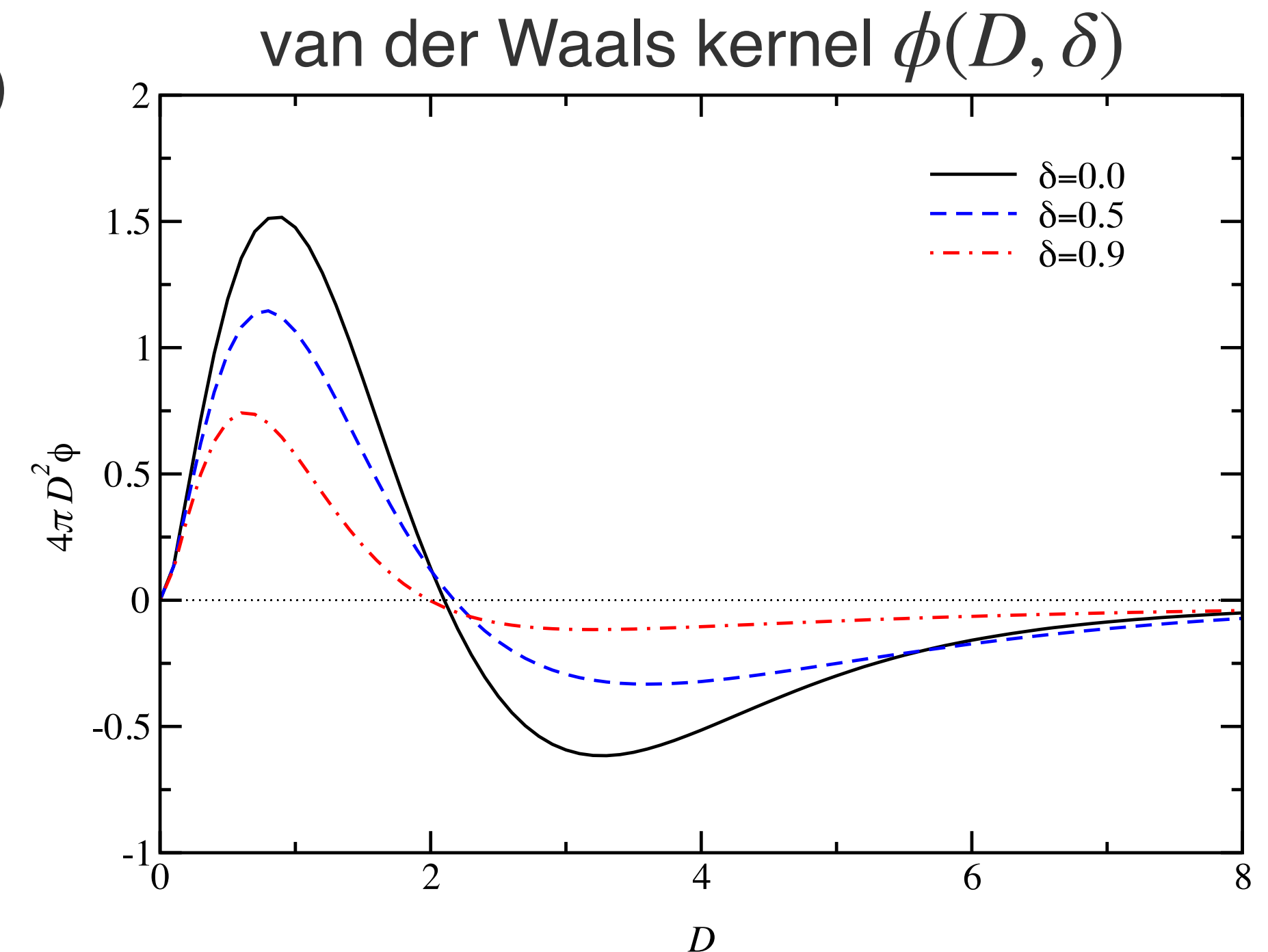
Nonlocal correlation

$$E_c^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

$$|r_1 - r_2| \rightarrow \infty$$

$$\phi \rightarrow -\frac{3}{2} \frac{1}{|r_1 - r_2|^6} \frac{1}{q_0^2(\mathbf{r}_1) q_0^2(\mathbf{r}_2) [q_0^2(\mathbf{r}_1) + q_0^2(\mathbf{r}_2)]}$$



van der Waals density functional (vdW-DF)

Nonlocal correlation

$$E_c^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

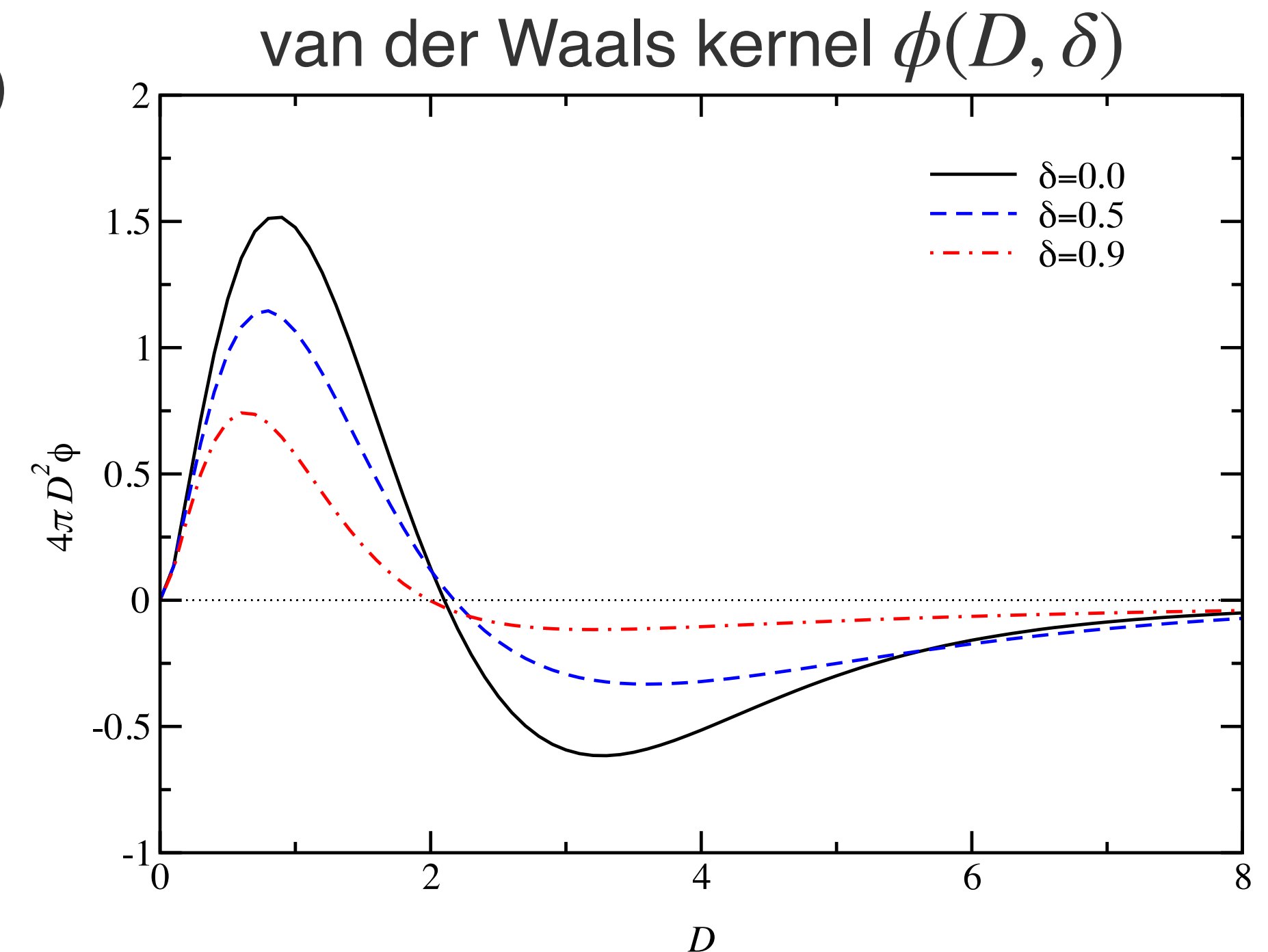
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Uniform (homogeneous) limit

$$E_c^{\text{nl}} \rightarrow 0$$



van der Waals density functional (vdW-DF)

XC functional in vdW-DF

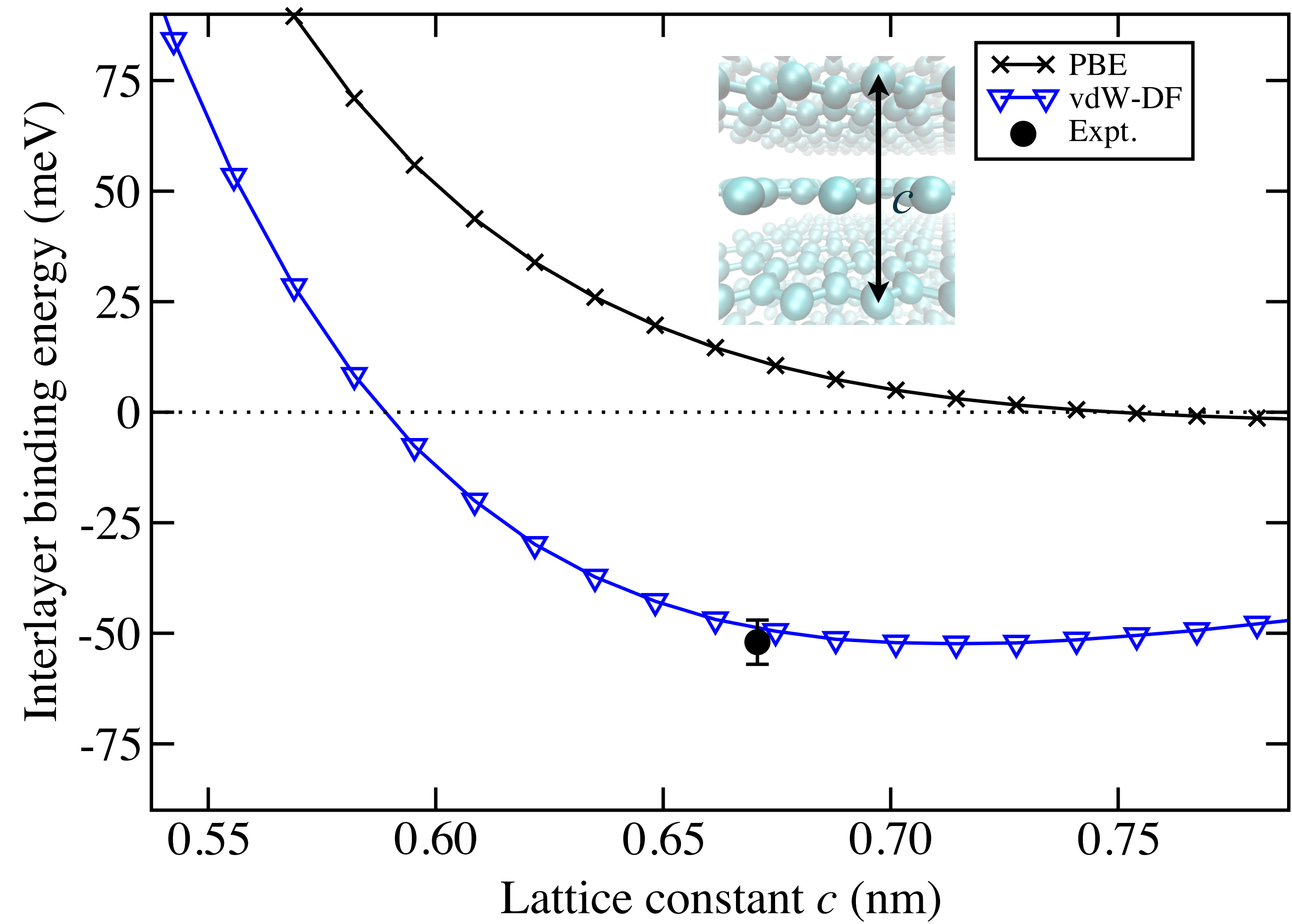
$$E_{\text{xc}}^{\text{vdW-DF}} = E_{\text{x}}^{\text{GGA}} + E_{\text{c}}^{\text{LDA}} + E_{\text{c}}^{\text{nl}}$$

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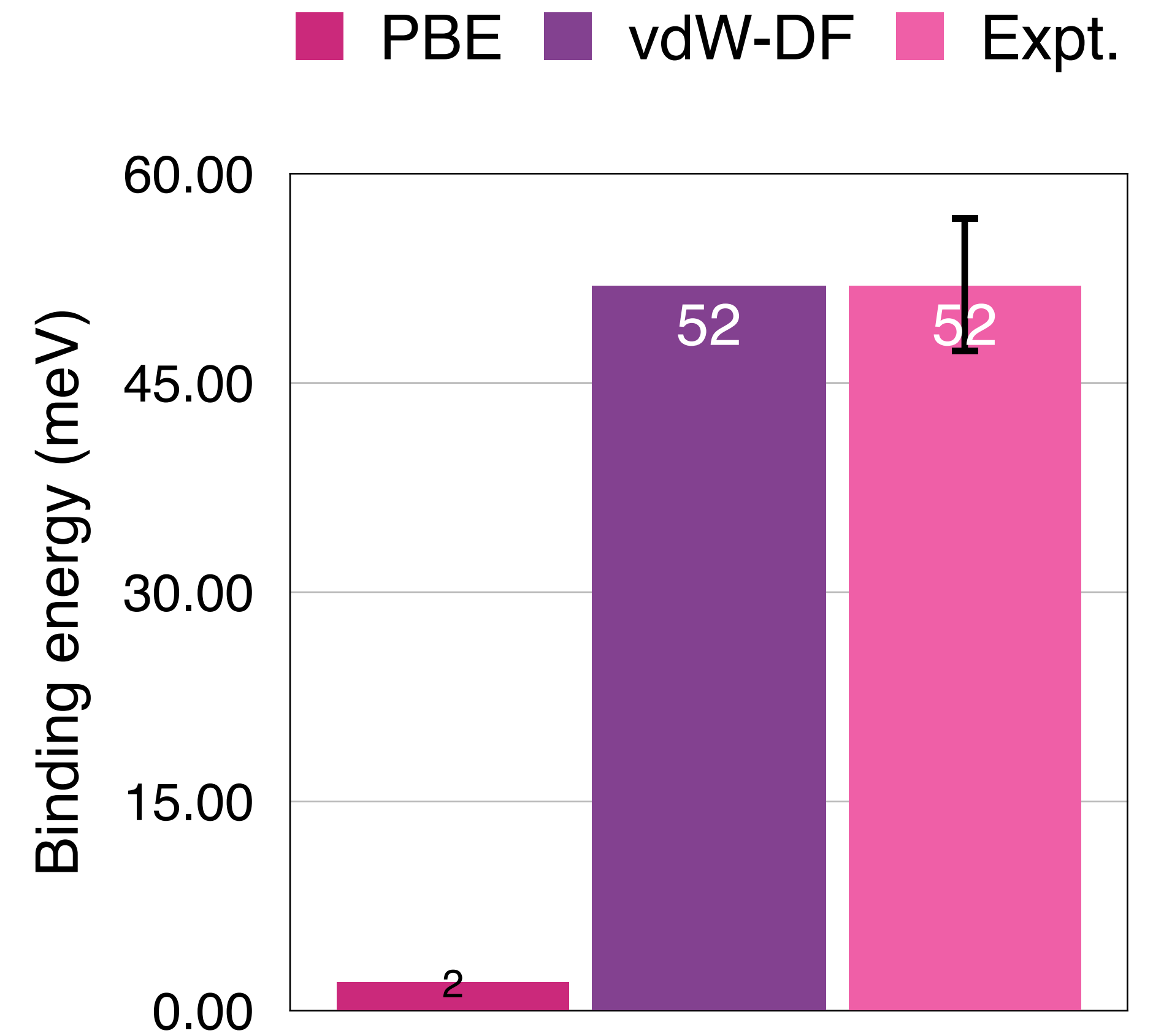
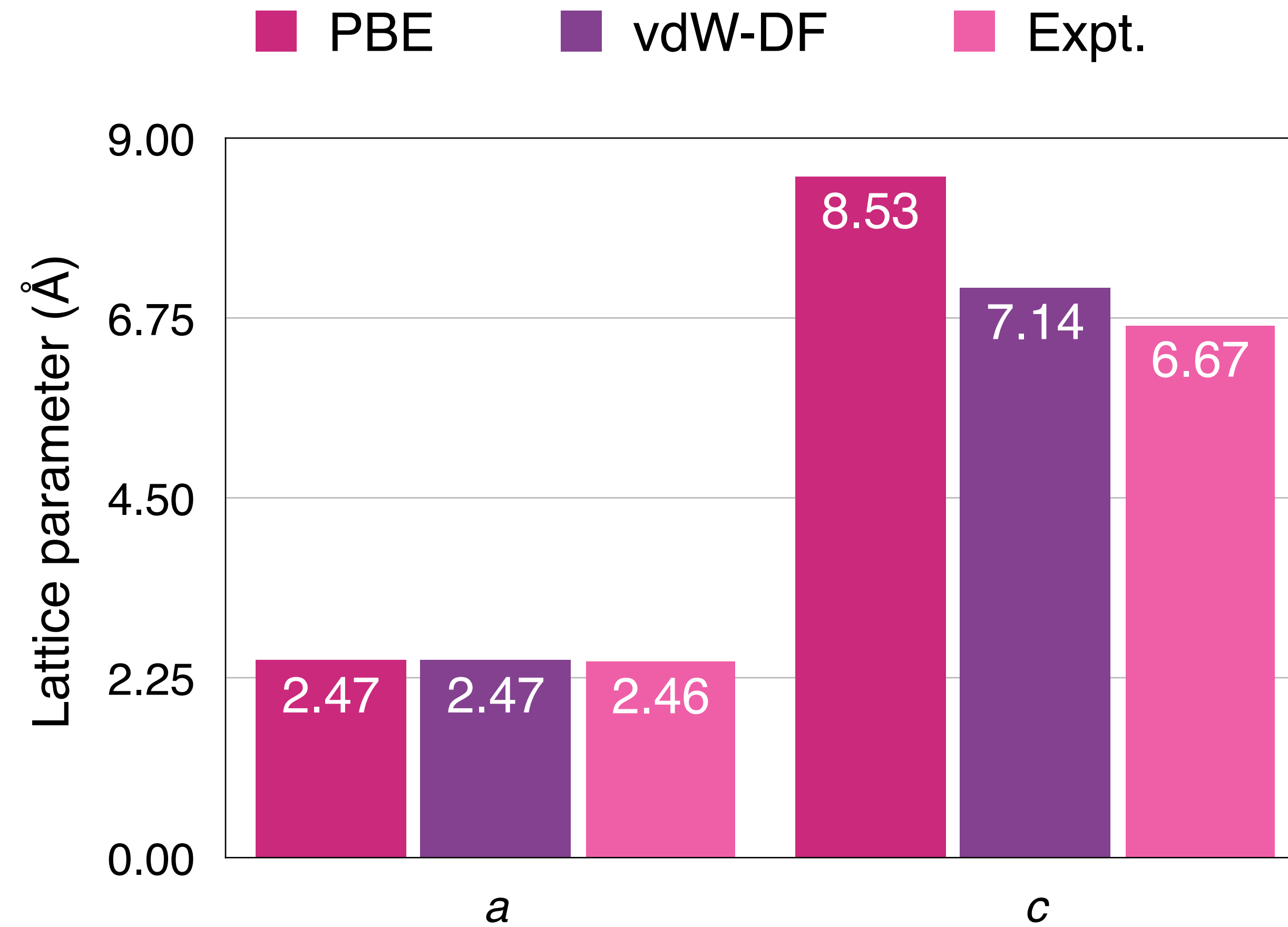
$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

- ☺ No empirical parameters
- ☺ Seamless description of covalent and van der Waals bondings

Graphite

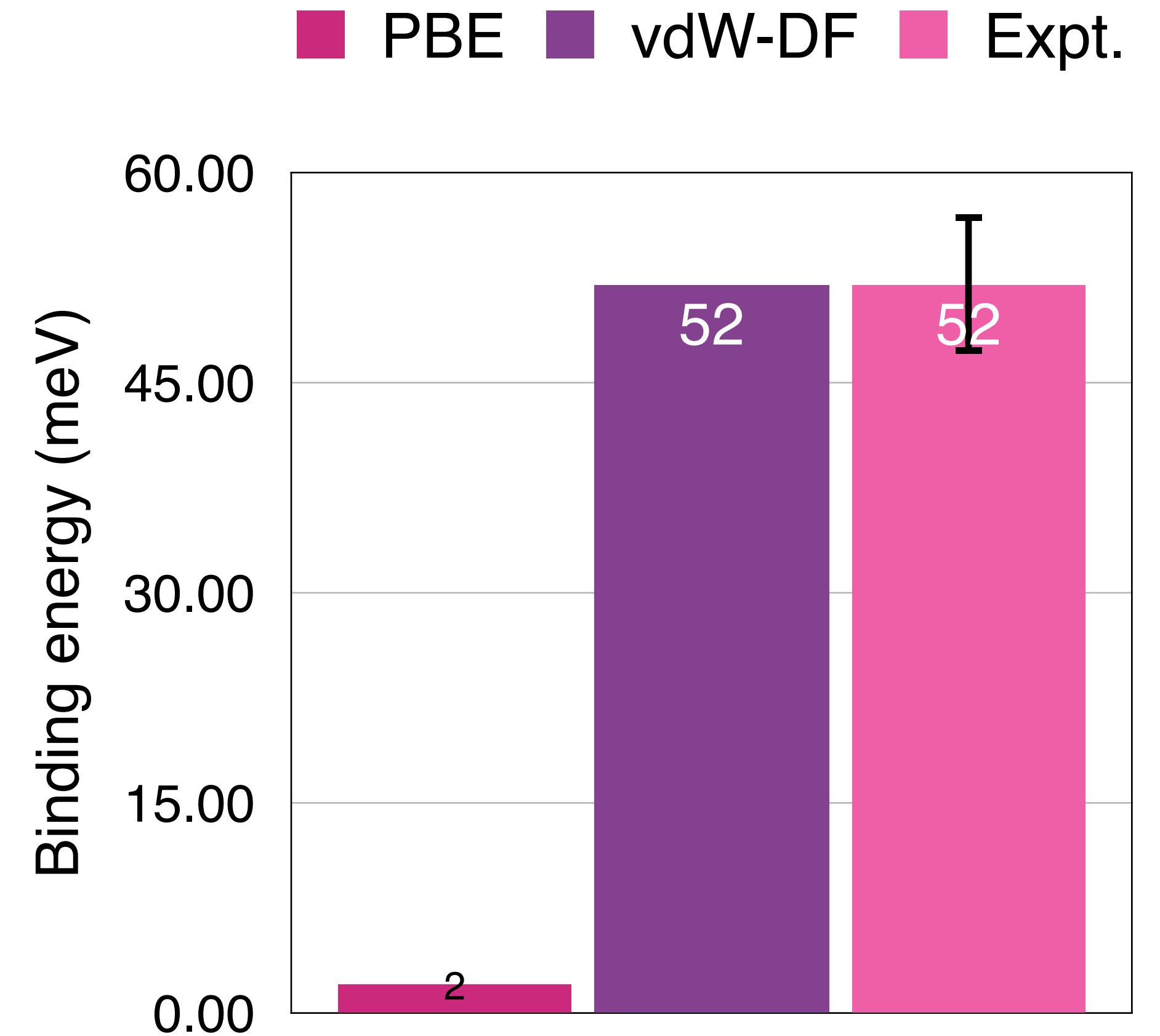
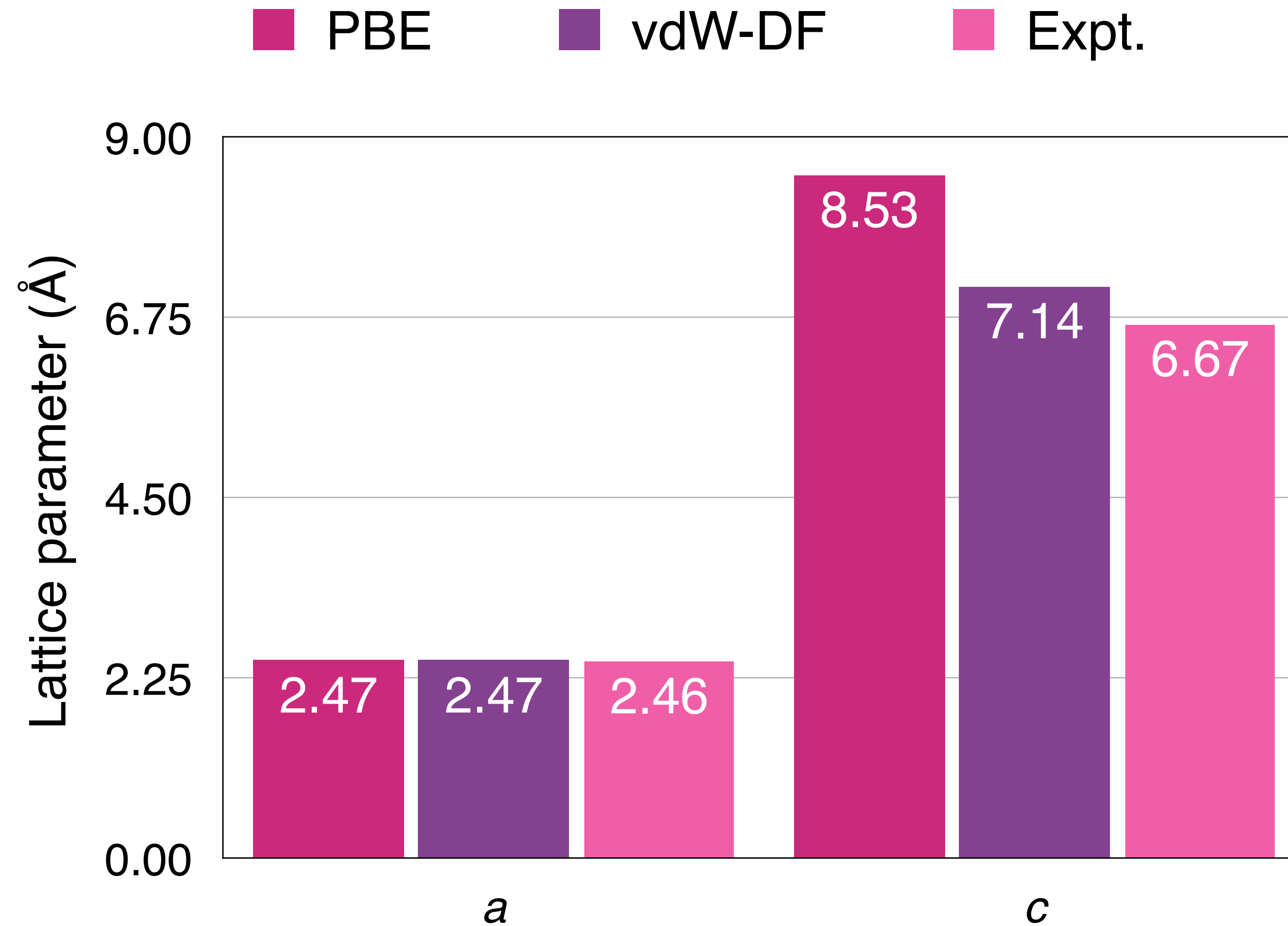


Graphite



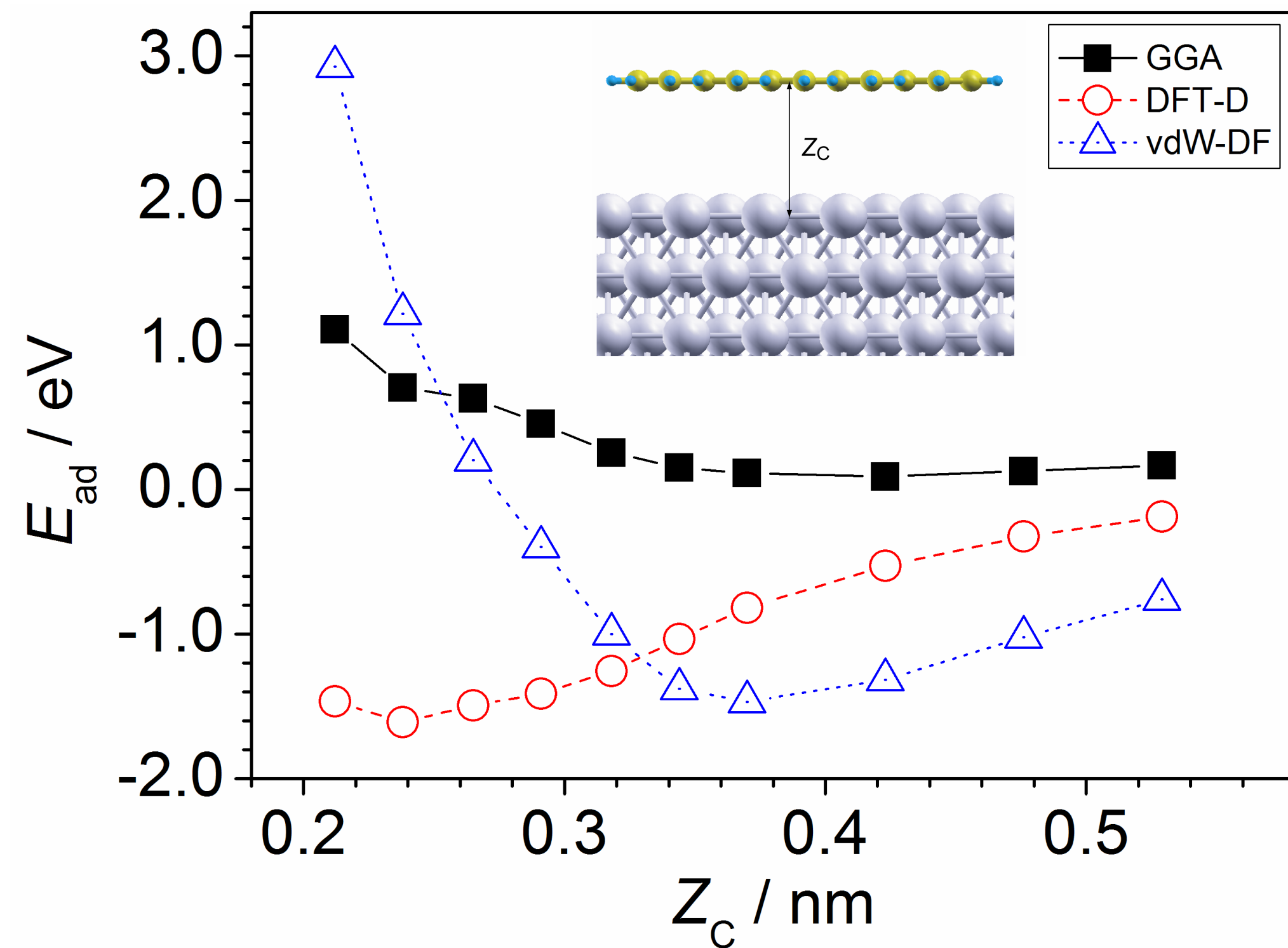
- vdW-DF can describe both intra- and interlayer bindings

Graphite

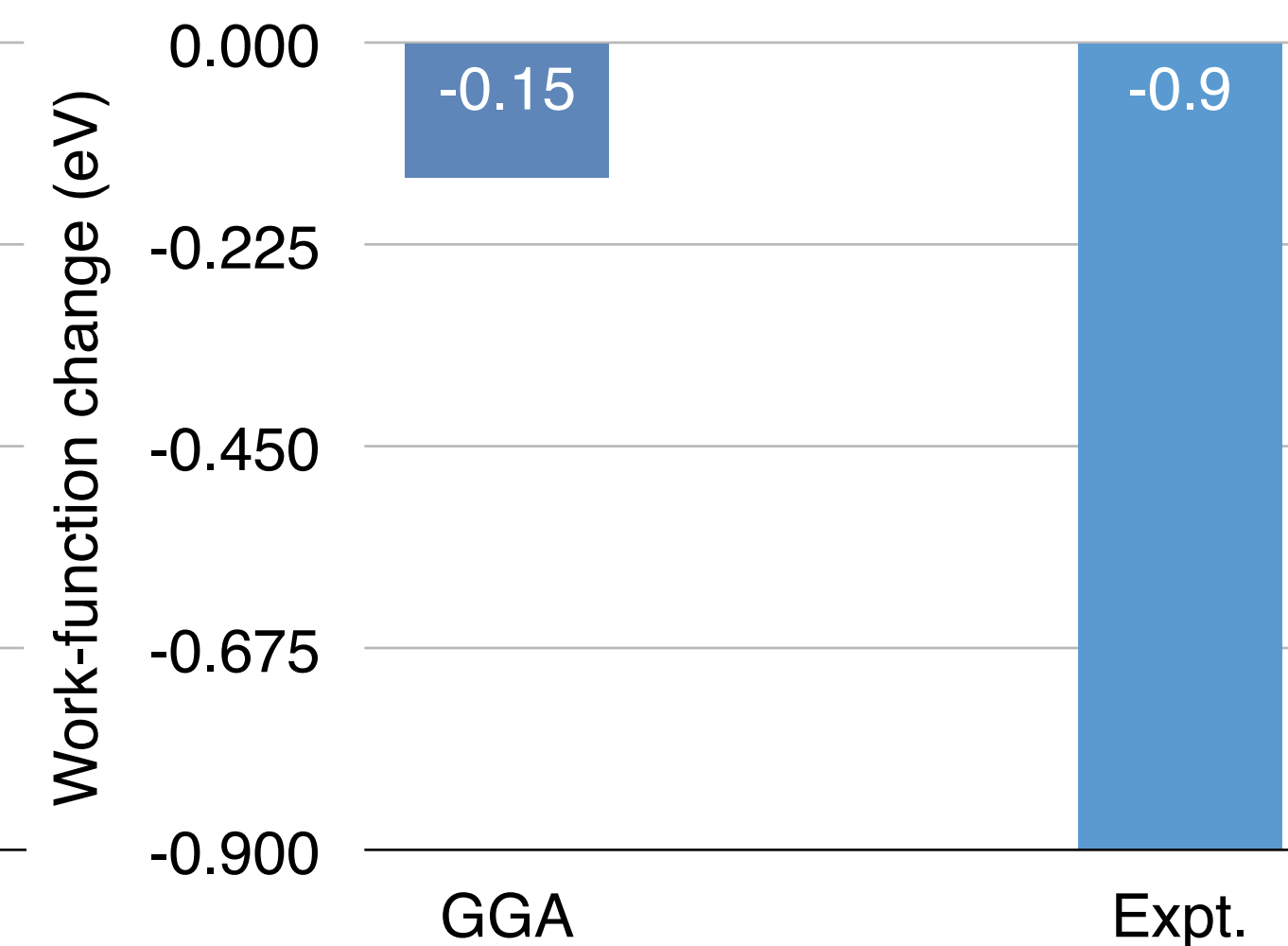
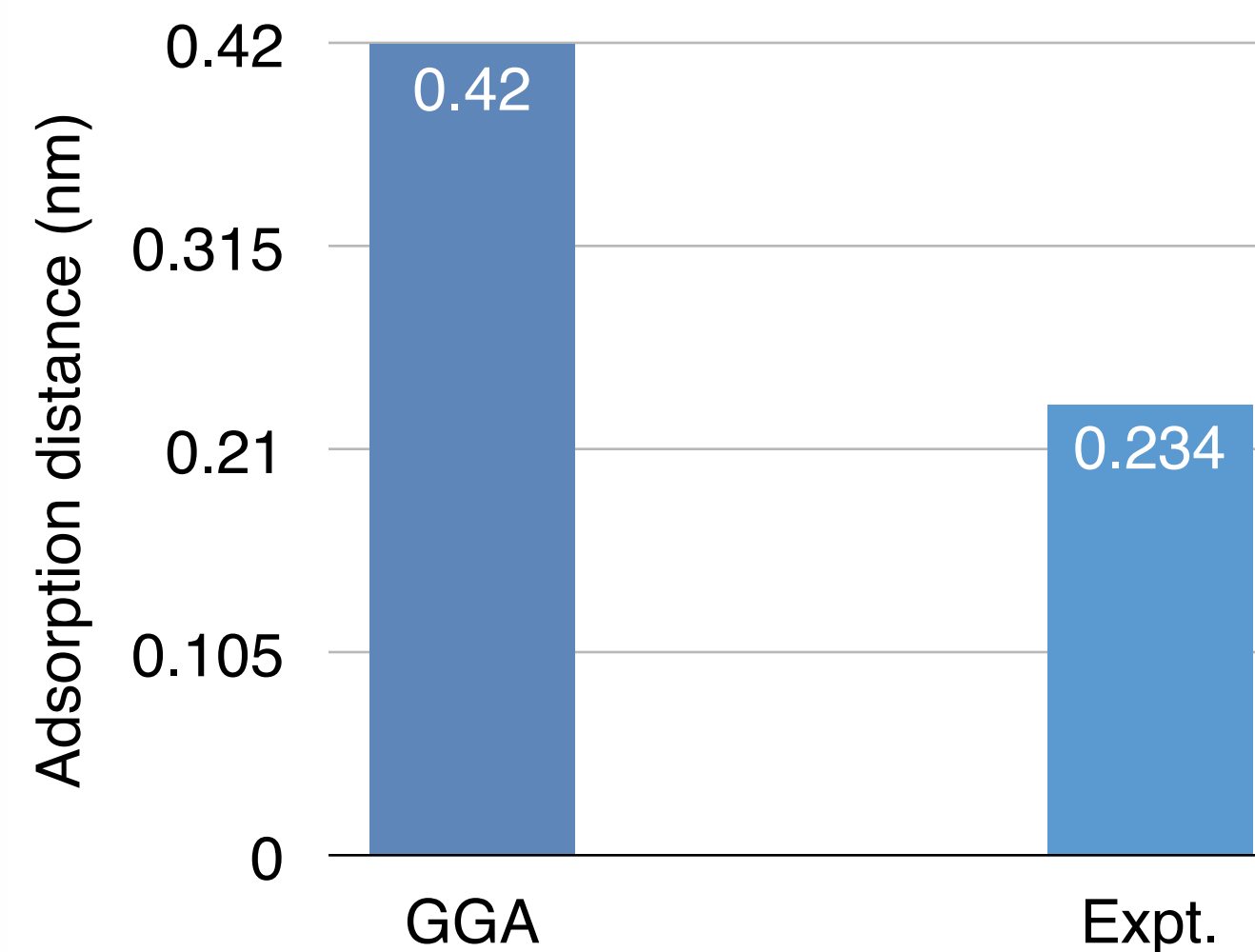
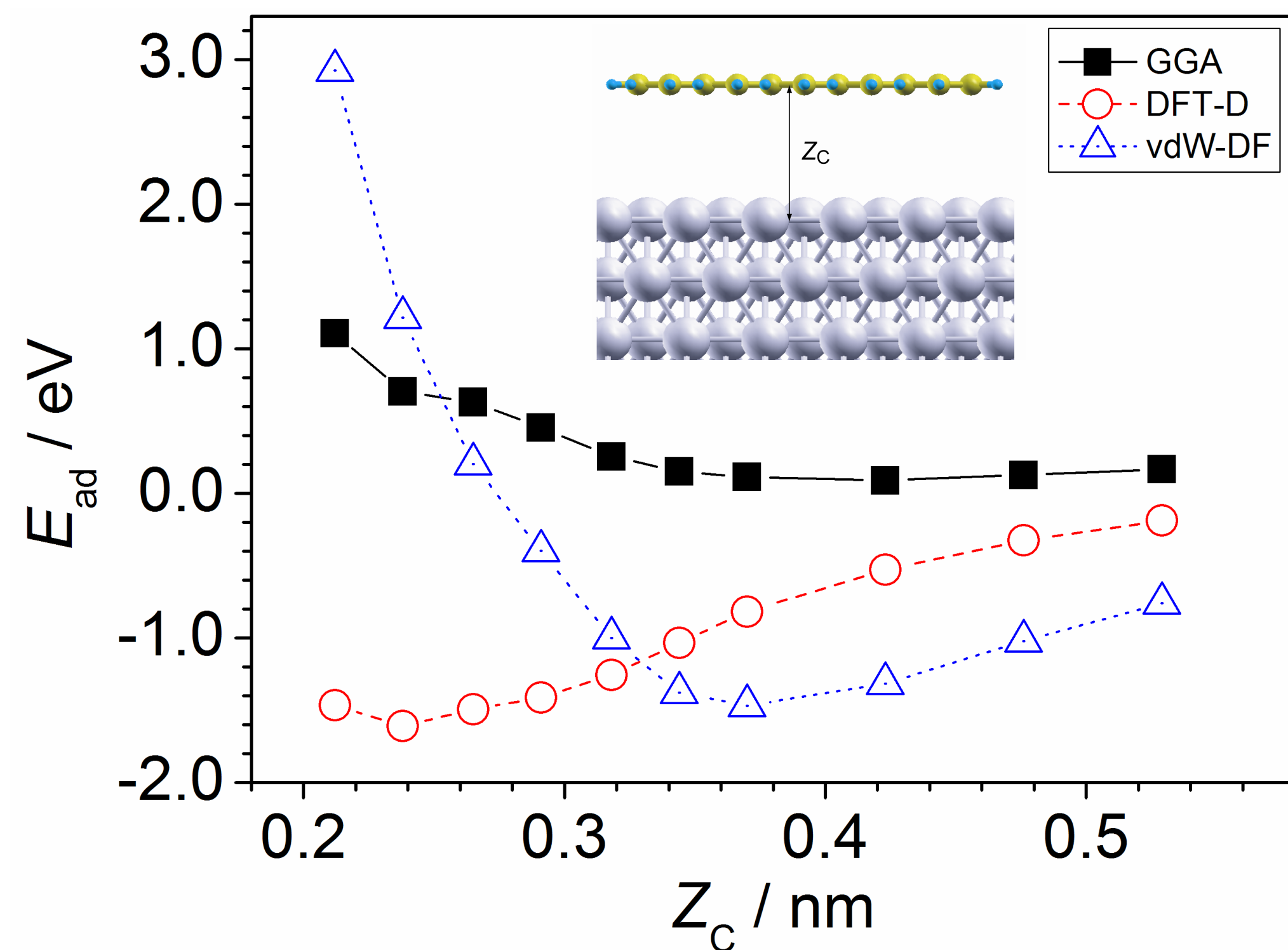


- vdW-DF can describe both intra- and interlayer bindings
- Binding energy is accurate, but interlayer distance is overestimated

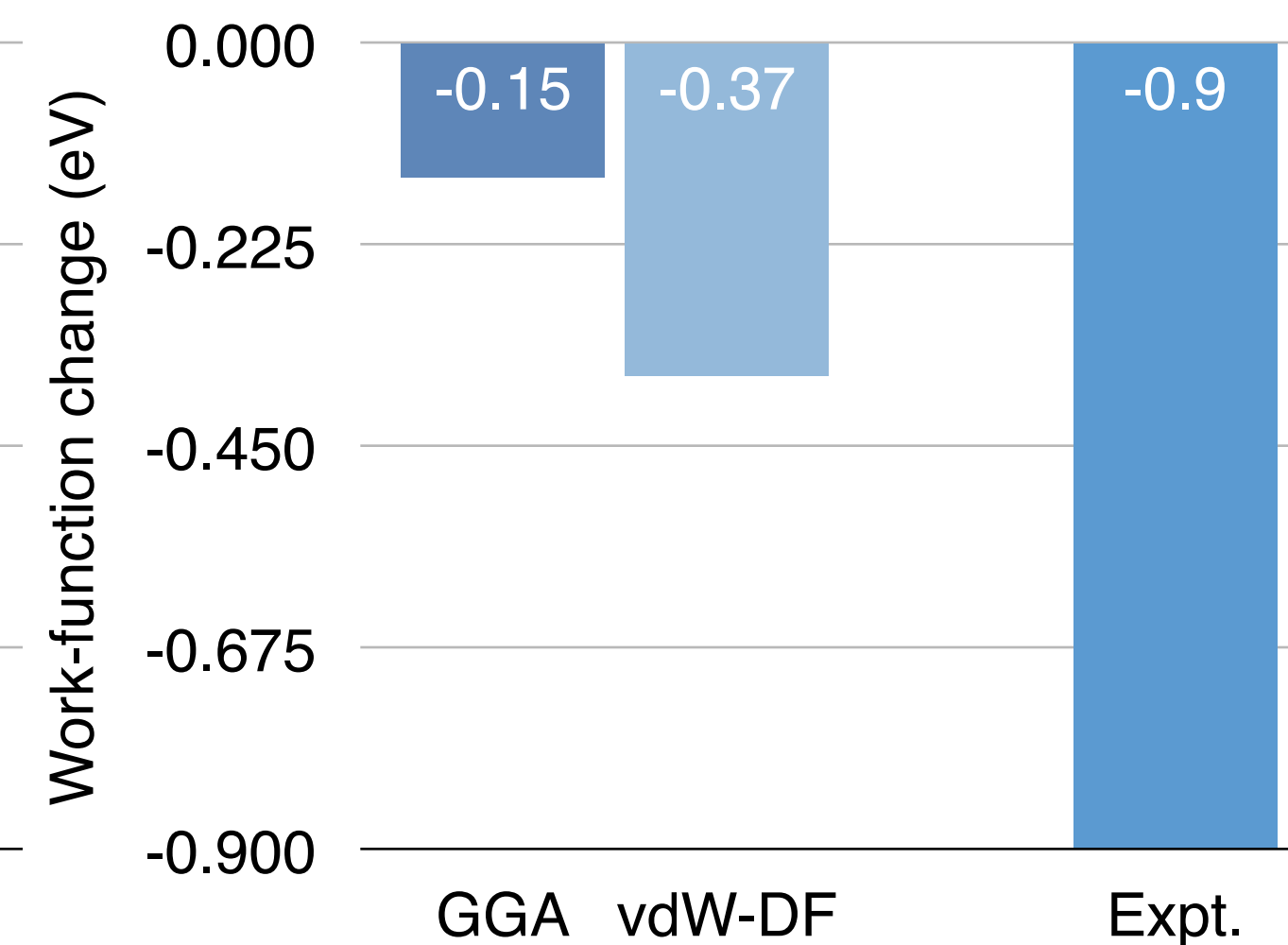
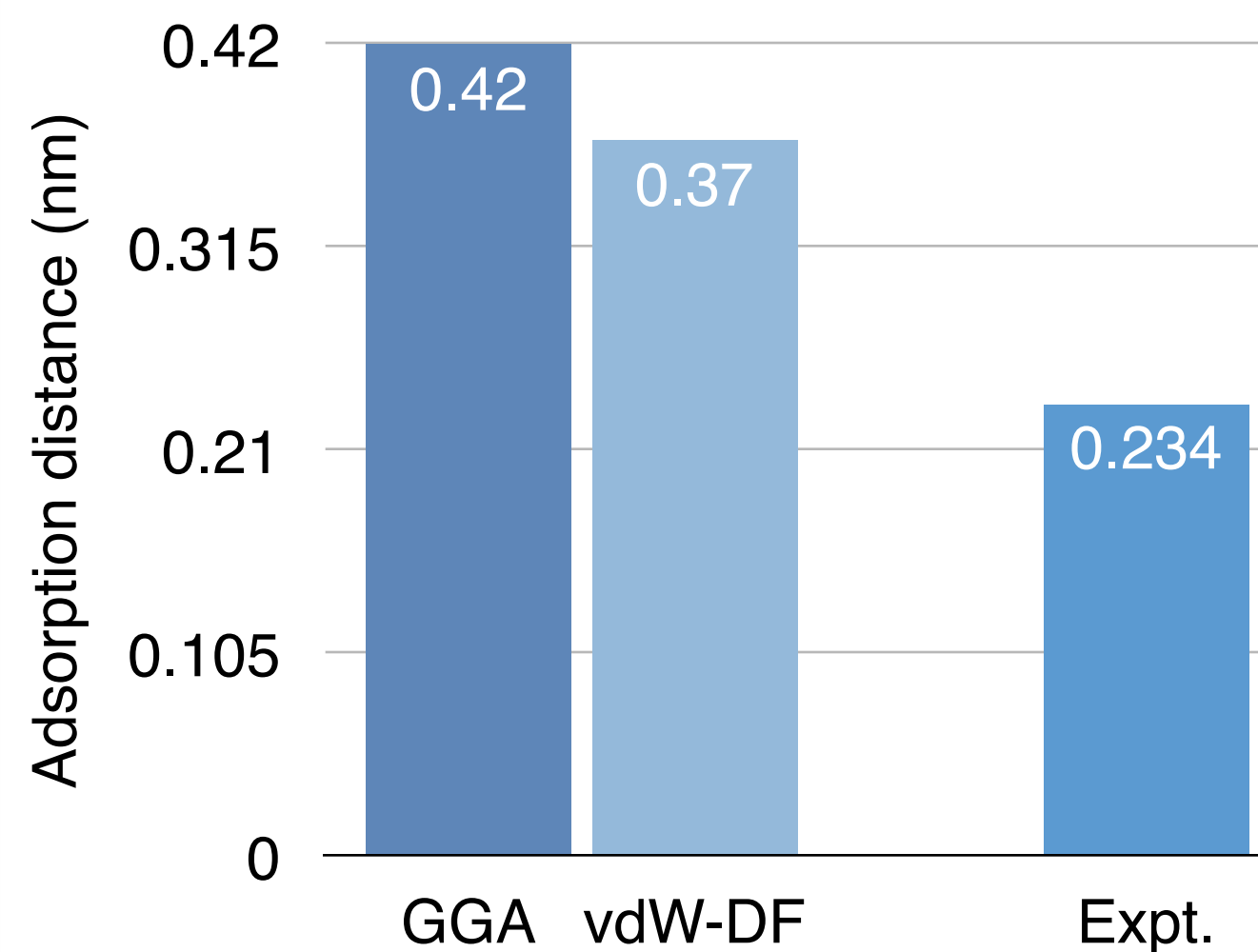
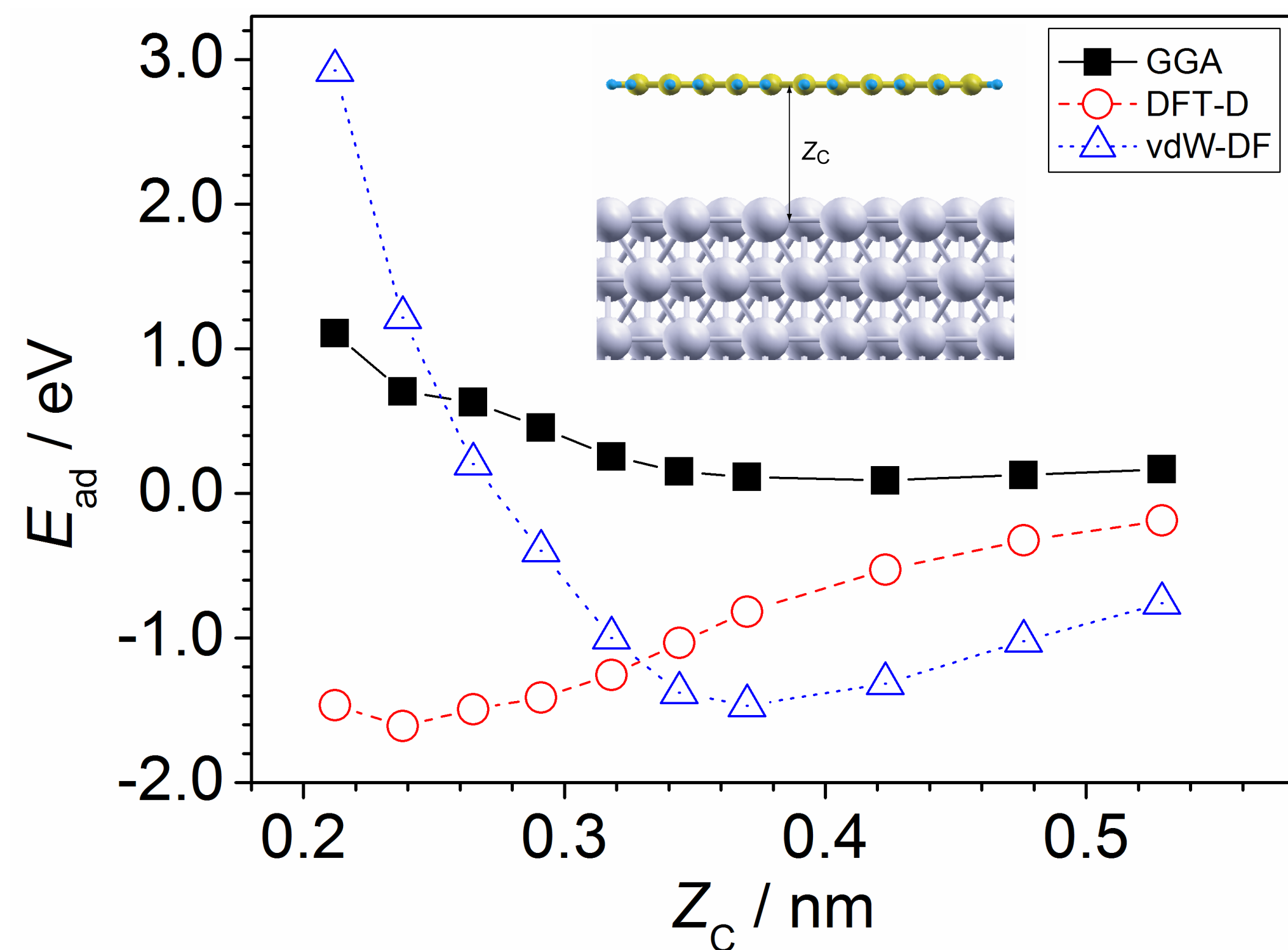
Work-function change in pentacene/Cu(111)



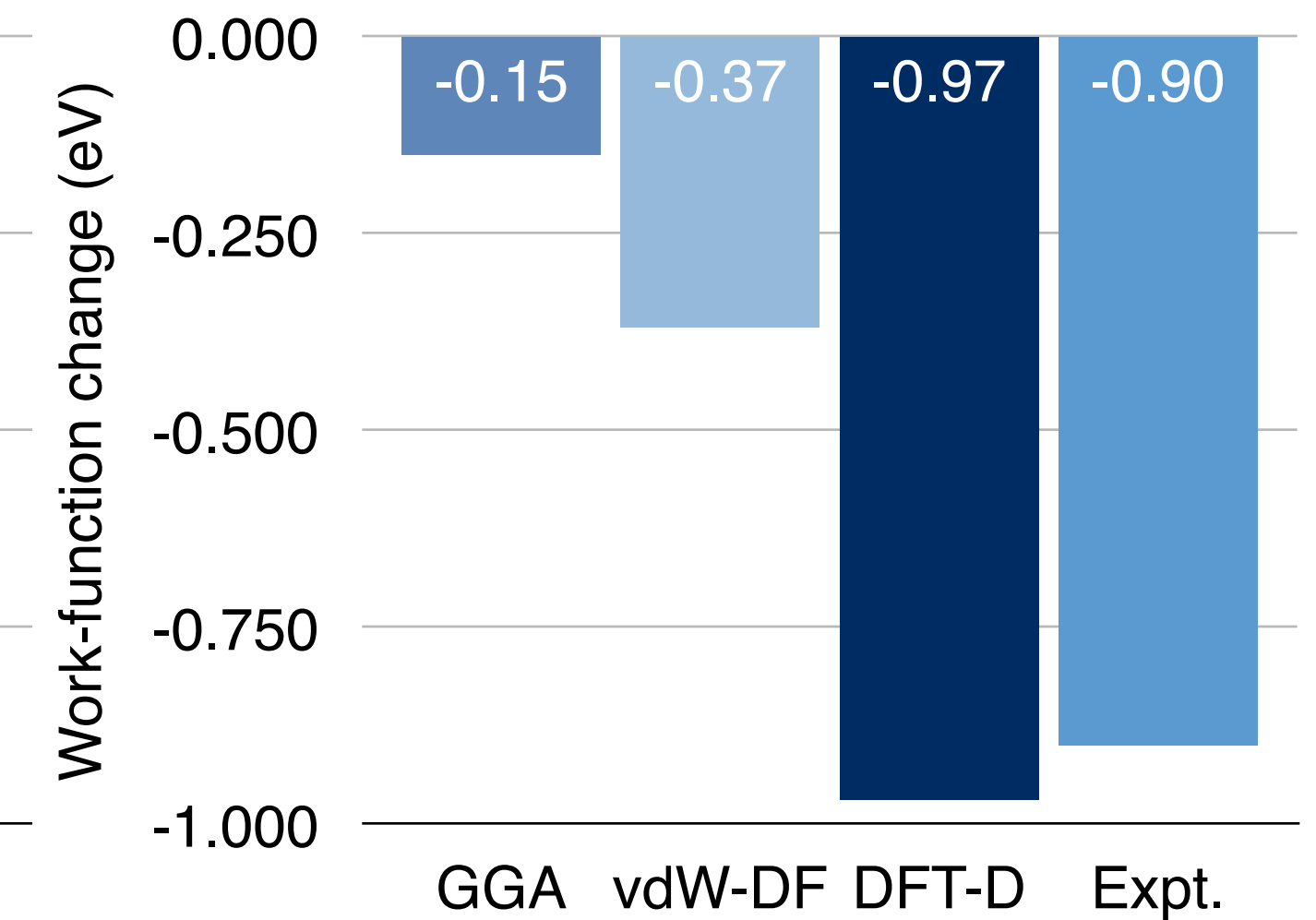
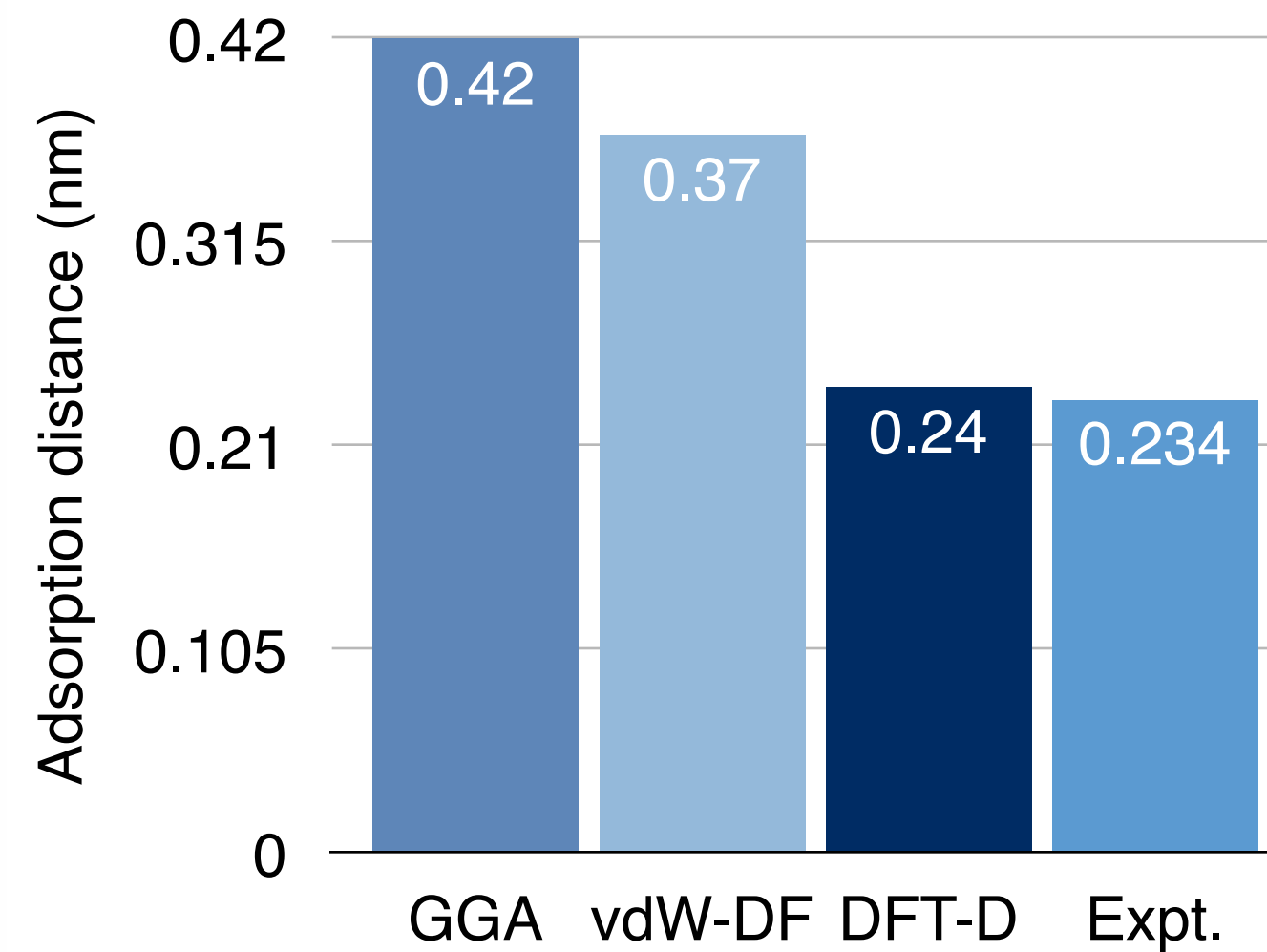
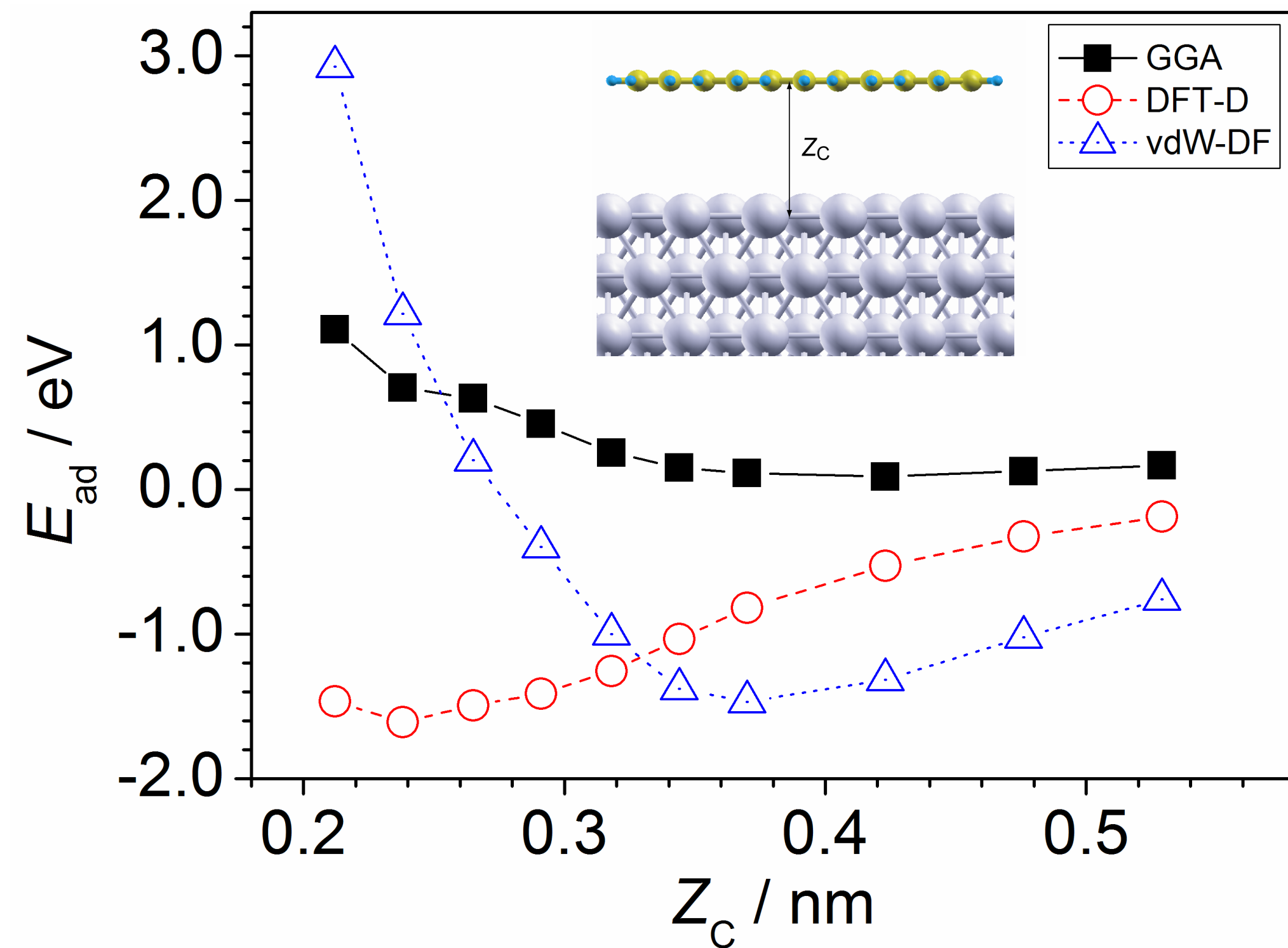
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Work-function change in pentacene/Cu(111)



Work-function change in pentacene/Cu(111)

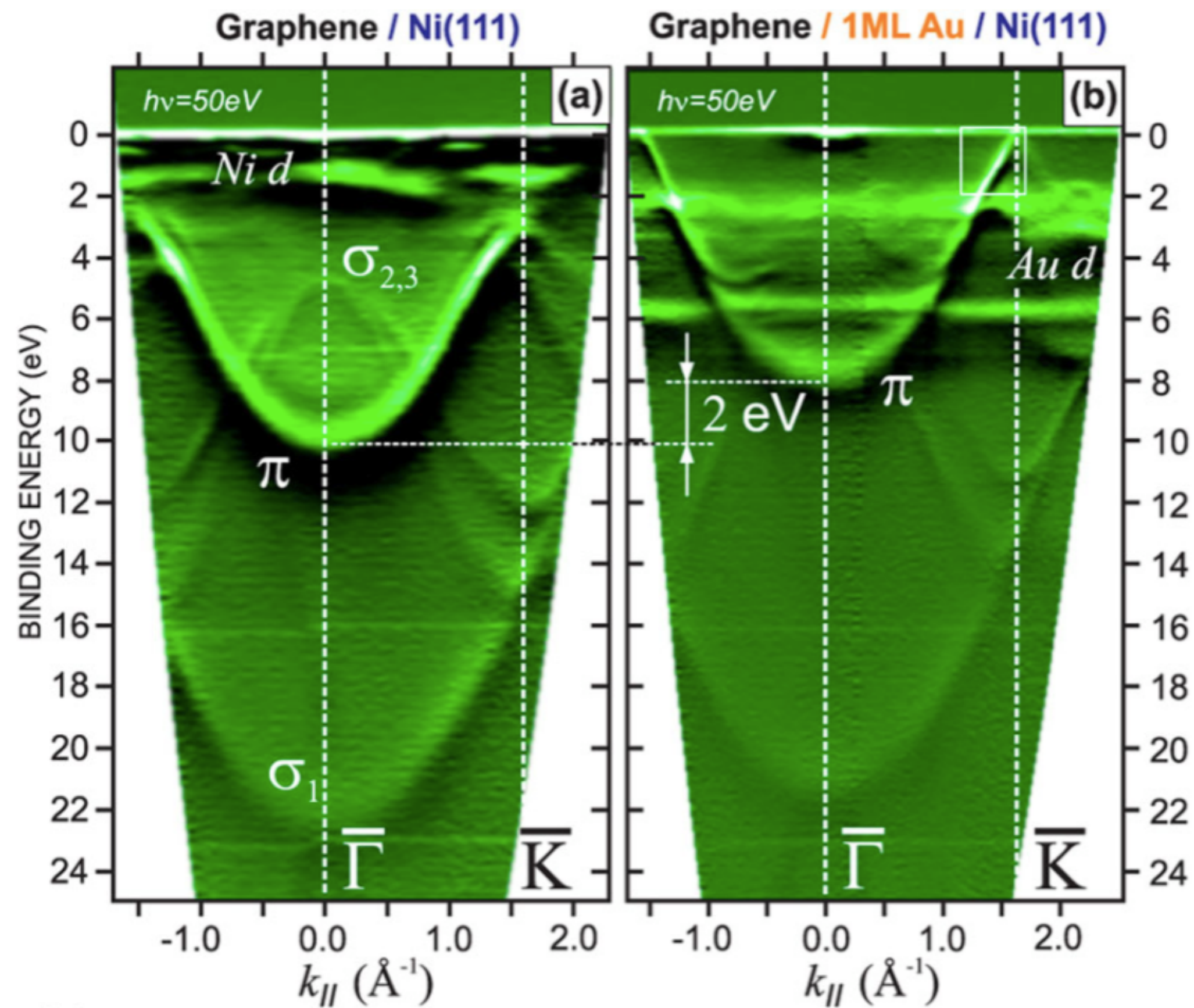


☹ Overestimated binding distance, leading to incorrect interface electronic structure

Band-gap opening at the K point of graphene on Ni(111)

Band-gap opening at the K point of graphene on Ni(111)

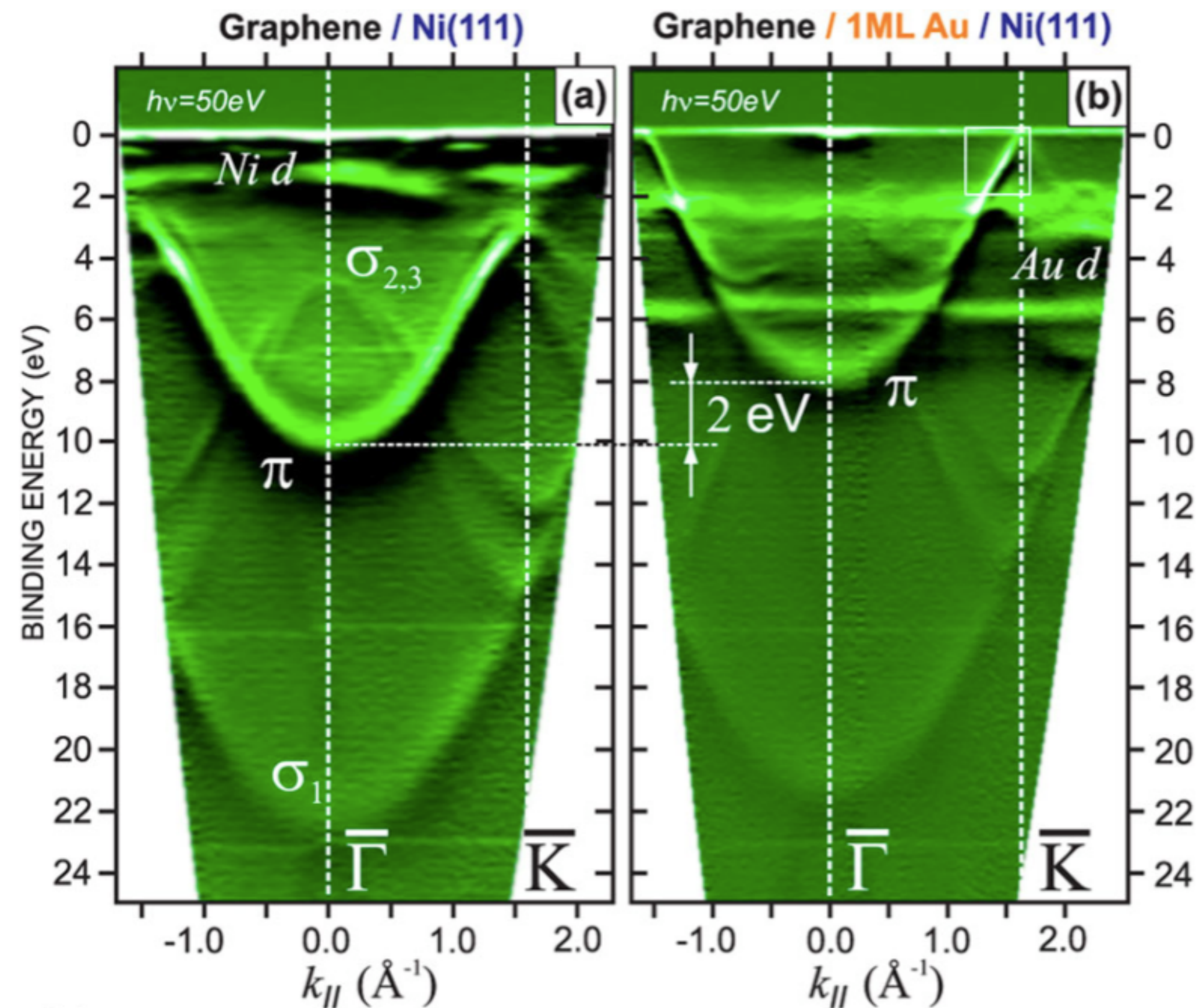
ARPES Expt.



Varykhalov *et al.*, *Phys. Rev. Lett.* **101**, 157601 (2008).

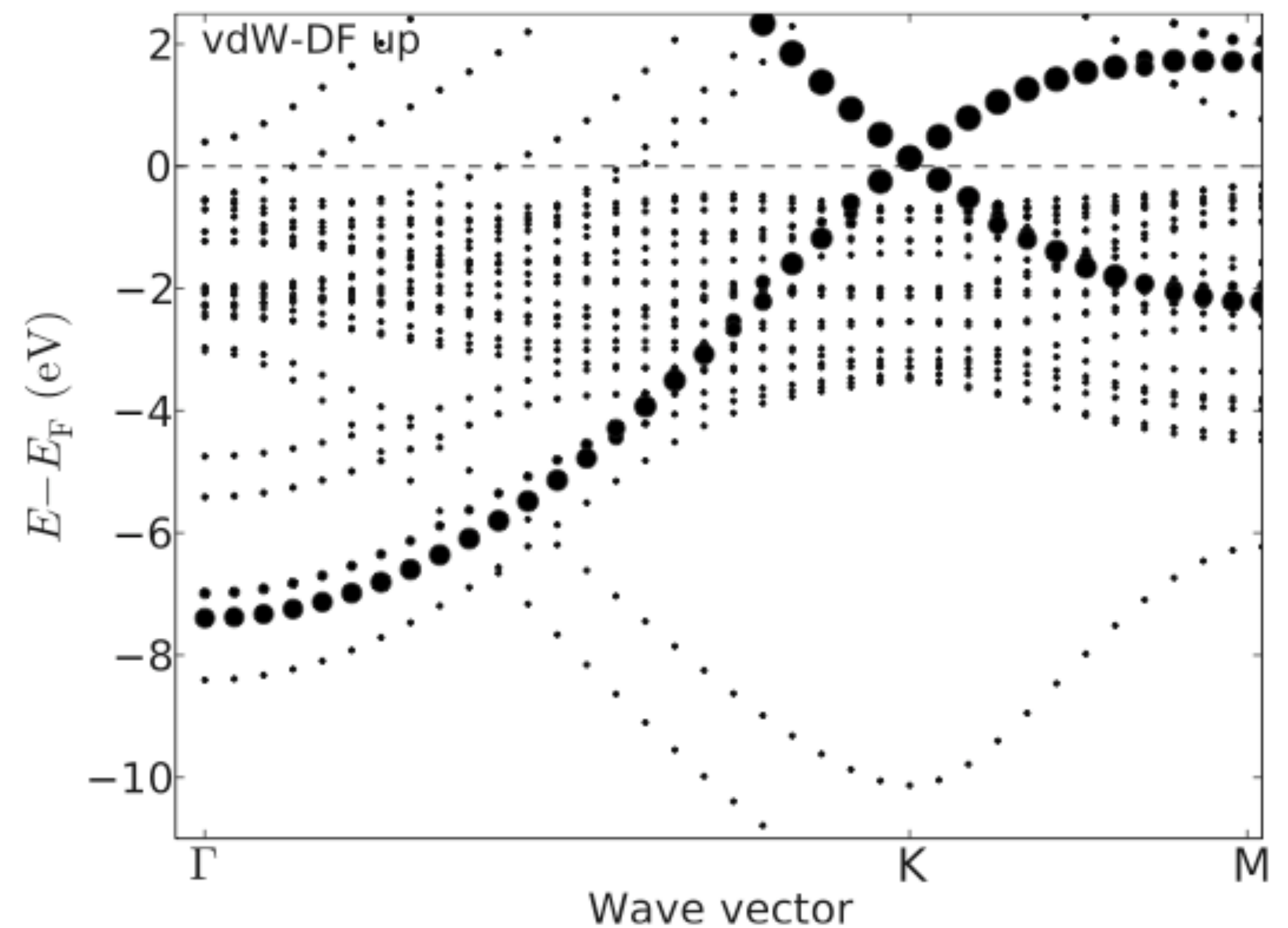
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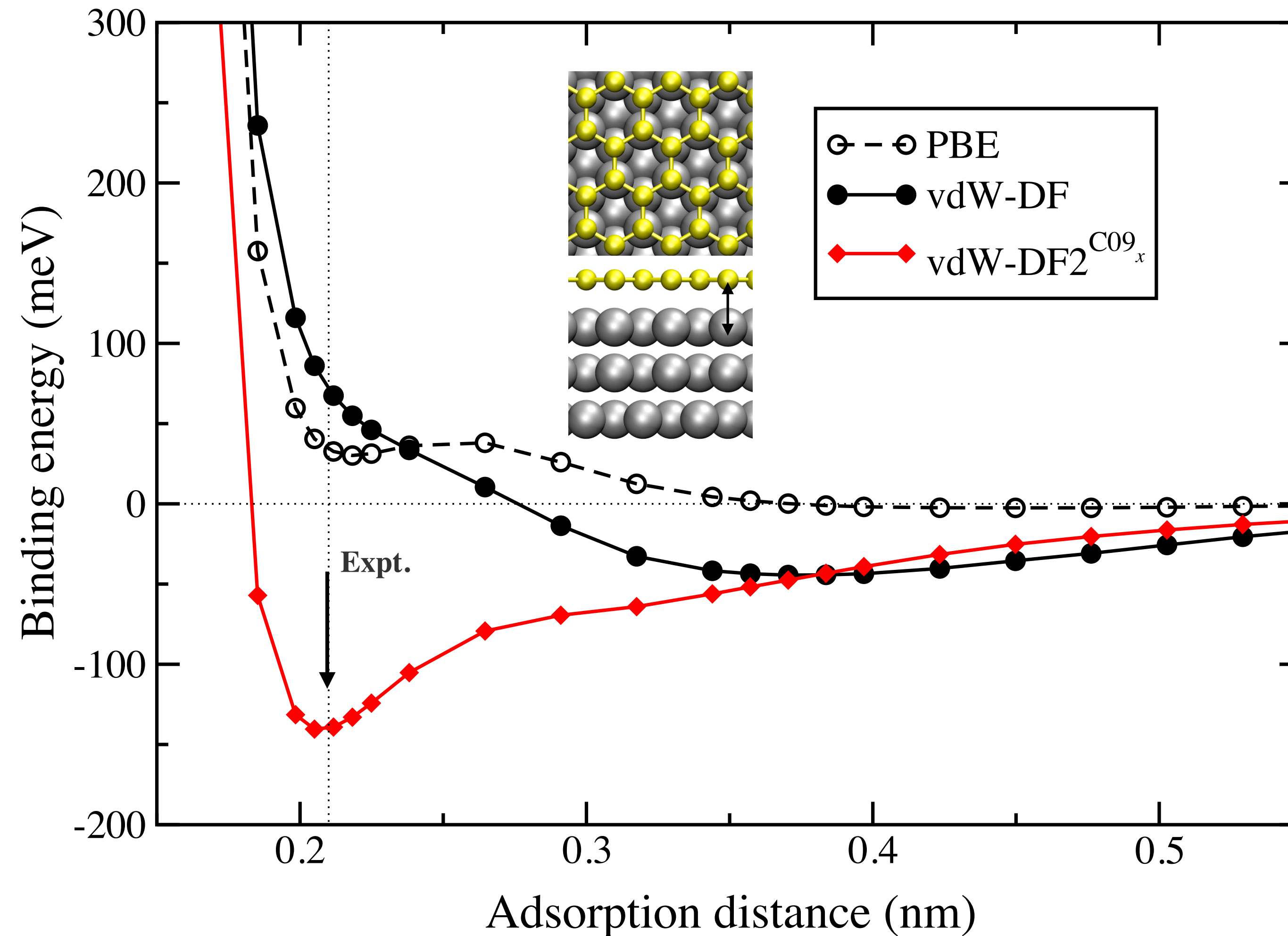
Varykhalov *et al.*, *Phys. Rev. Lett.* **101**, 157601 (2008).

vdW-DF Calc.



Vanin *et al.*, *Phys. Rev. B* **81**, 081408(R) (2010).

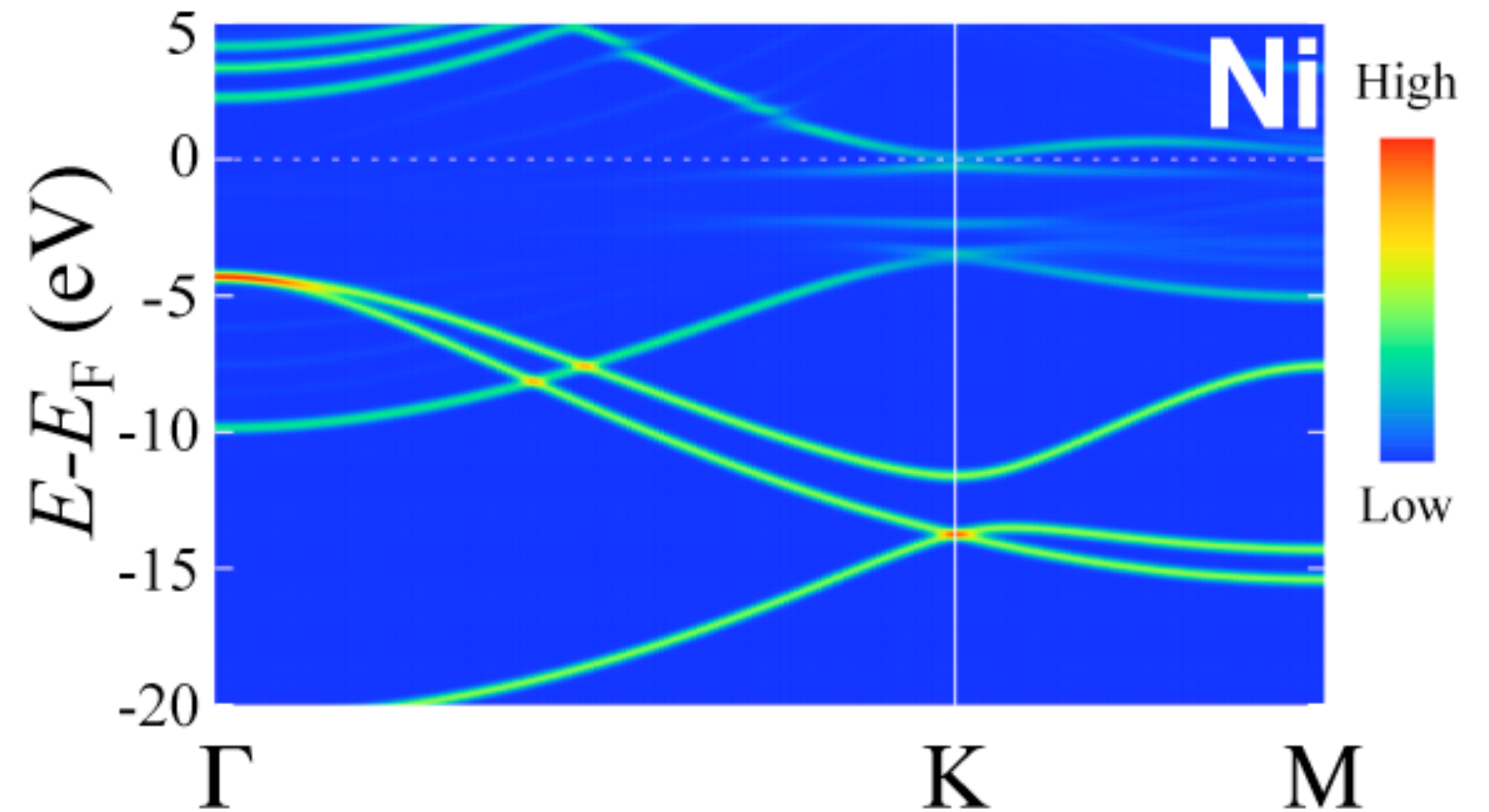
Binding energy curves of graphene on Ni(111)



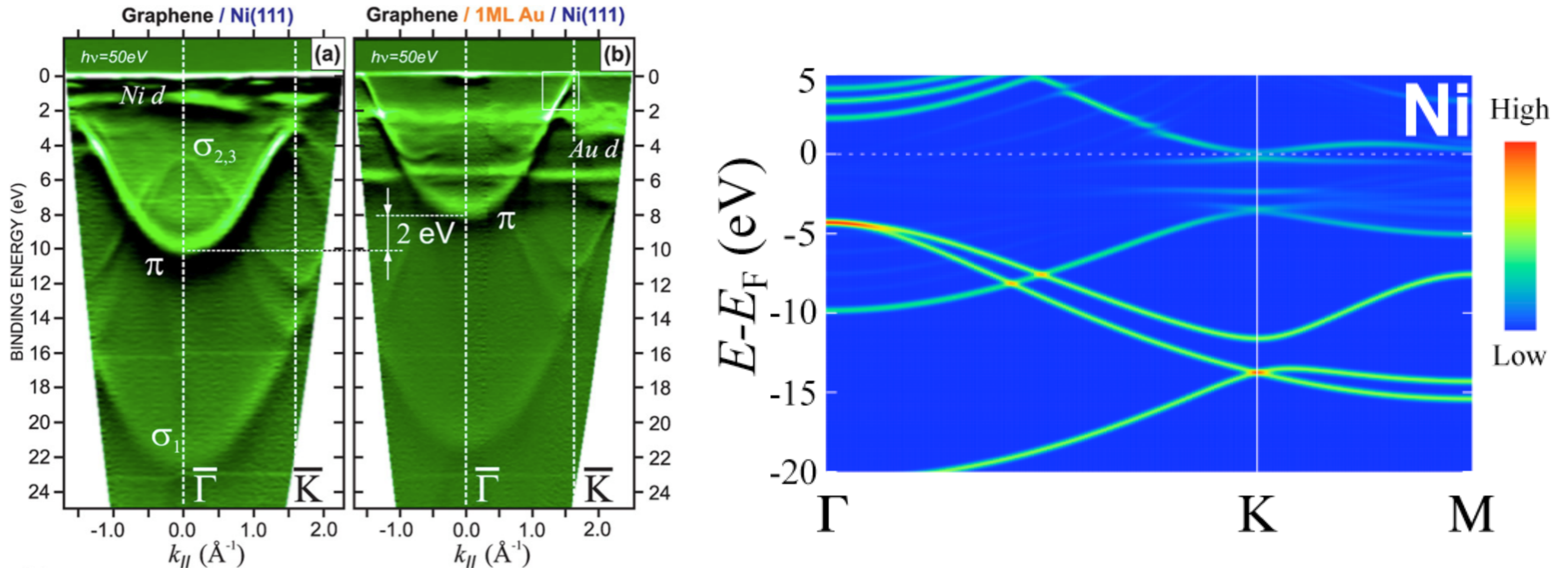
vdW-DF2-C09 improves the adsorption distance of graphene on Ni(111)

Band structure of graphene on Ni(111)

Band structure of graphene on Ni(111)



Band structure of graphene on Ni(111)



Vanin *et al.*, *Phys. Rev. B* **81**, 081408(R) (2010).

Calculated band structure with vdW-DF2-C09 agrees well with experiment when the correct adsorption distance is used

IH and M. Otani, *Phys. Rev. B* **82**, 153412 (2010).

van der Waals density functional (vdW-DF)

XC functional in vdW-DF

$$E_{\text{xc}}^{\text{vdW-DF}} = E_{\text{x}}^{\text{GGA}} + E_{\text{c}}^{\text{LDA}} + E_{\text{c}}^{\text{nl}}$$

$$E_{\text{c}}^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

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$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

- ☺ No empirical parameters
- ☺ Seamless description of covalent and van der Waals bondings

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- ☺ No empirical parameters
- ☺ Seamless description of covalent and van der Waals bondings
- ☹ No unique choice of local exchange and correlation to be accurate functional

van der Waals density functional (vdW-DF)

XC functional in vdW-DF

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van der Waals density functional (vdW-DF)

XC functional in vdW-DF

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$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

1st generation: vdW-DF(1), optB88-vdW, optB86b-vdW, vdW-DF-C09, vdW-DF-cx

2nd generation: vdW-DF2, vdW-DF2-C09, rev-vdW-DF2 (vdW-DF2-B86R)

3rd generation: vdW-DF3-opt1, vdW-DF3-opt2

van der Waals density functional (vdW-DF)

XC functional in rev-vdW-DF2 (vdW-DF2-B86R)

$$E_{\text{xc}}^{\text{rev-vdW-DF2}} = E_{\text{x}}^{\text{B86R}} + E_{\text{c}}^{\text{LDA}} + E_{\text{c,nl}}^{\text{vdW-DF2}}$$

$$E_{\text{c}}^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

1st generation: vdW-DF(1), optB88-vdW, optB86b-vdW, vdW-DF-C09, vdW-DF-cx

2nd generation: vdW-DF2, vdW-DF2-C09, rev-vdW-DF2 (vdW-DF2-B86R)

3rd generation: vdW-DF3-opt1, vdW-DF3-opt2

Hamada, *Phys. Rev. B* **89**, 121103 (2014).

Revised B86b exchange functional (B86R)

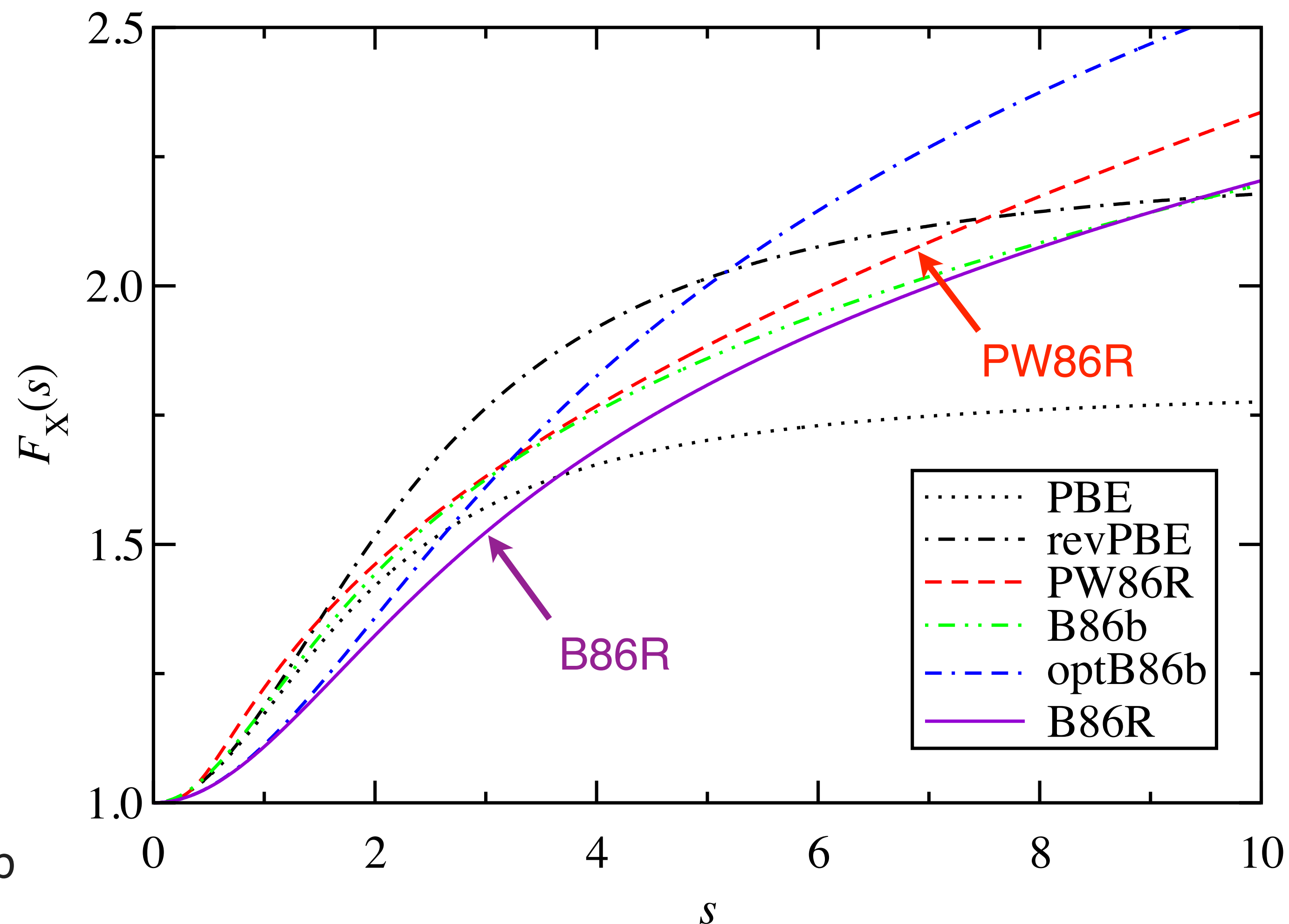
$$E_X^{\text{GGA}} = \int d\mathbf{r} n(\mathbf{r}) \epsilon_X^{\text{unif}}(n) F_X(s)$$

$$F_X^{\text{B86b}}(s) = 1 + \frac{\mu s^2}{(1 + \mu s^2/\kappa)^{4/5}}$$

$$s = |\nabla n|/(2k_F n)$$

Becke, *J. Chem. Phys.* **85**, 7184 (1986).

- μ and κ are determined so as to match
 - GEA at the slowly varying density limit
(important for covalent bonding)
 - Large density gradient behavior in the B86b
(important for the vdW bonding)



Revised B86b exchange functional (B86R)

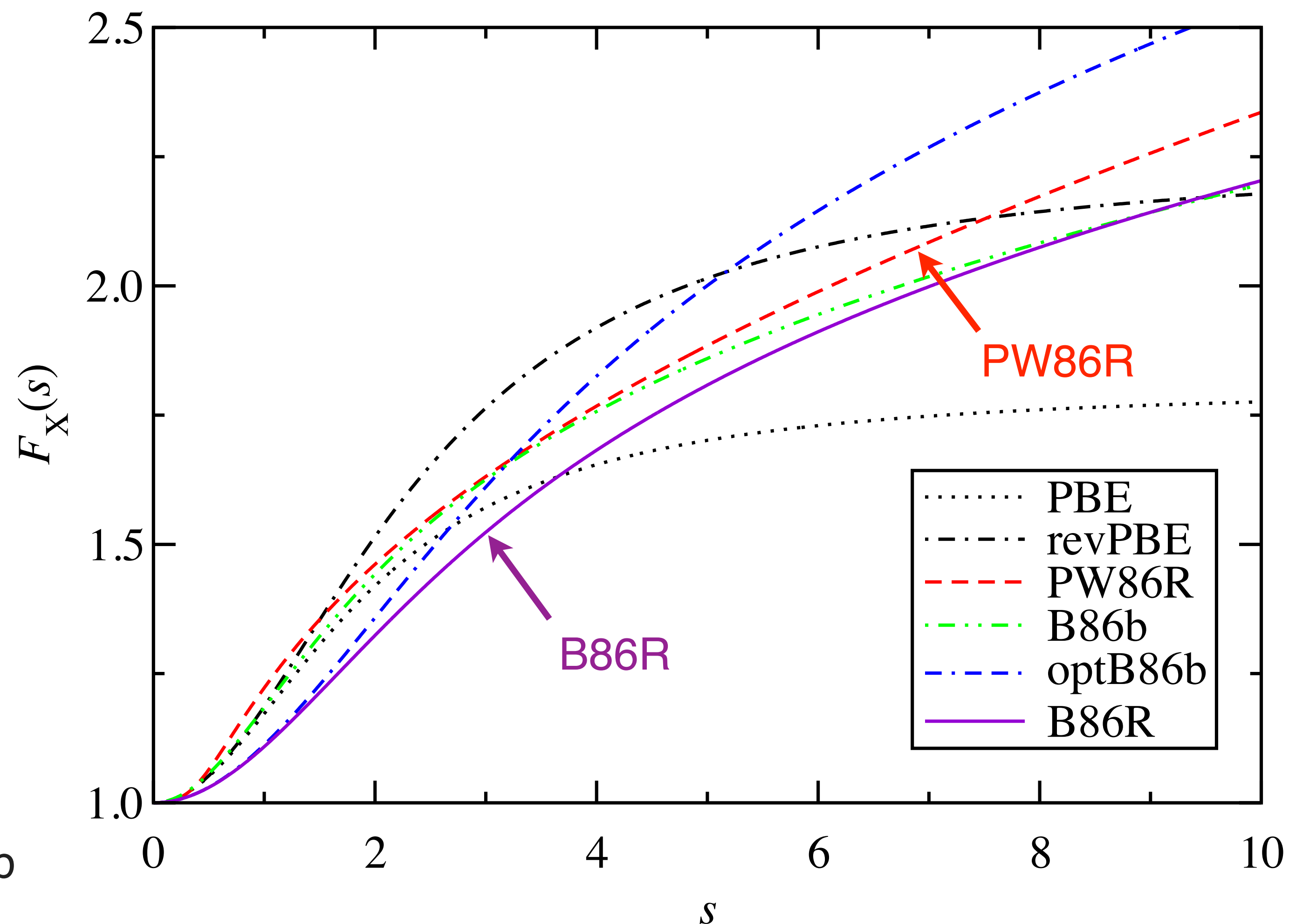
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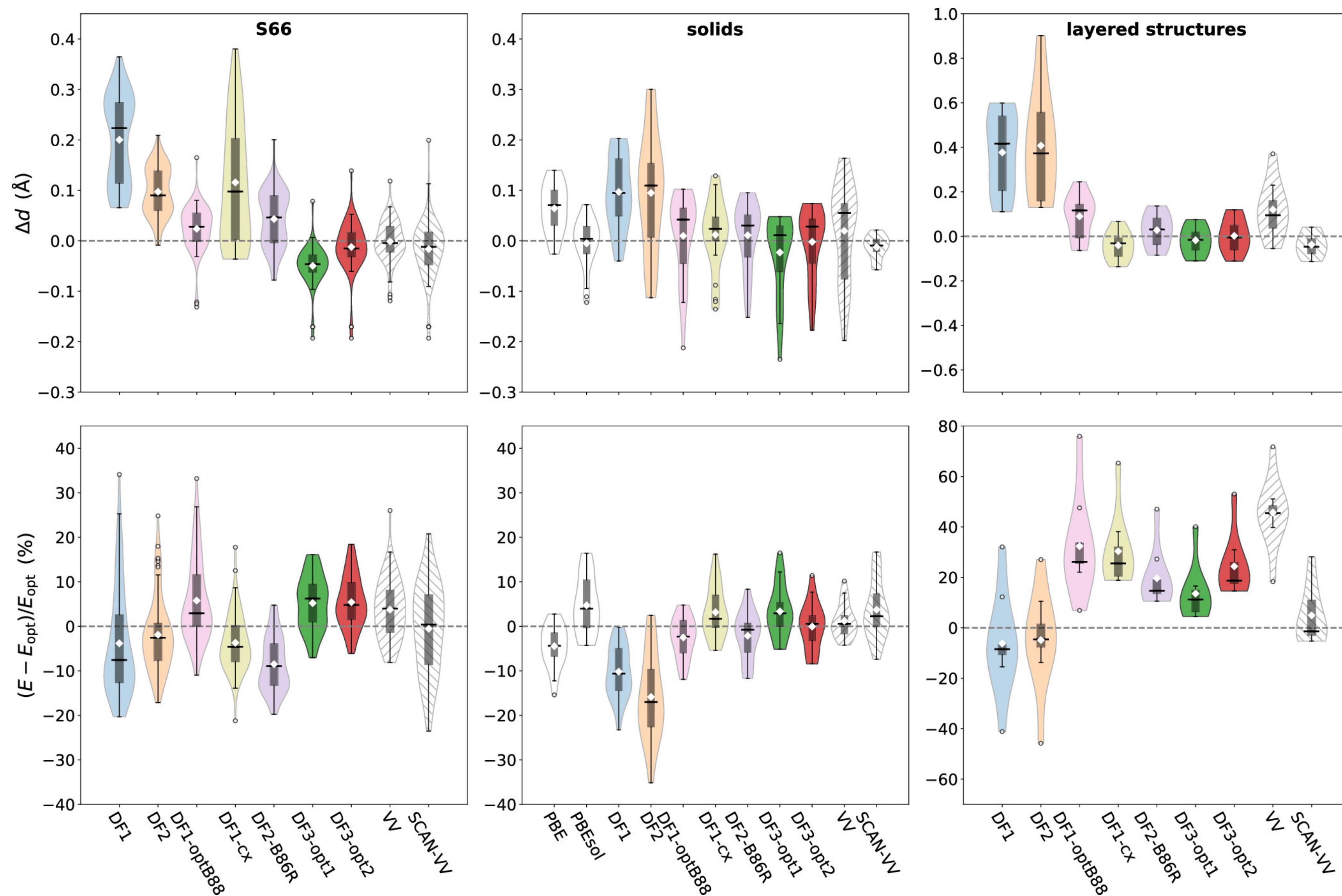
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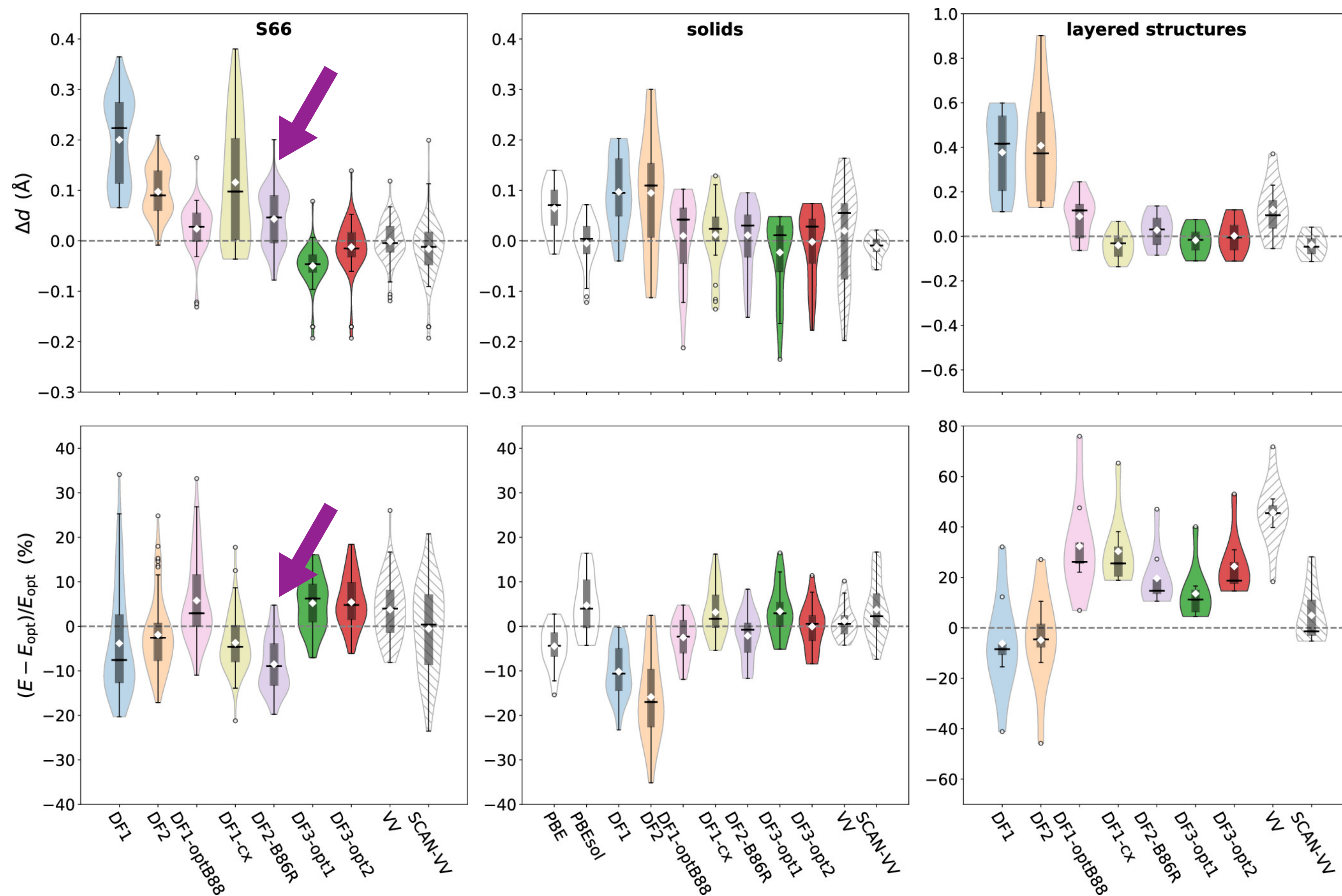
Outline

- Background: Dispersion forces in density functional theory
- Theory and application of van der Waals density functional (vdW-DF)
 - **Benchmark for molecules, materials, and interfaces**
 - Case study
 - Layered materials
 - Organic crystals
 - Recent development
- Summary

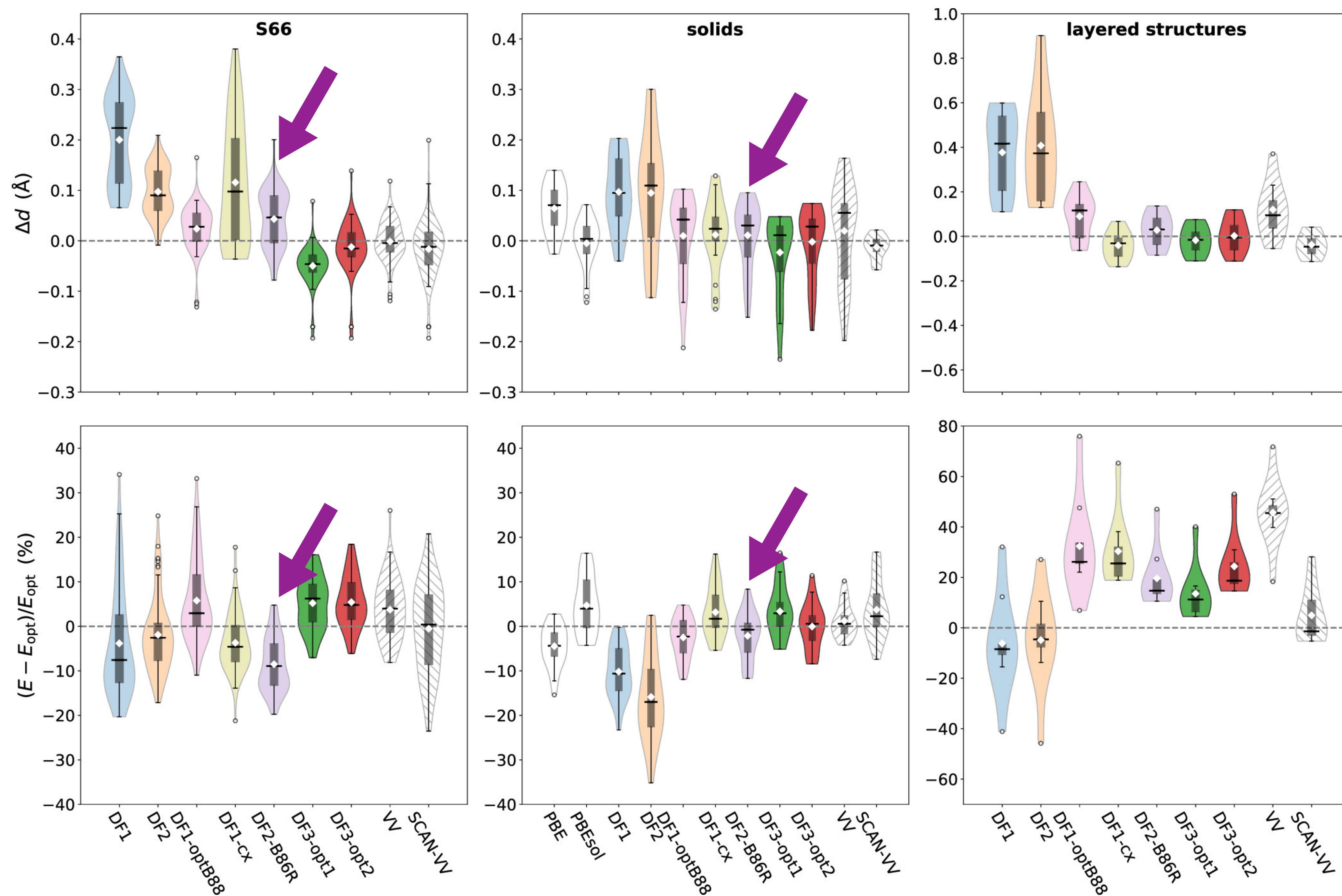
Comparison of vdW-DFs for molecules and solids



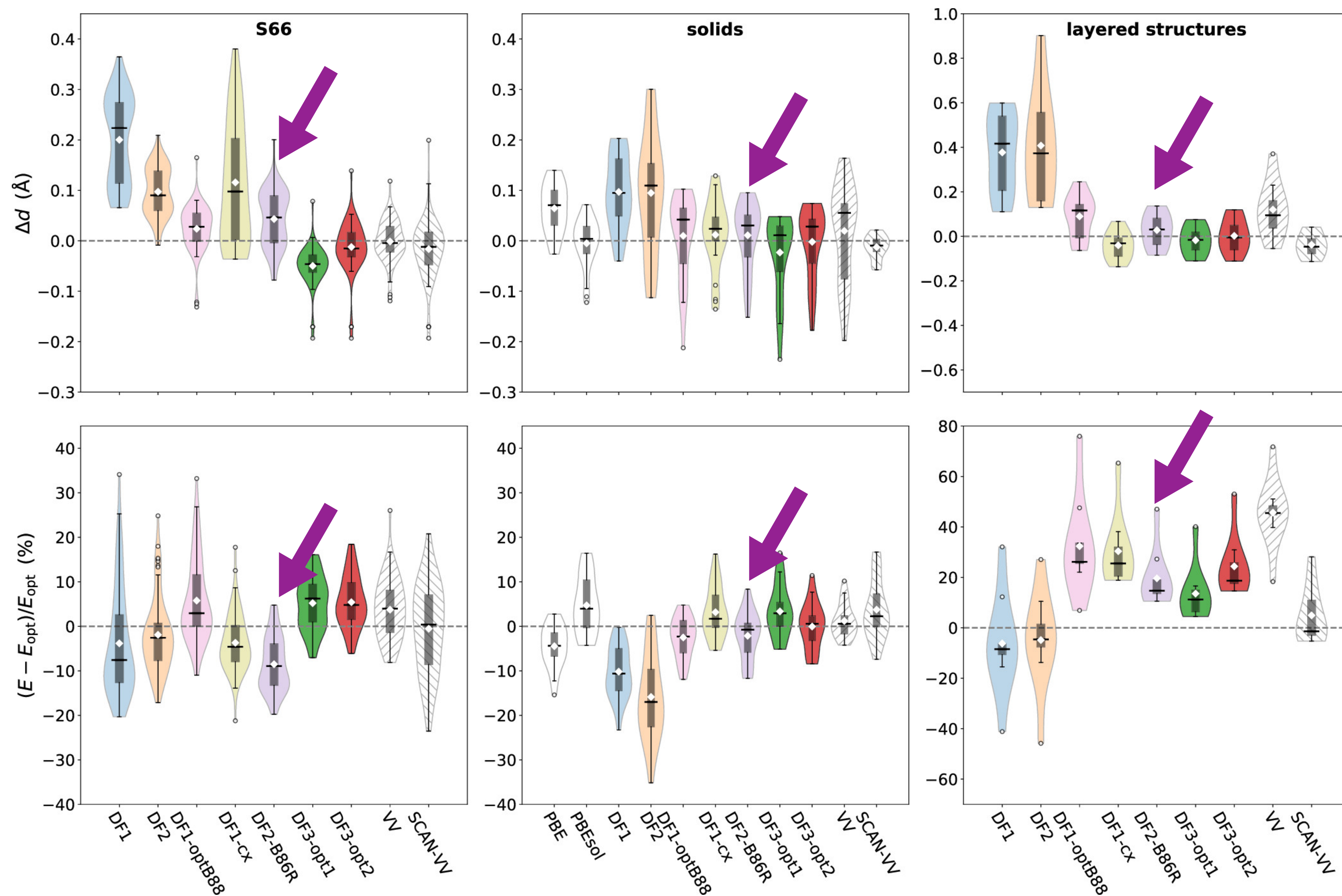
Comparison of vdW-DFs for molecules and solids



Comparison of vdW-DFs for molecules and solids

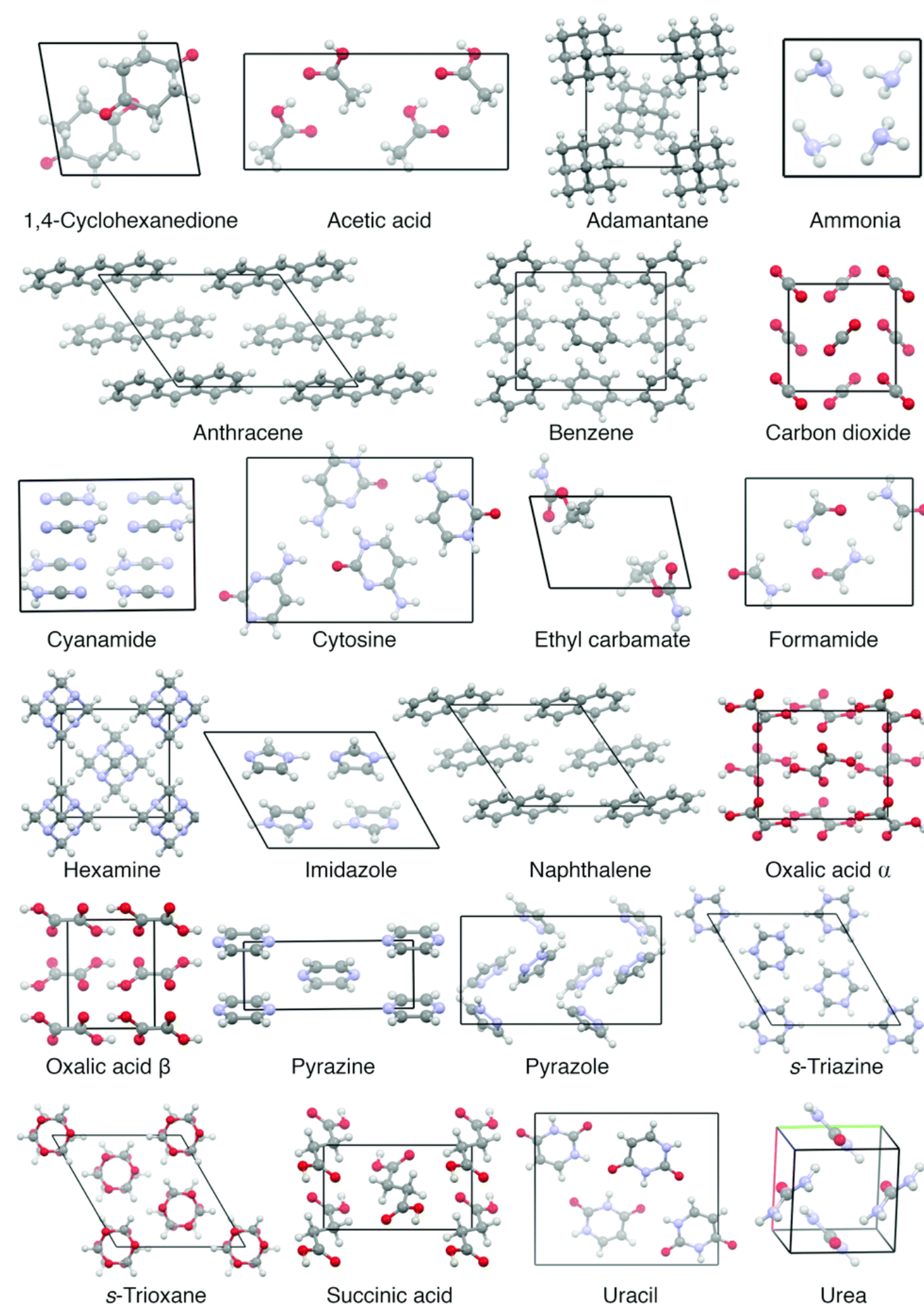


Comparison of vdW-DFs for molecules and solids



Comparison of vdW-DFs for molecular crystals

X23 dataset

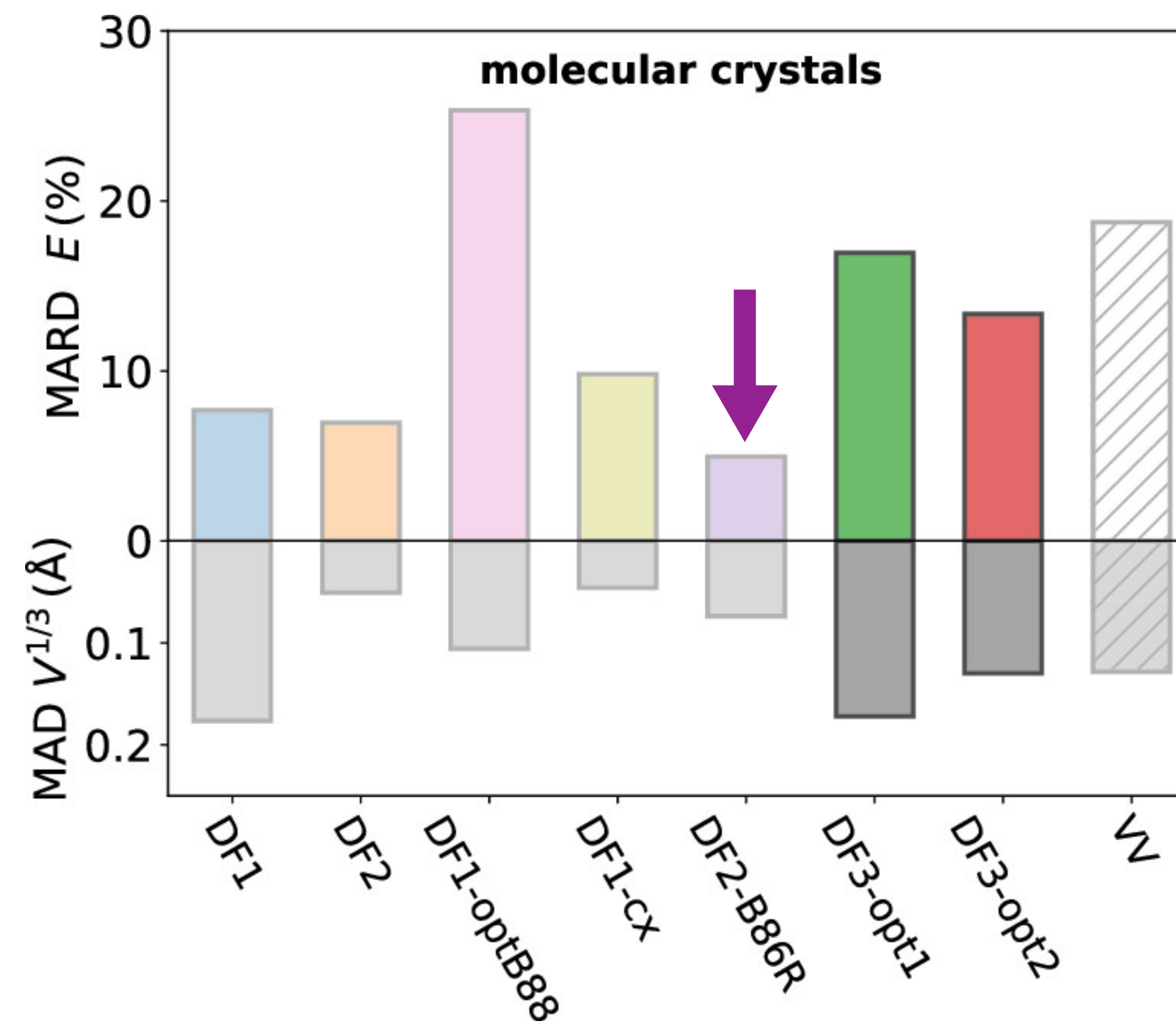
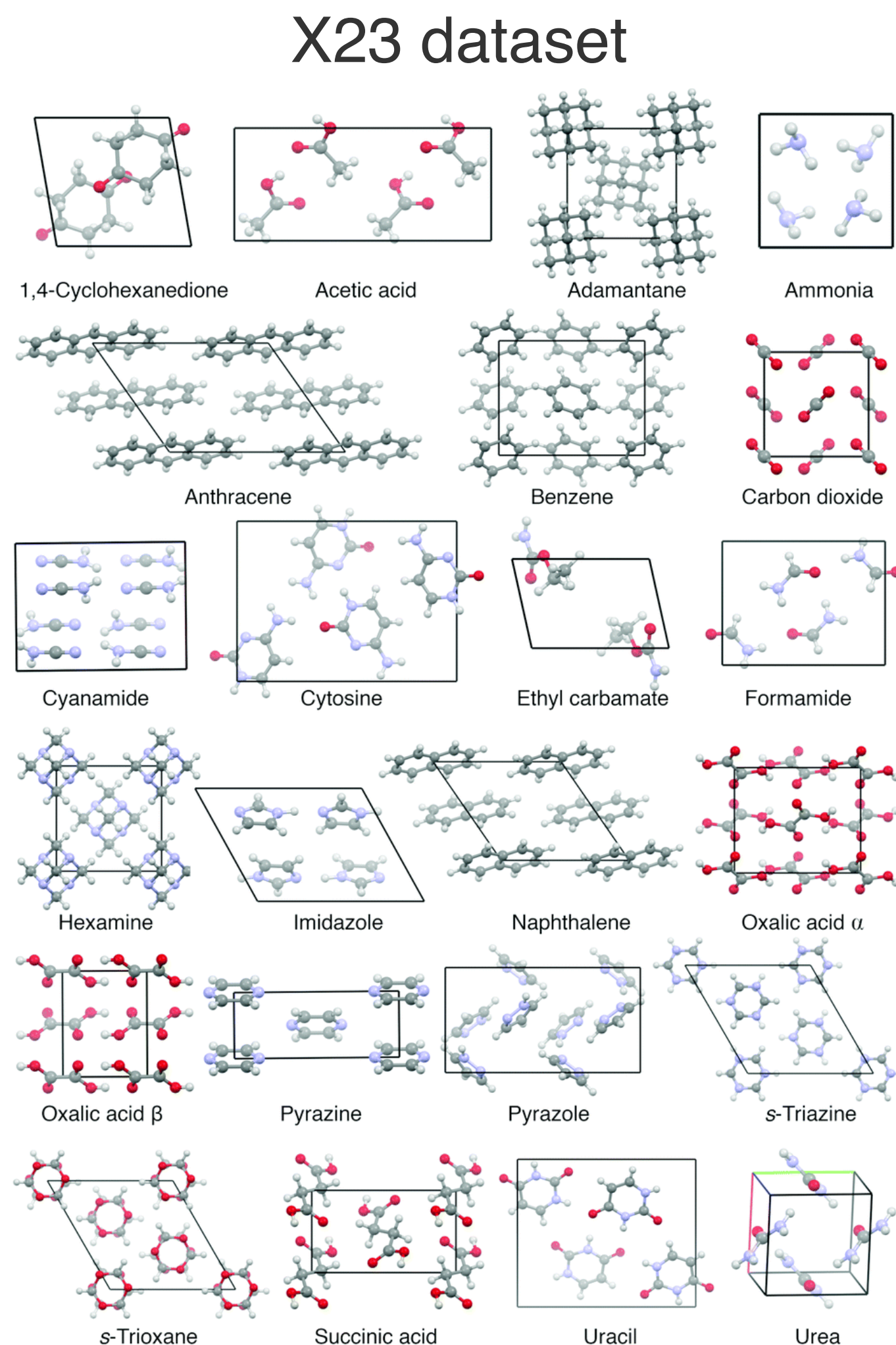


Reilly and Tkatchenko, *J. Chem. Phys.* **139**, 024705 (2013);

G. A. Dolgonos, J. Hoja, D. Boese, *Phys. Chem. Chem. Phys.* **21**, 24333 (2019).

Chakraborty, Berland, Thonhauser, *J. Chem. Theory Comput.* **16**, 5893 (2020).

Comparison of vdW-DFs for molecular crystals



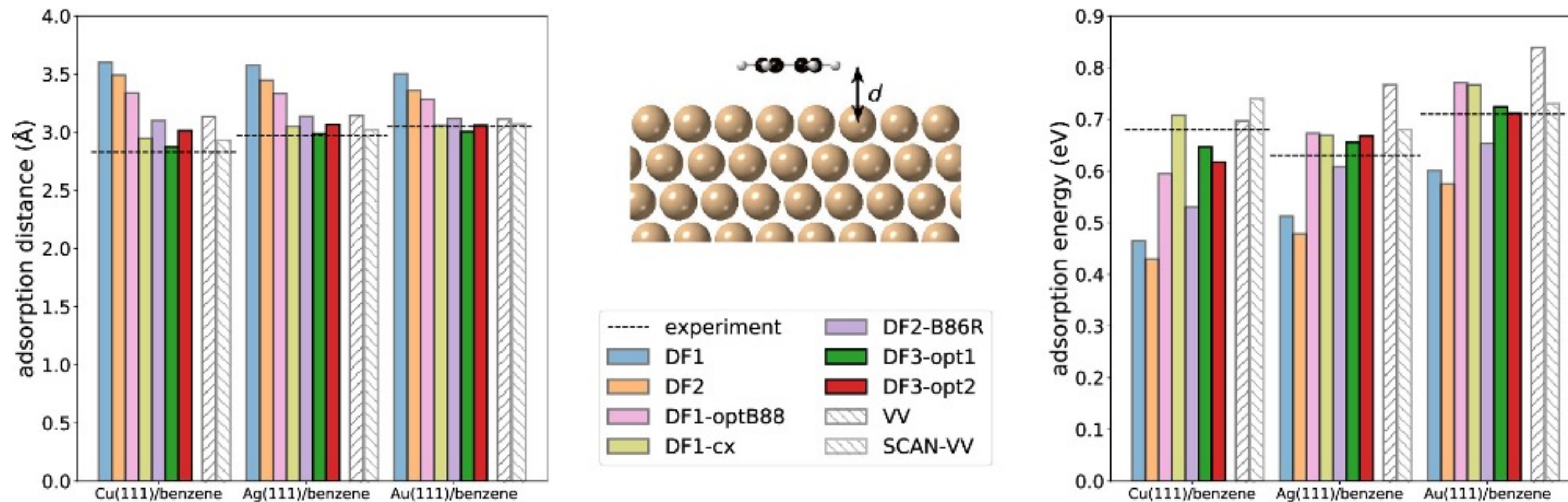
Reilly and Tkatchenko, *J. Chem. Phys.* **139**, 024705 (2013);

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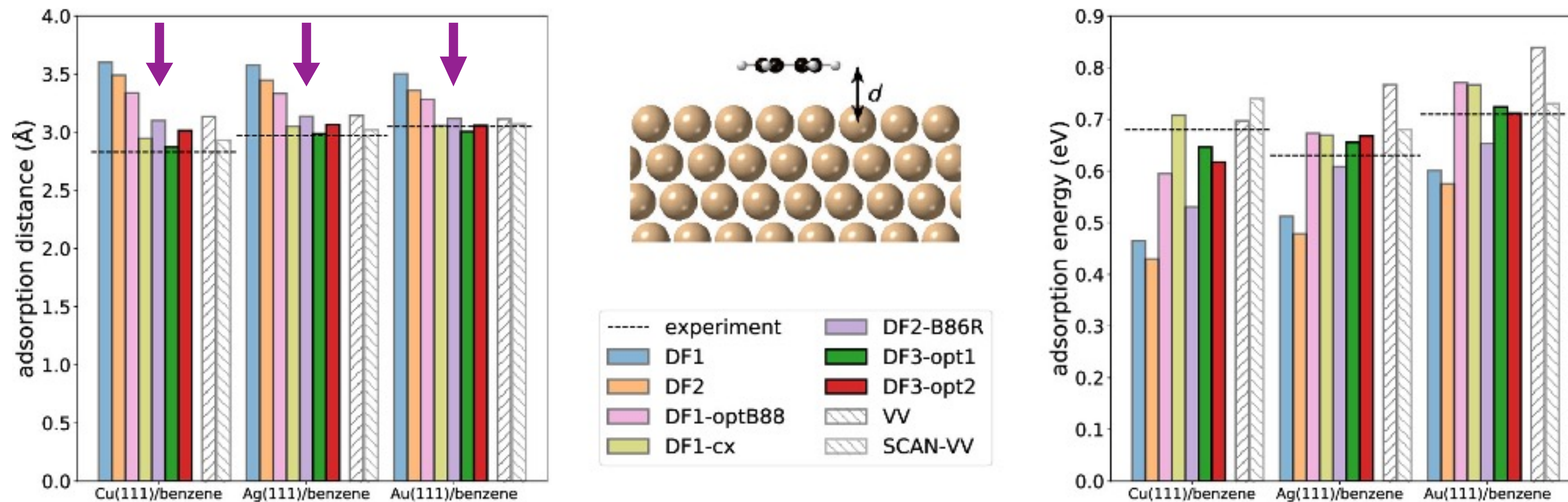
Comparison of vdW-DFs for adsorption systems

Benzene on Cu, Ag, and Au(111) surfaces



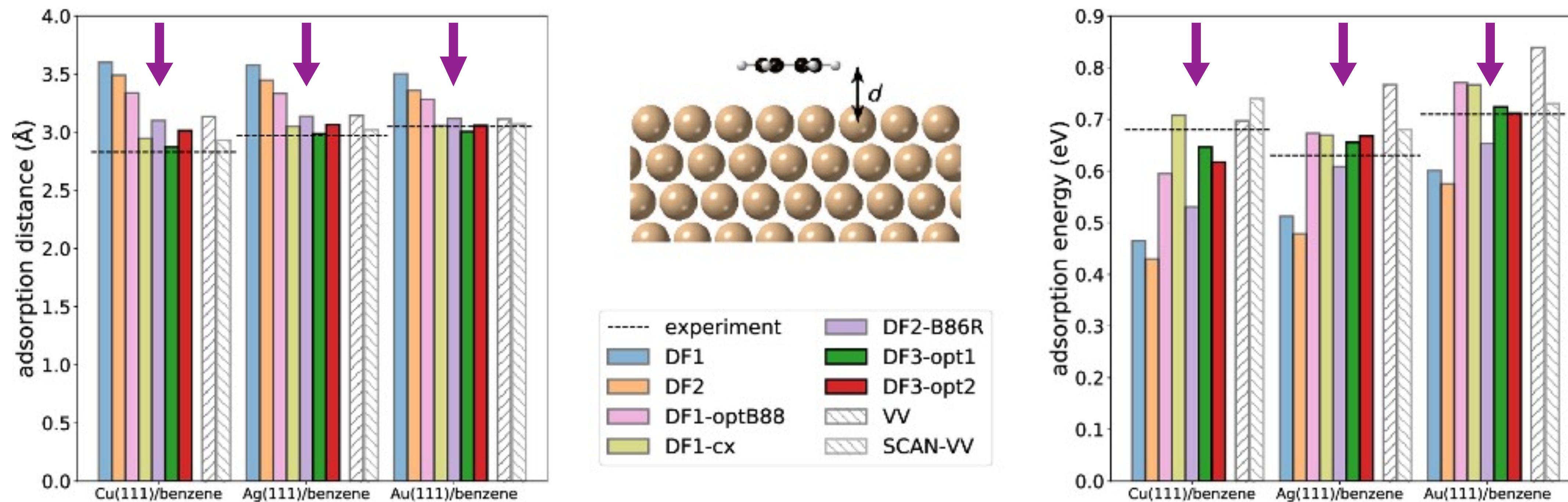
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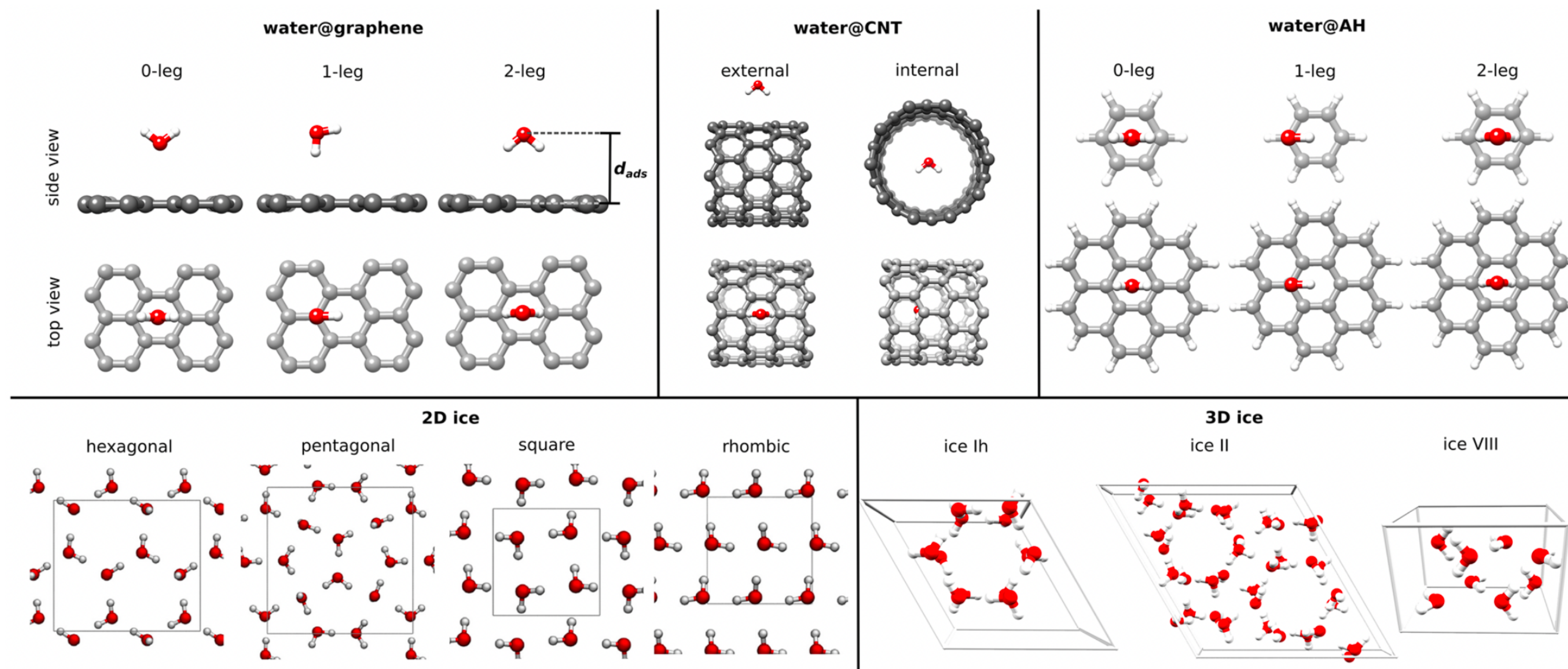


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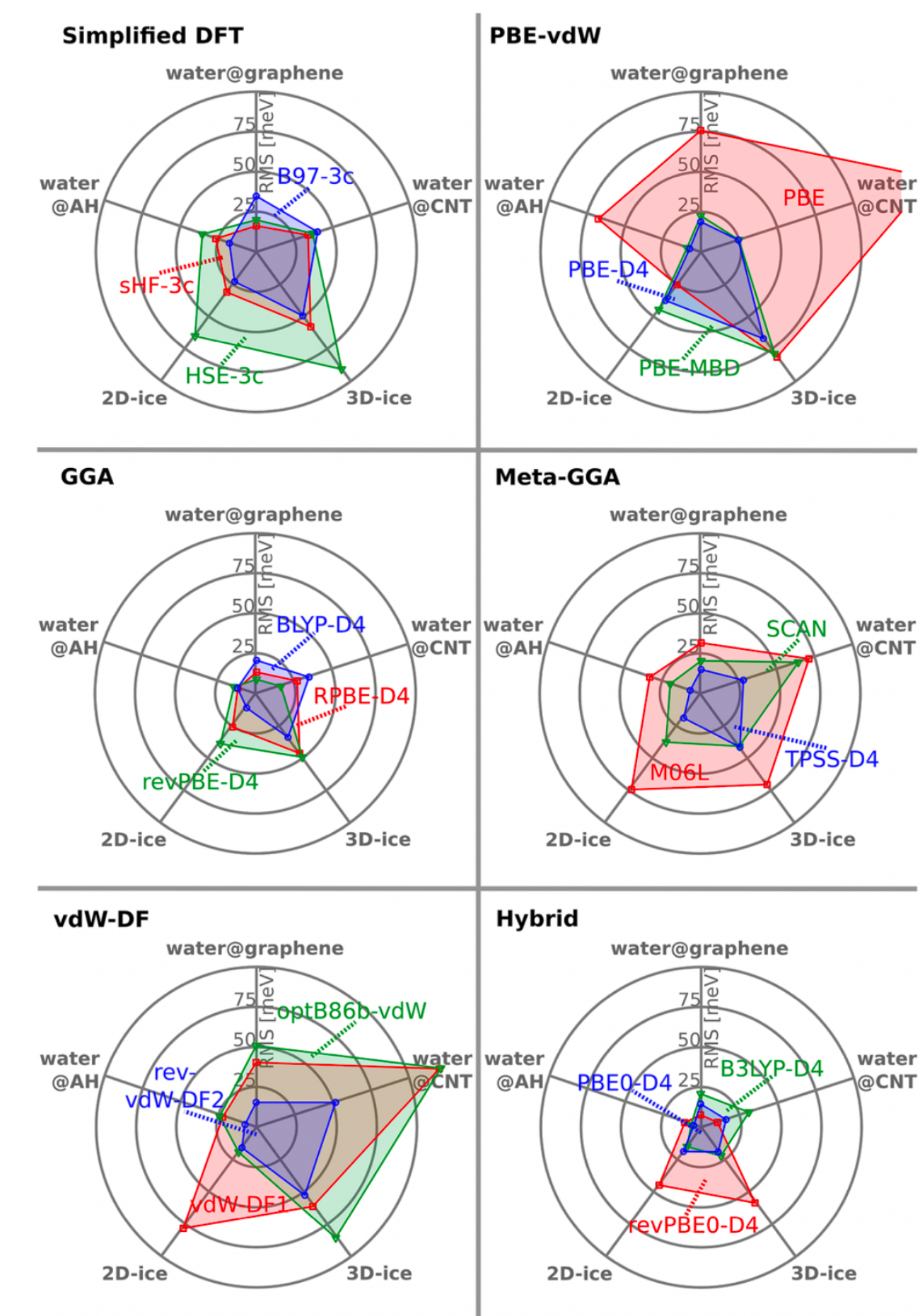
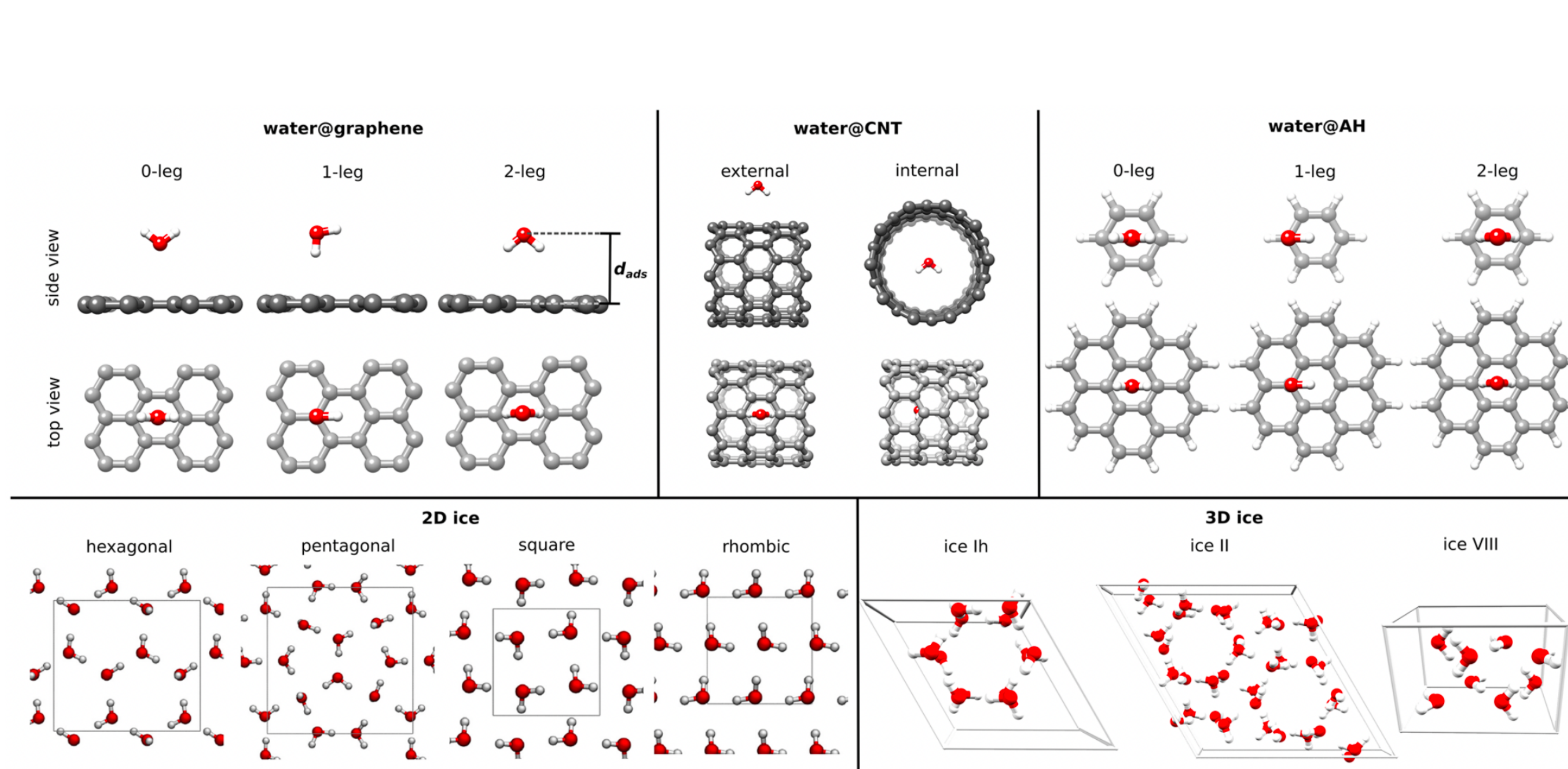


Comparison of vdW-DFs for WaC18 dataset



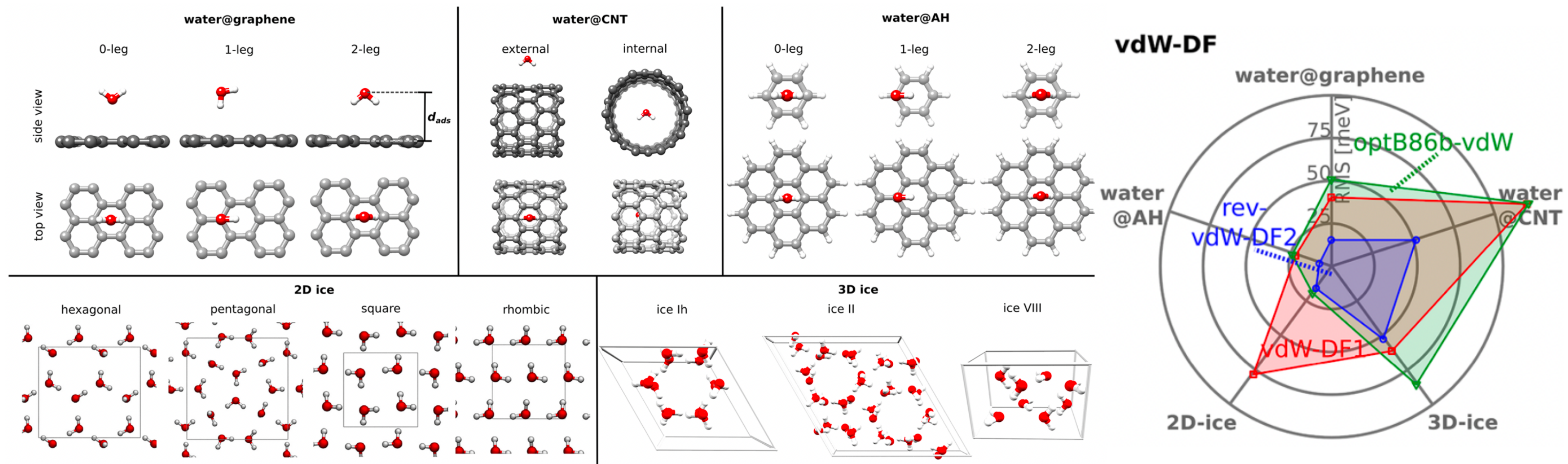
Benchmark quantum Monte Carlo results of
18 water-water and water-carbon systems

Comparison of vdW-DFs for WaC18 dataset



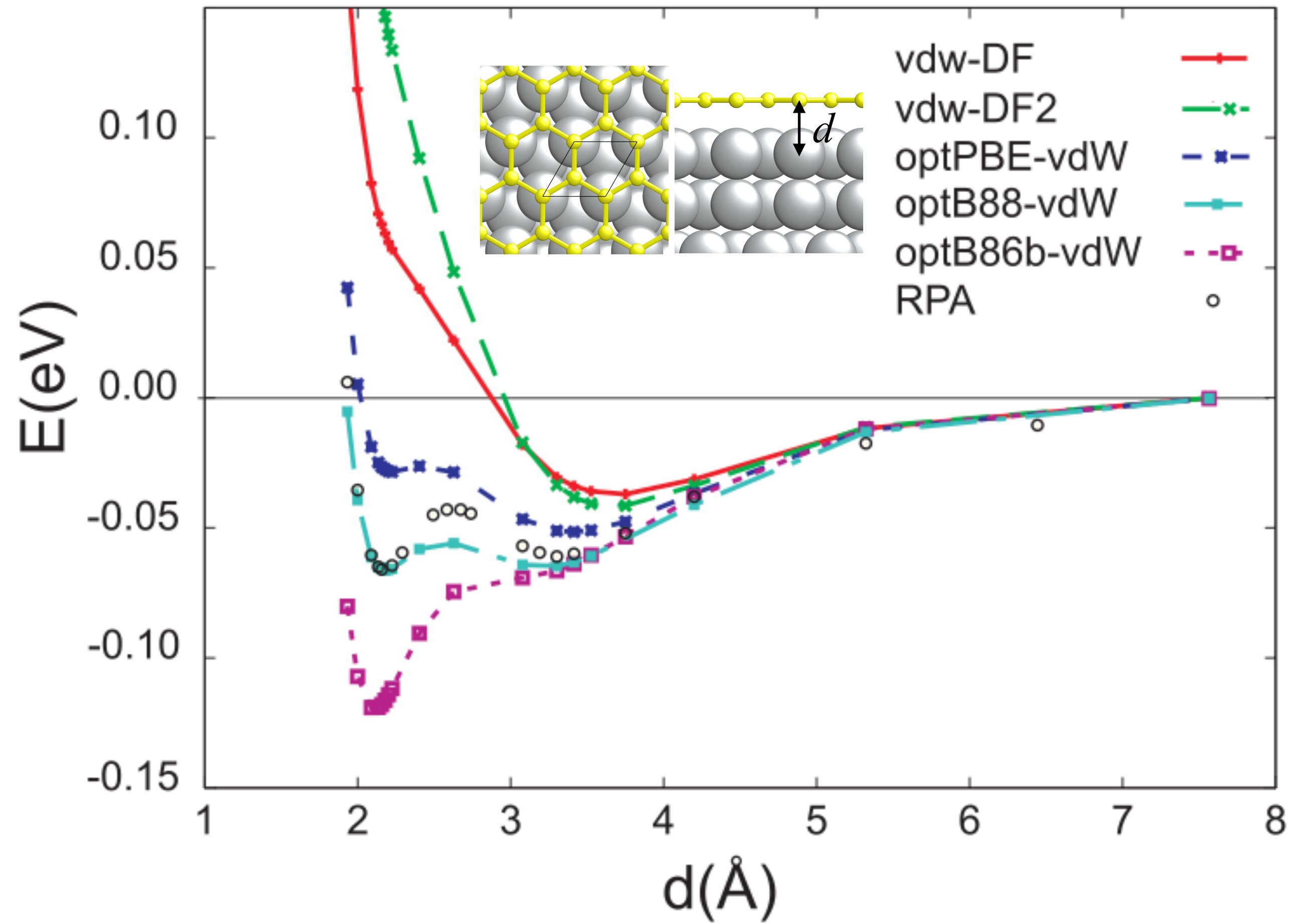
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Comparison of vdW-DFs for WaC18 dataset

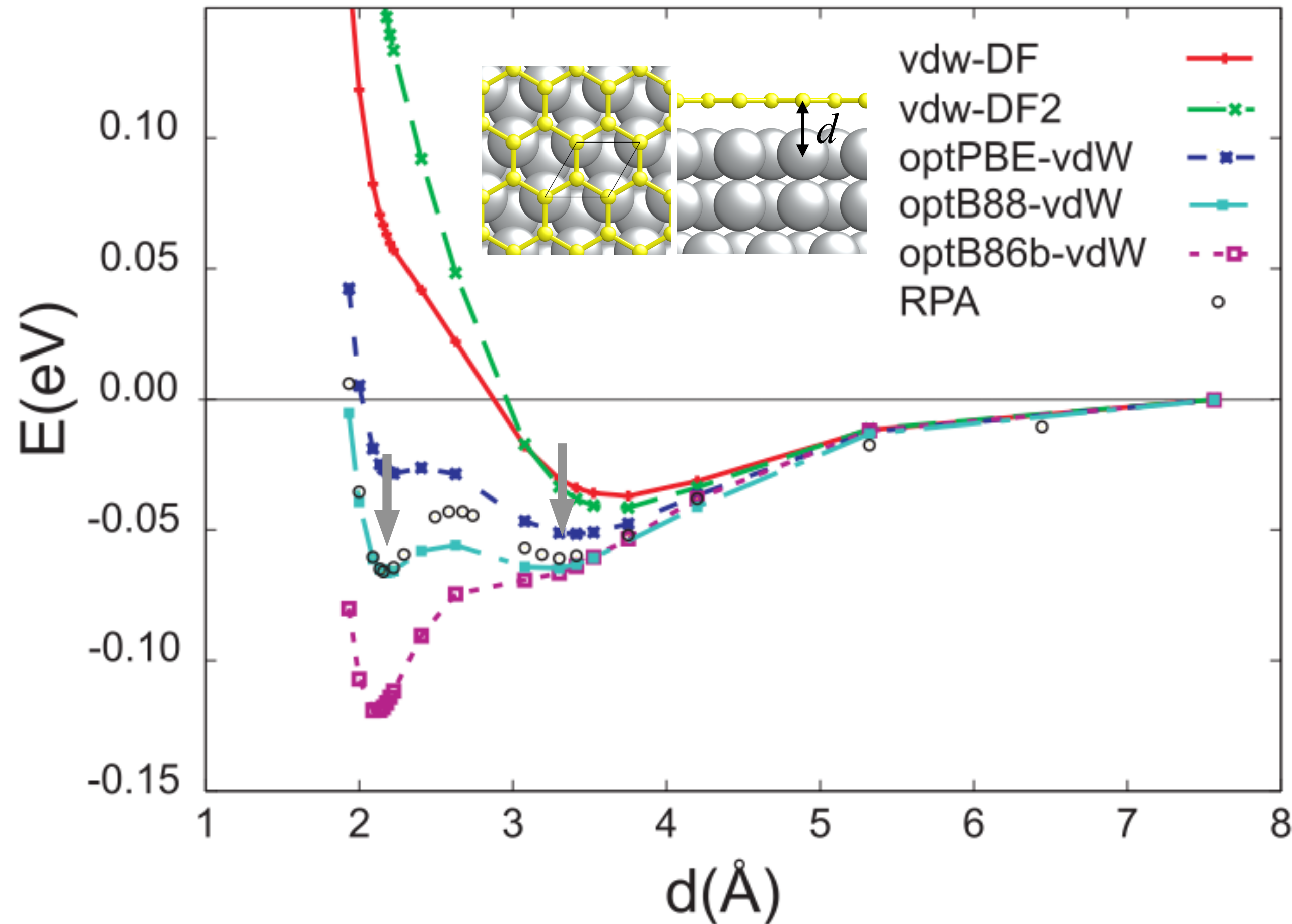


Benchmark quantum Monte Carlo results of
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Graphene/Ni(111)

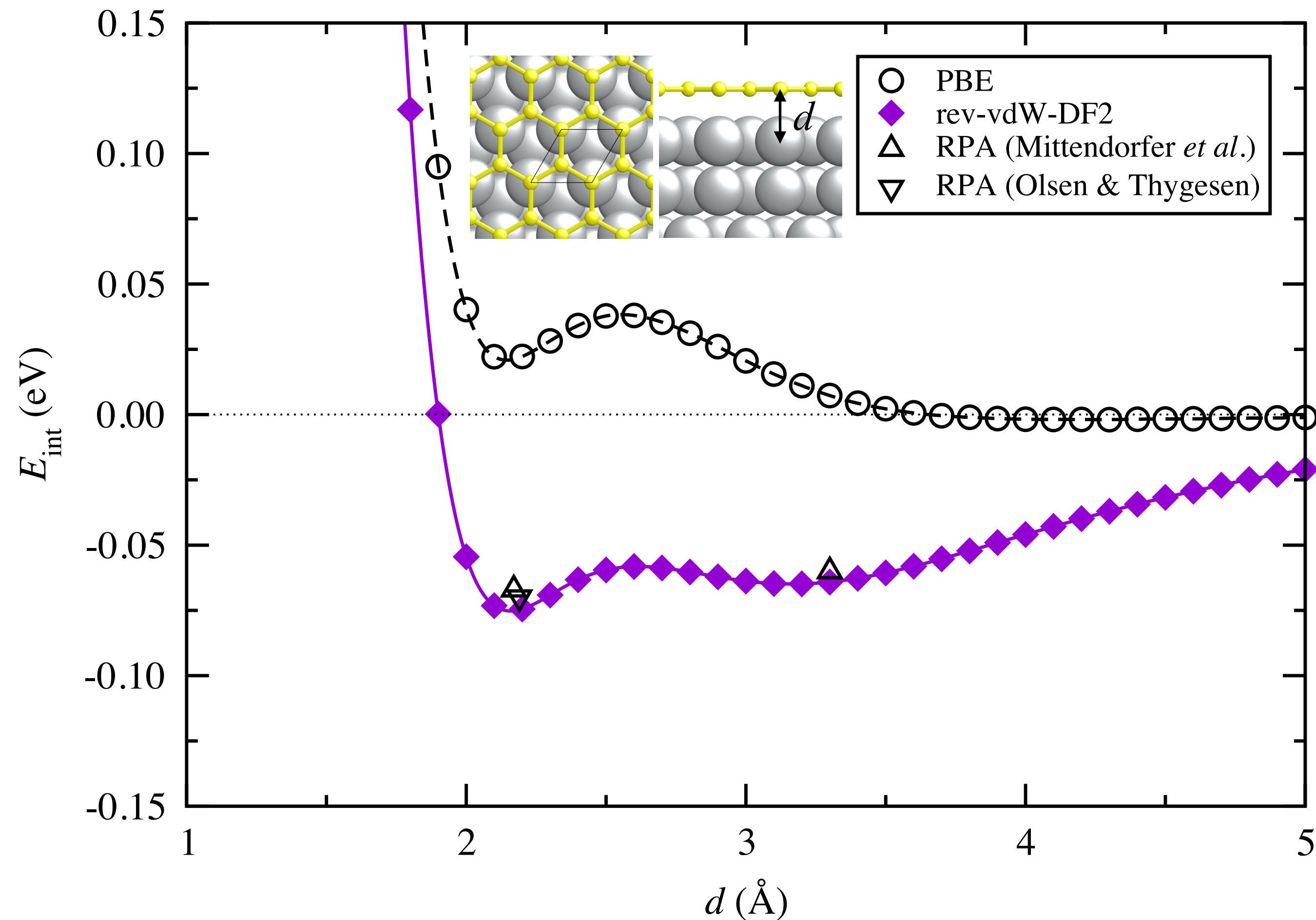


Graphene/Ni(111)



- Two minima in the binding energy curve for chemisorption and physisorption
- Good system to test the accuracy to describe both strong and weak interactions

Graphene/Ni(111)



- rev-vdW-DF2 Interaction energies are in good agreement with the RPA ones
- Two minima in the interaction energy curve are nicely reproduced.

Outline

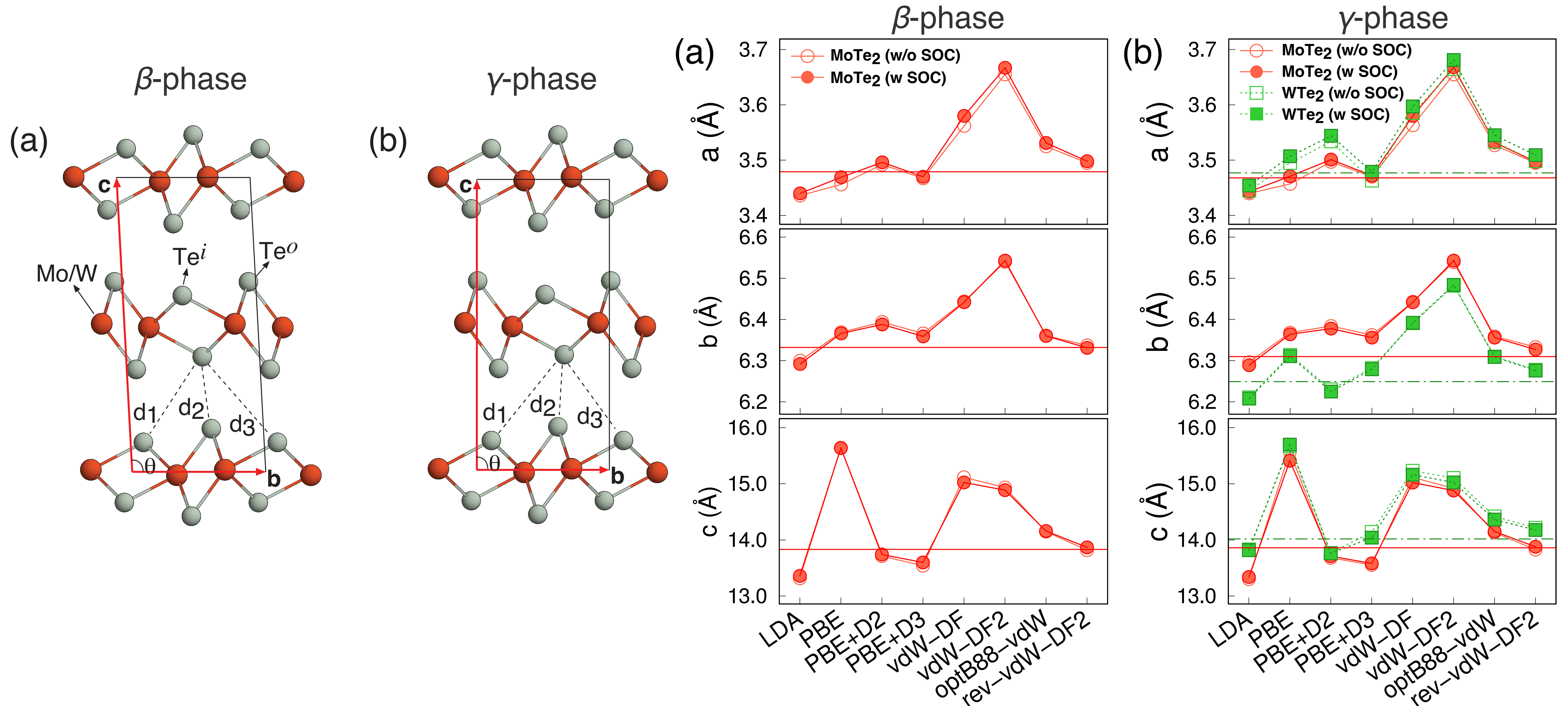
- Background: Dispersion forces in density functional theory
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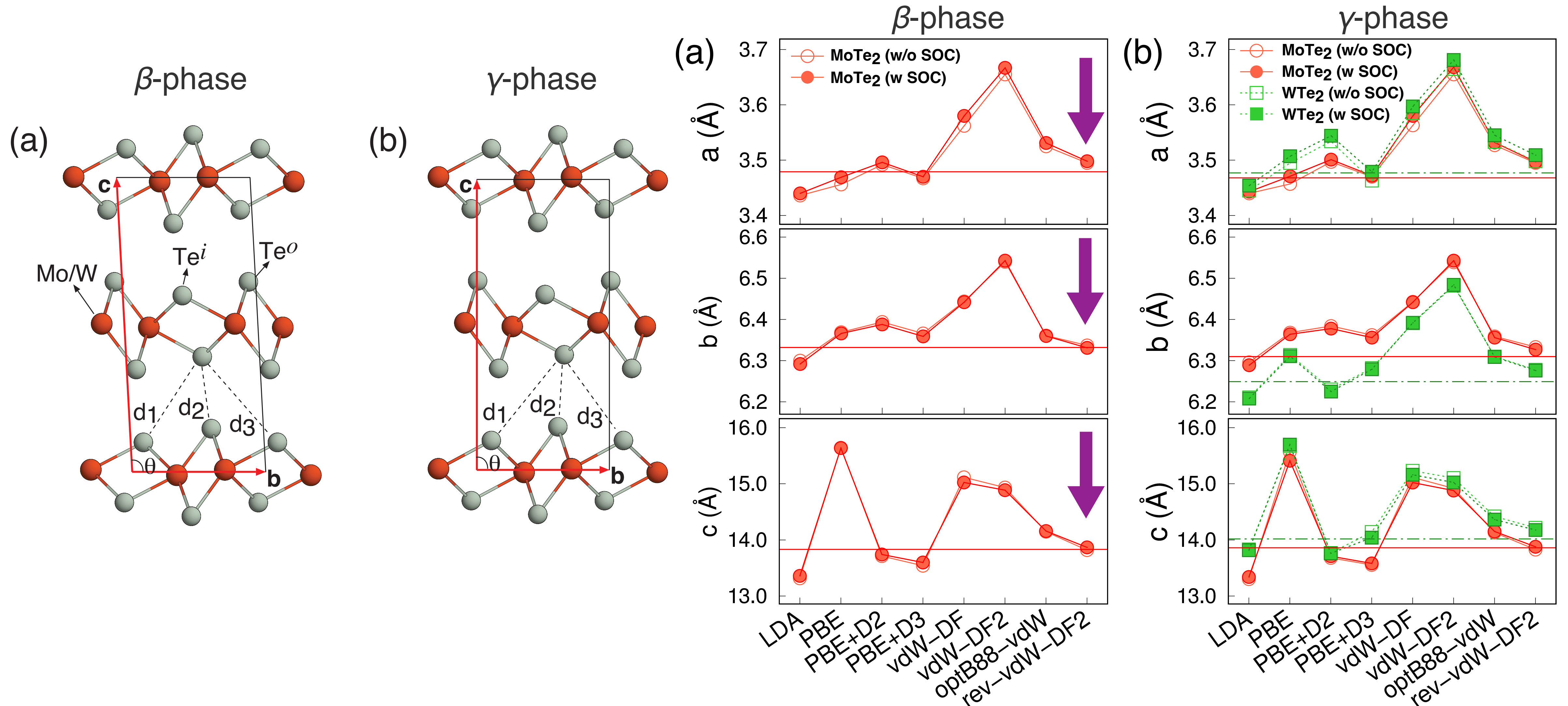
Layered transition metal dichalcogenides

MoTe₂ and WTe₂



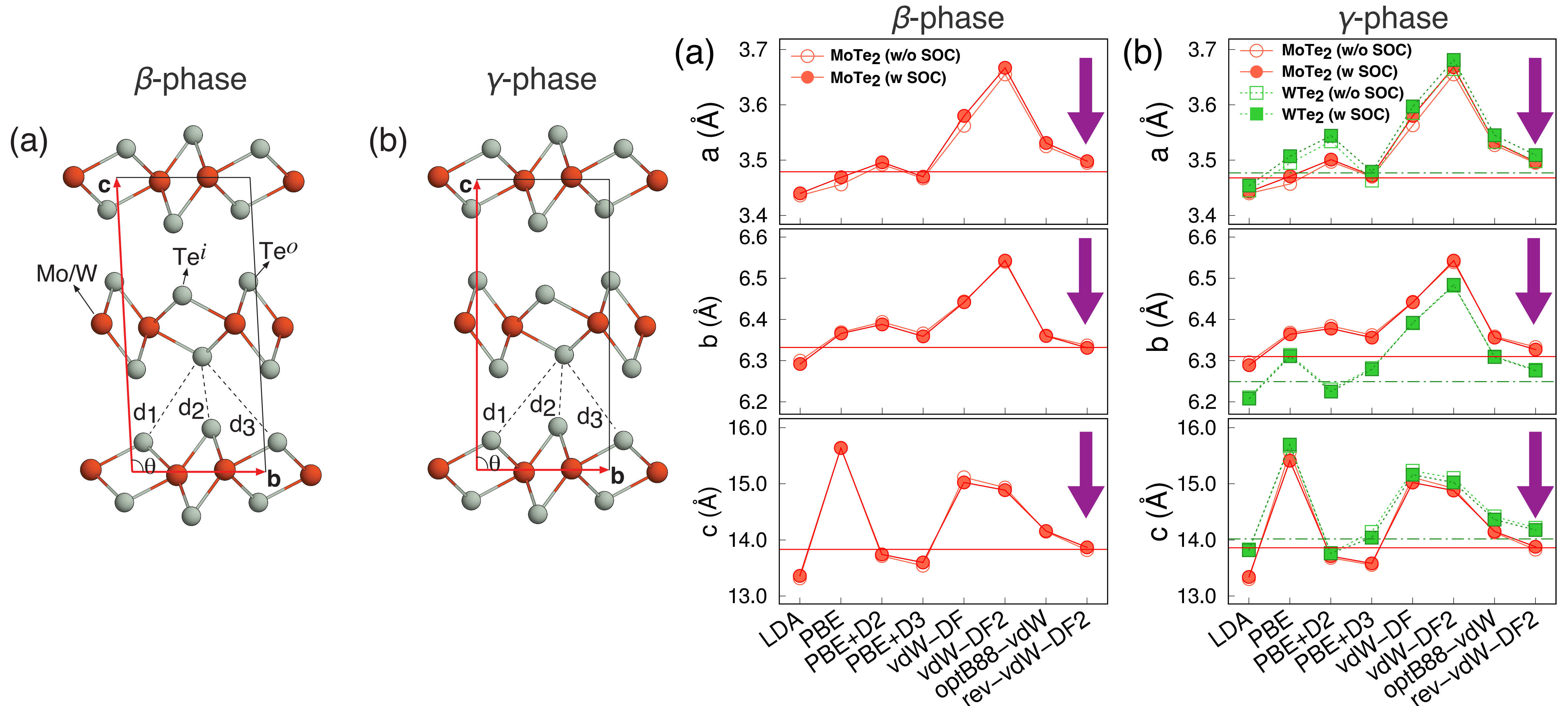
Layered transition metal dichalcogenides

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Layered transition metal dichalcogenides

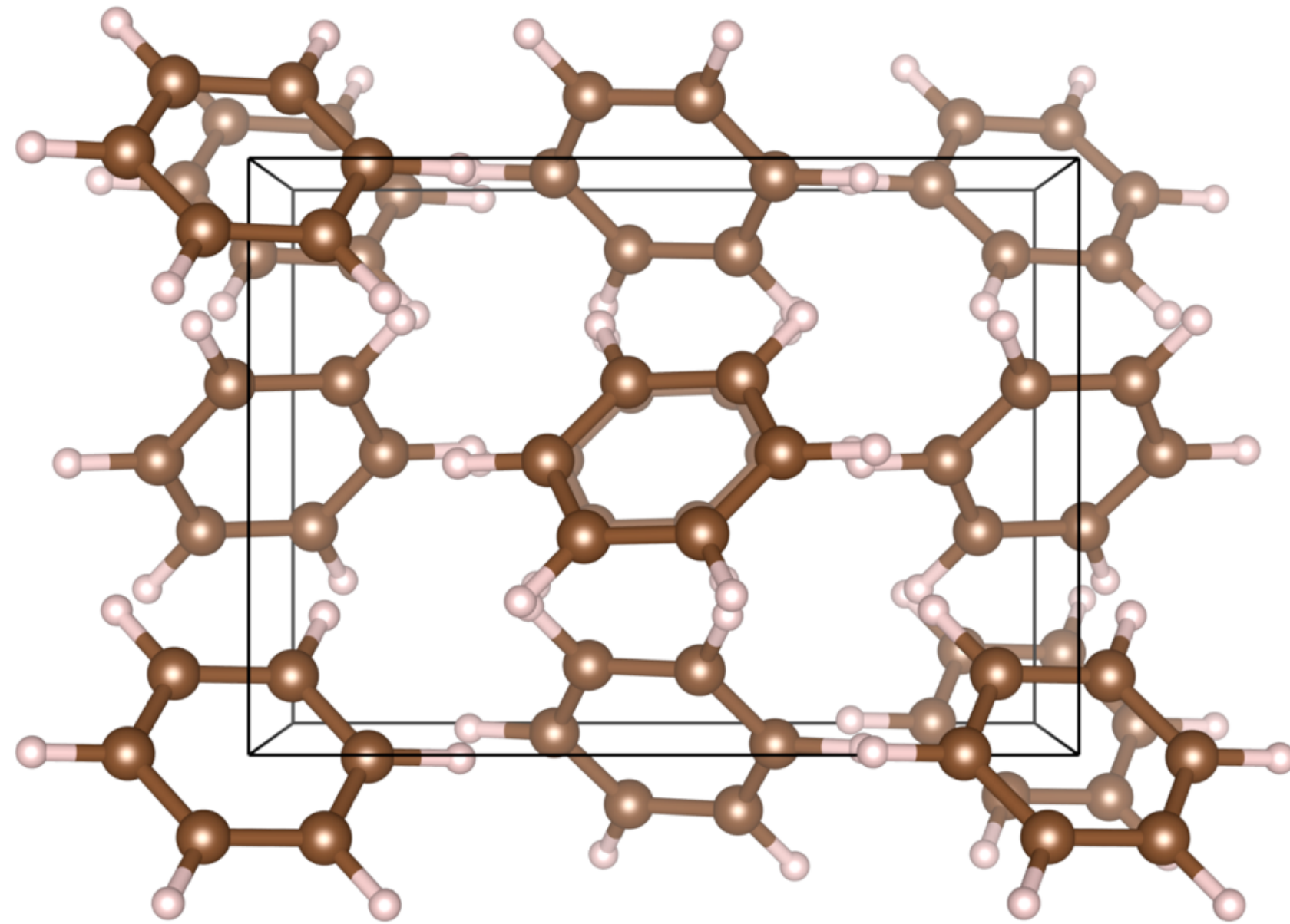
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Structure of an organic solid: Benzene crystal (*Pbca*)

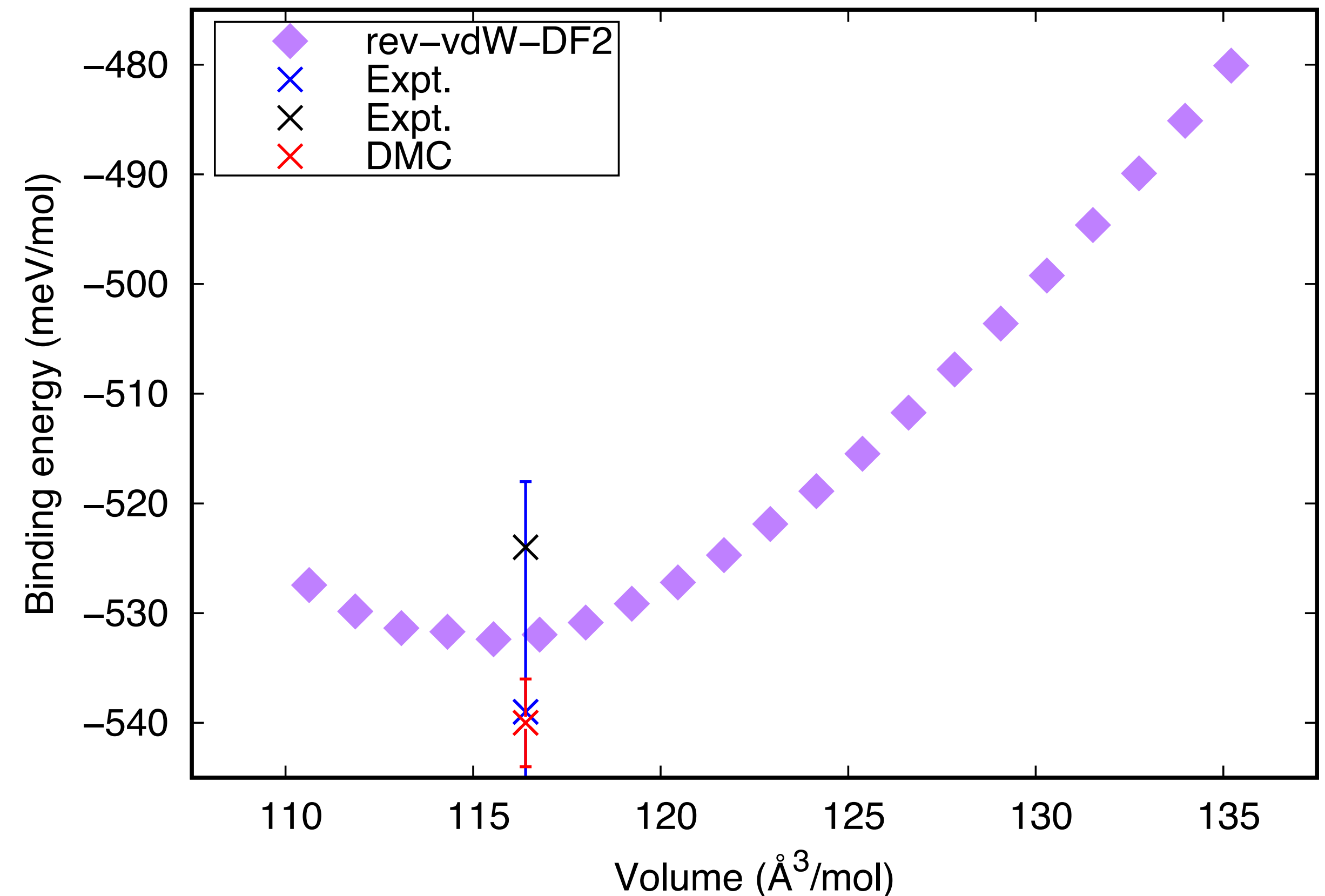
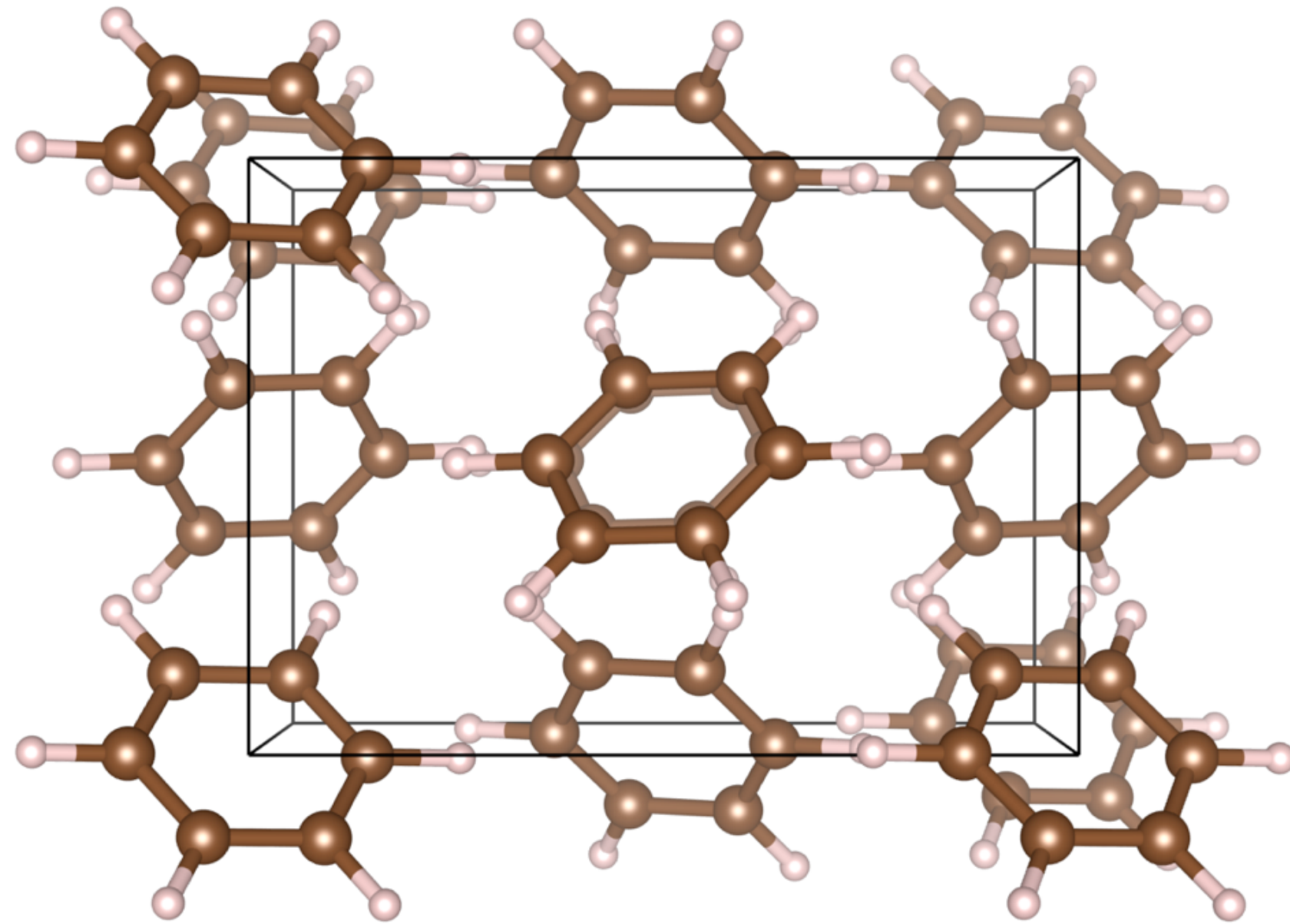


Expt. : Cox, *Rev. Mod. Phys.* **30**, 159 (1958)

RPA: Lu, Li, Rocca, & Galli, *Phys. Rev. Lett.* **102**, 206411 (2009).

DMC: Zen *et al.*, *Proc. Natl. Acad. Sci.* **115**, 1724 (2018).

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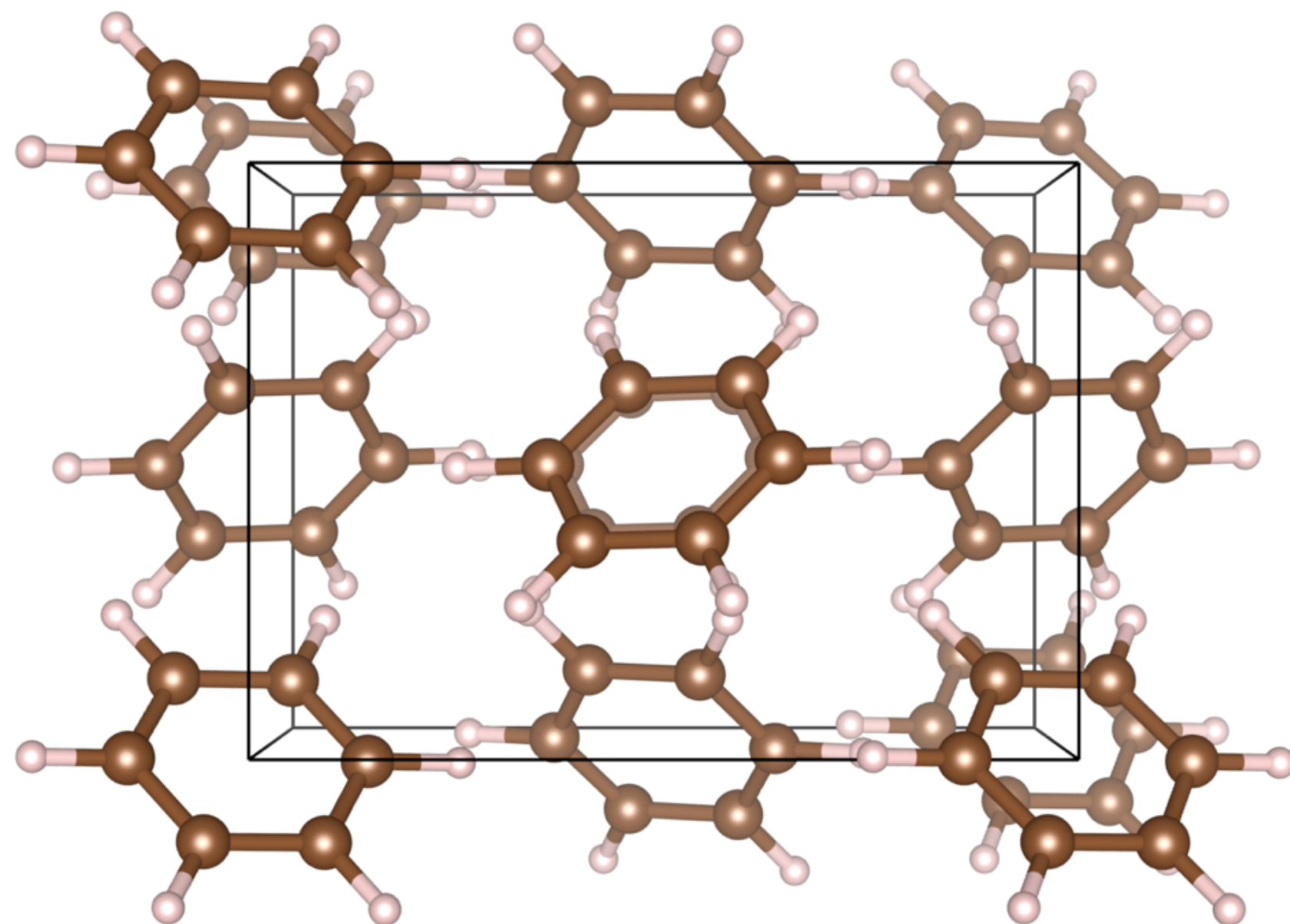


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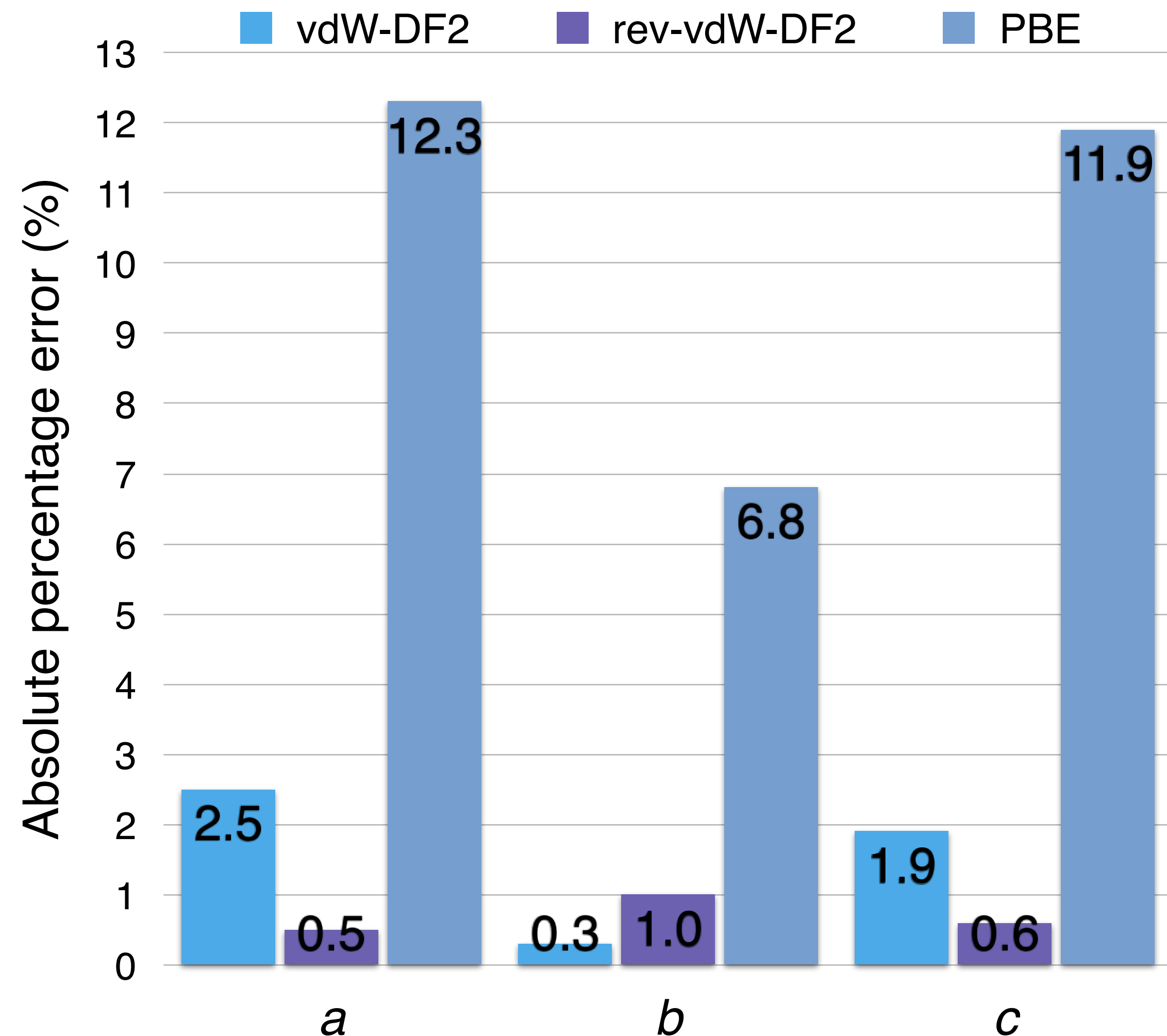
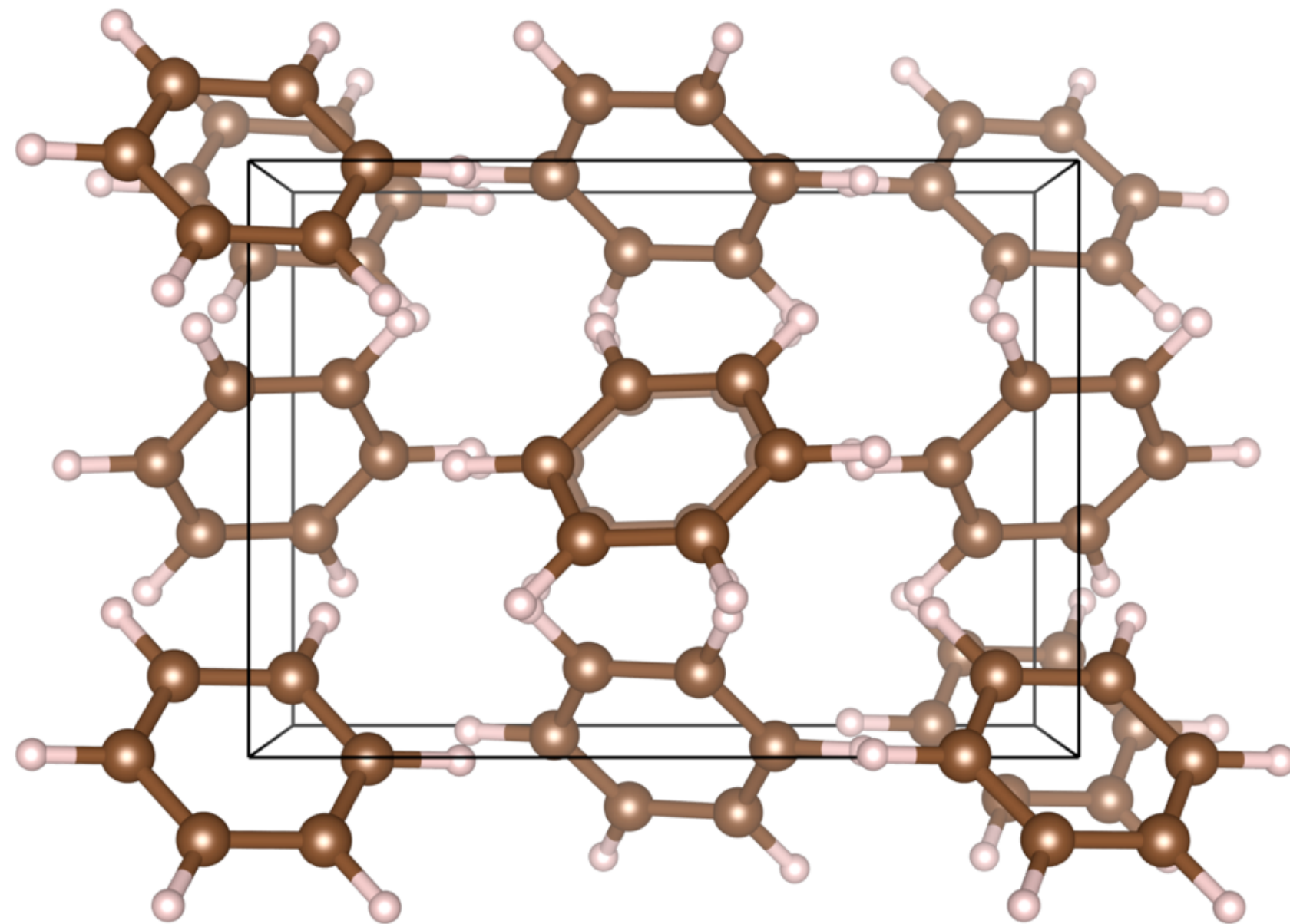
	a (Å)	b (Å)	c (Å)	V_0 (Å ³)	E_b (eV/mol)
PBE	8.19	10.118	7.541	624.9	
vdW-DF2	7.477	9.496	6.872	487.9	0.563
rev-vdW-DF2	7.33	9.38	6.734	463	0.53
RPA@PBE				1.00× V_0	0.487
DMC					0.540±4
Expt. (-195°C)	7.292	9.471	6.742	465.6	0.542

Expt. : Cox, *Rev. Mod. Phys.* **30**, 159 (1958)

RPA: Lu, Li, Rocca, & Galli, *Phys. Rev. Lett.* **102**, 206411 (2009).

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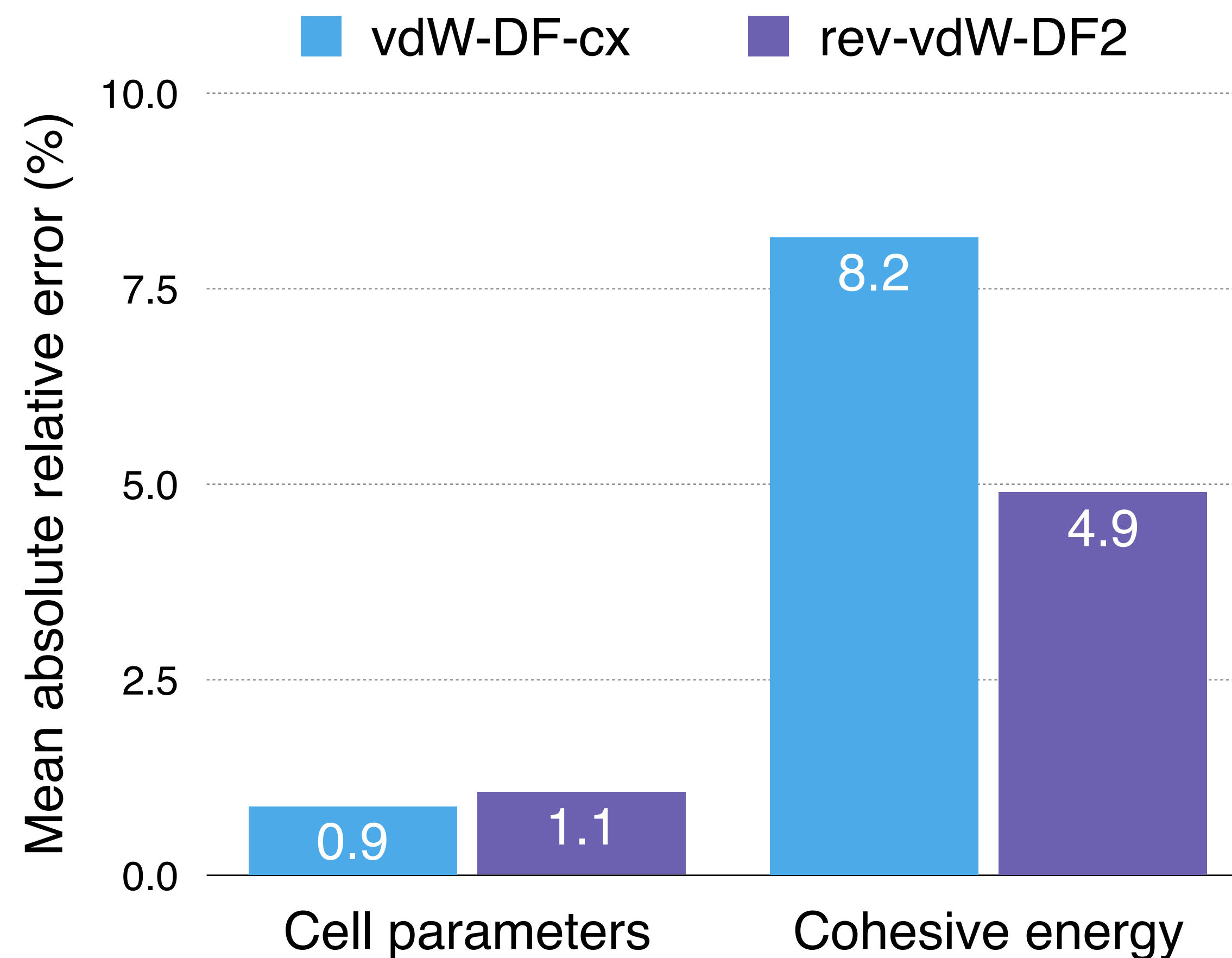
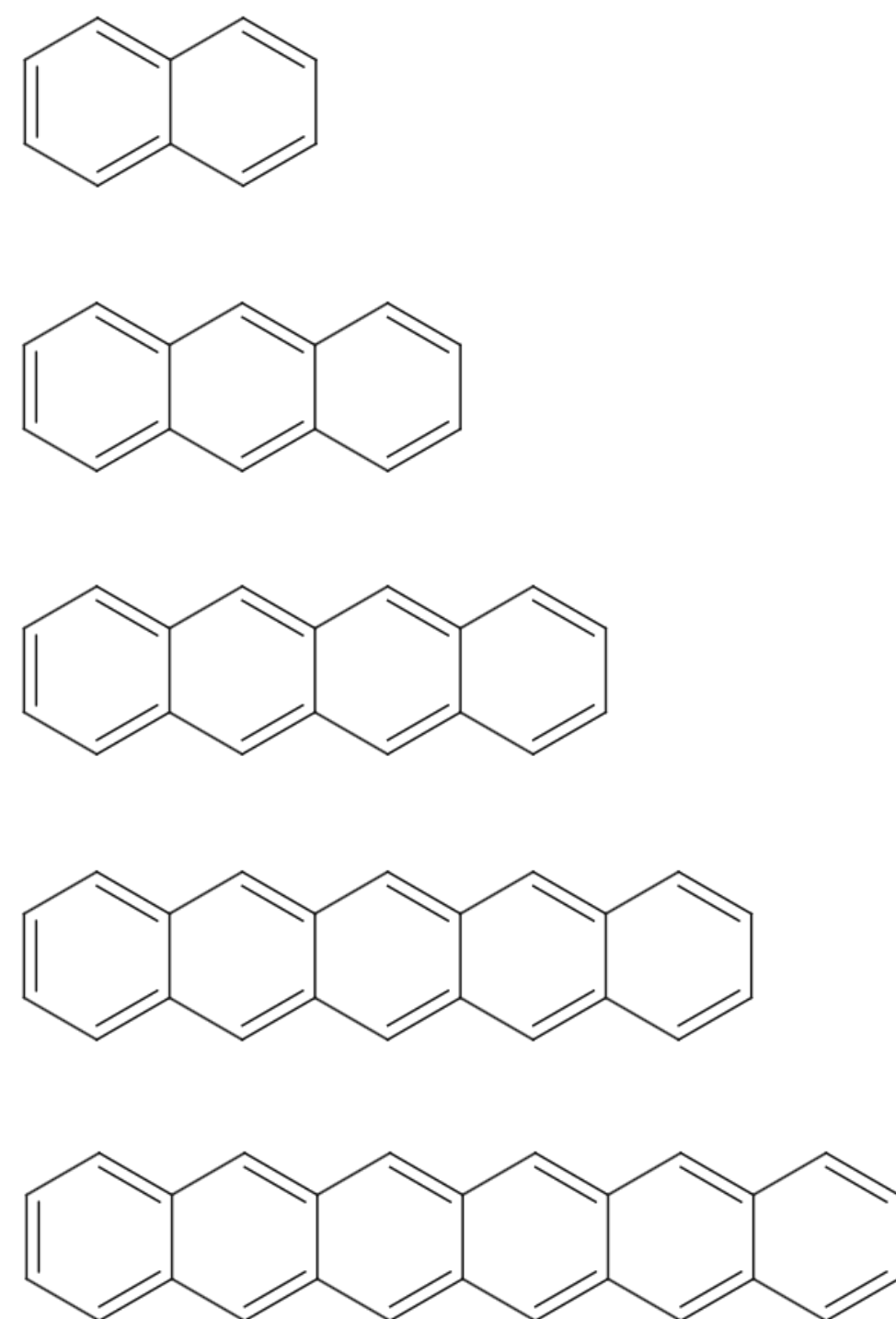


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DMC: Zen *et al.*, *Proc. Natl. Acad. Sci.* **115**, 1724 (2018).

Structures of organic solids: Oligoacene crystals

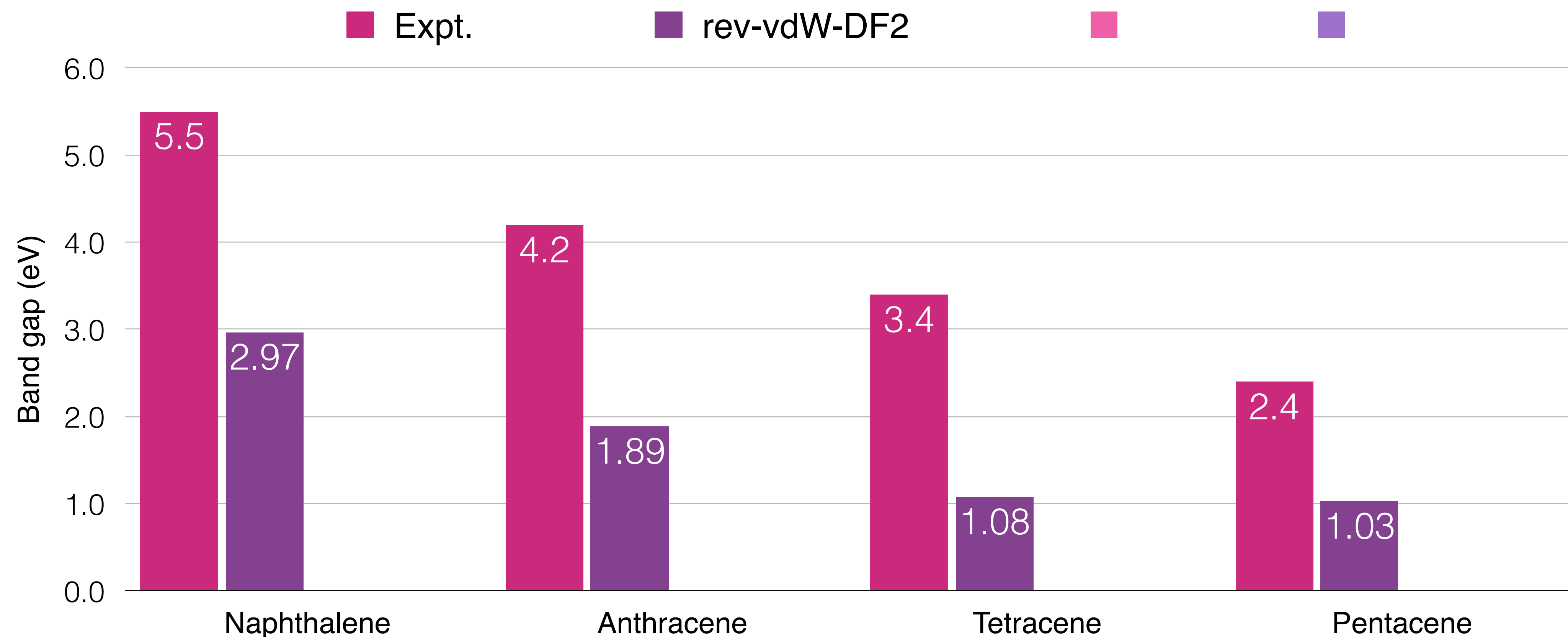


Electronic structure — Band gaps of oligoacene crystals

Yanagisawa and IH, *J. Appl. Phys.* **121**, 045501 (2017).

Yanagisawa and IH in *Theoretical Chemistry for Advanced Nanomaterials
Functional Analysis by Computation and Experiment*. (Springer-Nature, 2020).

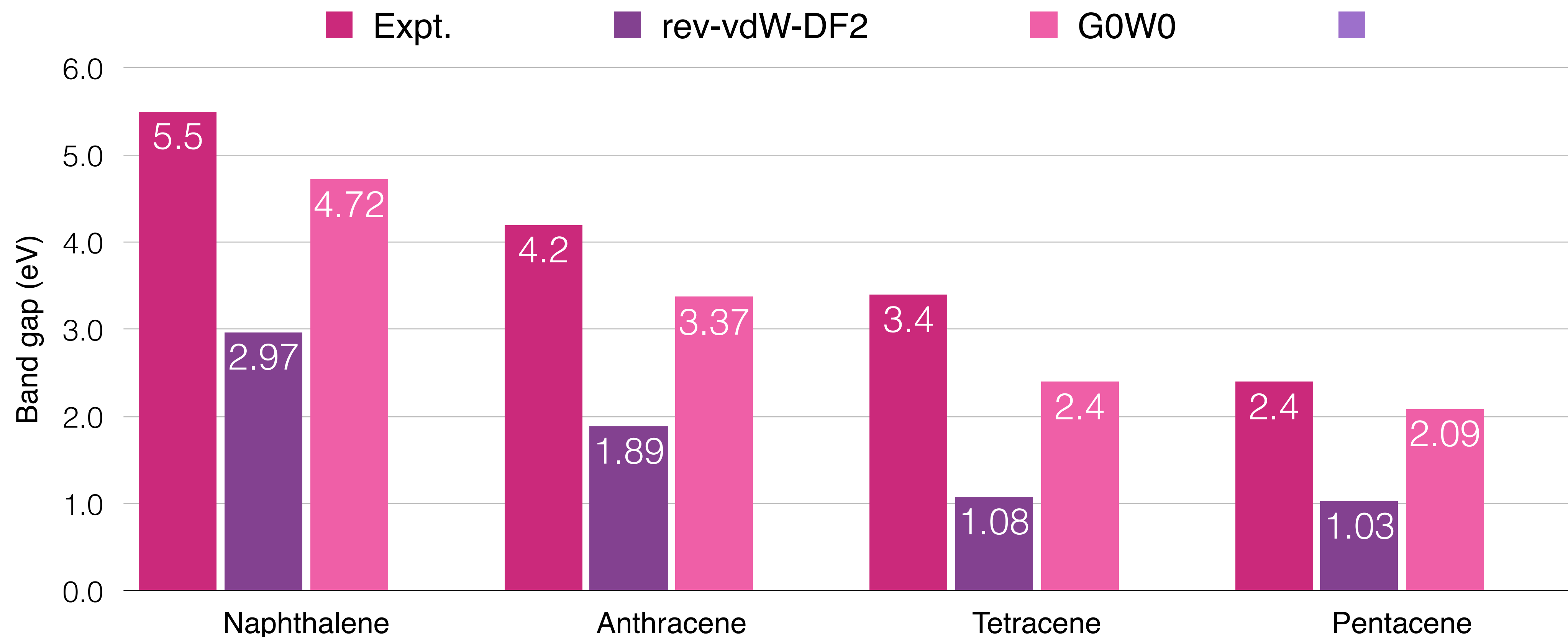
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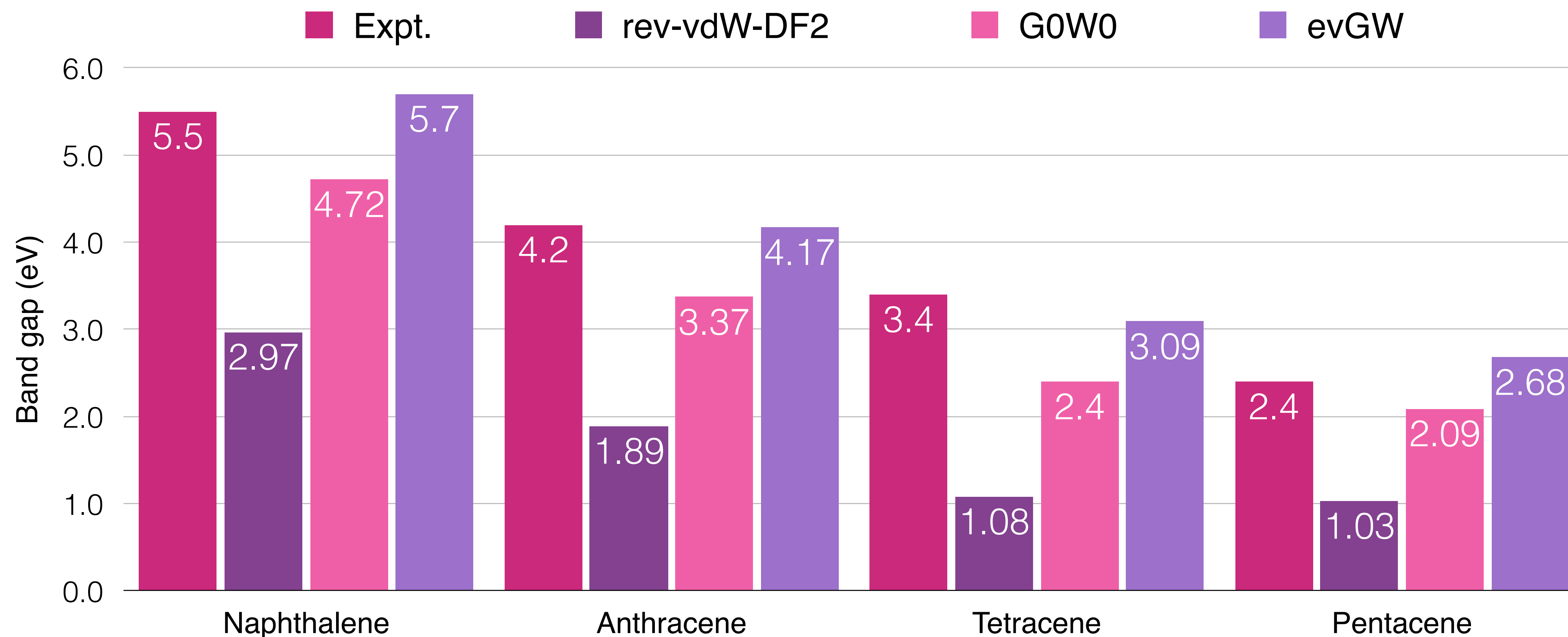
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Lattice dynamics

Lattice dynamics

Diagonalization of the dynamical matrix

$$|D_{s\alpha,t\beta}(\mathbf{q}) - \omega_{\mathbf{q}}^2 \delta_{s,t} \delta_{\alpha,\beta}| = 0$$

$$D_{s\alpha,t\beta}(\mathbf{q}) = \frac{1}{\sqrt{M_s M_t}} \tilde{C}_{s\alpha,t\beta}(\mathbf{q})$$

Lattice dynamics

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Fourier transform of the interatomic force constant $[C_{s\alpha,t\beta}(I, J)]$

$$\tilde{C}_{s\alpha,t\beta}(\mathbf{q}) = \frac{1}{N} \sum_{I,J} C_{s\alpha,t\beta}(I, J) e^{-i\mathbf{q} \cdot (\mathbf{R}_I - \mathbf{R}_J)}$$

Lattice dynamics from the perturbation theory

Diagonalization of the dynamical matrix

$$|D_{s\alpha,t\beta}(\mathbf{q}) - \omega_{\mathbf{q}}^2 \delta_{s,t} \delta_{\alpha,\beta}| = 0$$

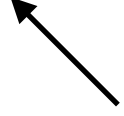
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$$D_{s\alpha,t\beta}(\mathbf{q}) = \frac{1}{\sqrt{M_s M_t}} \frac{\partial^2 E_{\text{tot}}}{\partial u_{s,\mathbf{q}}^{\alpha*} \partial u_{t,\mathbf{q}}^{\beta}}$$

Fourier transform of atomic displacement



Lattice dynamics from the perturbation theory

Diagonalization of the dynamical matrix

$$|D_{s\alpha,t\beta}(\mathbf{q}) - \omega_{\mathbf{q}}^2 \delta_{s,t} \delta_{\alpha,\beta}| = 0$$

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$$\frac{\partial^2 E_{\text{tot}}}{\partial u_{s,\mathbf{q}}^{\alpha*} \partial u_{t,\mathbf{q}}^{\beta}} = \sum_m^{\text{occ}} \sum_{\mathbf{k} \in \text{BZ}} \left(2 \left\langle \frac{\partial \psi_{m,\mathbf{k}}}{\partial u_{s,\mathbf{q}}^{\alpha}} \left| \frac{\partial v_{\text{ext}}}{\partial u_{t,\mathbf{q}}^{\beta}} \right| \psi_{m,\mathbf{k}} \right\rangle + \left\langle \psi_{m,\mathbf{k}} \left| \frac{\partial^2 v_{\text{ext}}}{\partial u_{s,\mathbf{q}}^{\alpha*} \partial u_{t,\mathbf{q}}^{\beta}} \right| \psi_{m,\mathbf{k}} \right\rangle \right) + \frac{\partial^2 E_{\text{II}}}{\partial u_{s,\mathbf{q}}^{\alpha*} \partial u_{t,\mathbf{q}}^{\beta}}$$

↑
First-order change in the wave function (modulated by $\mathbf{q}$)

Density functional perturbation theory

Linearized Kohn-Sham equation wrt external perturbation $\lambda_{\mathbf{q}}$ (Sternheimer equation)

$$\left(\hat{H} - \varepsilon_{m,\mathbf{k}} \right) \left| \frac{\partial \psi_{m,\mathbf{k}}}{\partial u_{s,\mathbf{q}}^{\alpha}} \right\rangle = - \left(\frac{\partial \hat{H}}{\partial u_{s,\mathbf{q}}^{\alpha}} - \frac{\partial \varepsilon_{m,\mathbf{k}}}{\partial u_{s,\mathbf{q}}^{\alpha}} \right) \left| \psi_{m,\mathbf{k}} \right\rangle$$

Density functional perturbation theory

Linearized Kohn-Sham equation wrt external perturbation $\lambda_{\mathbf{q}}$ (Sternheimer equation)

$$\left(\hat{H} - \varepsilon_{m,\mathbf{k}} \right) \hat{P}_{\mathbf{c},\mathbf{k}+\mathbf{q}} \left| \frac{\partial \psi_{m,\mathbf{k}}}{\partial u_{\mathbf{q}}^{\alpha}} \right\rangle = - \hat{P}_{\mathbf{c},\mathbf{k}+\mathbf{q}} \frac{\partial \hat{H}}{\partial u_{s,\mathbf{q}}^{\alpha}} \left| \psi_{m,\mathbf{k}} \right\rangle$$

$$\hat{P}_{\mathbf{c},\mathbf{k}+\mathbf{q}} = 1 - \sum_m^{\text{occ}} \left| \psi_{m,\mathbf{k}+\mathbf{q}} \right\rangle \left\langle \psi_{m,\mathbf{k}+\mathbf{q}} \right|$$

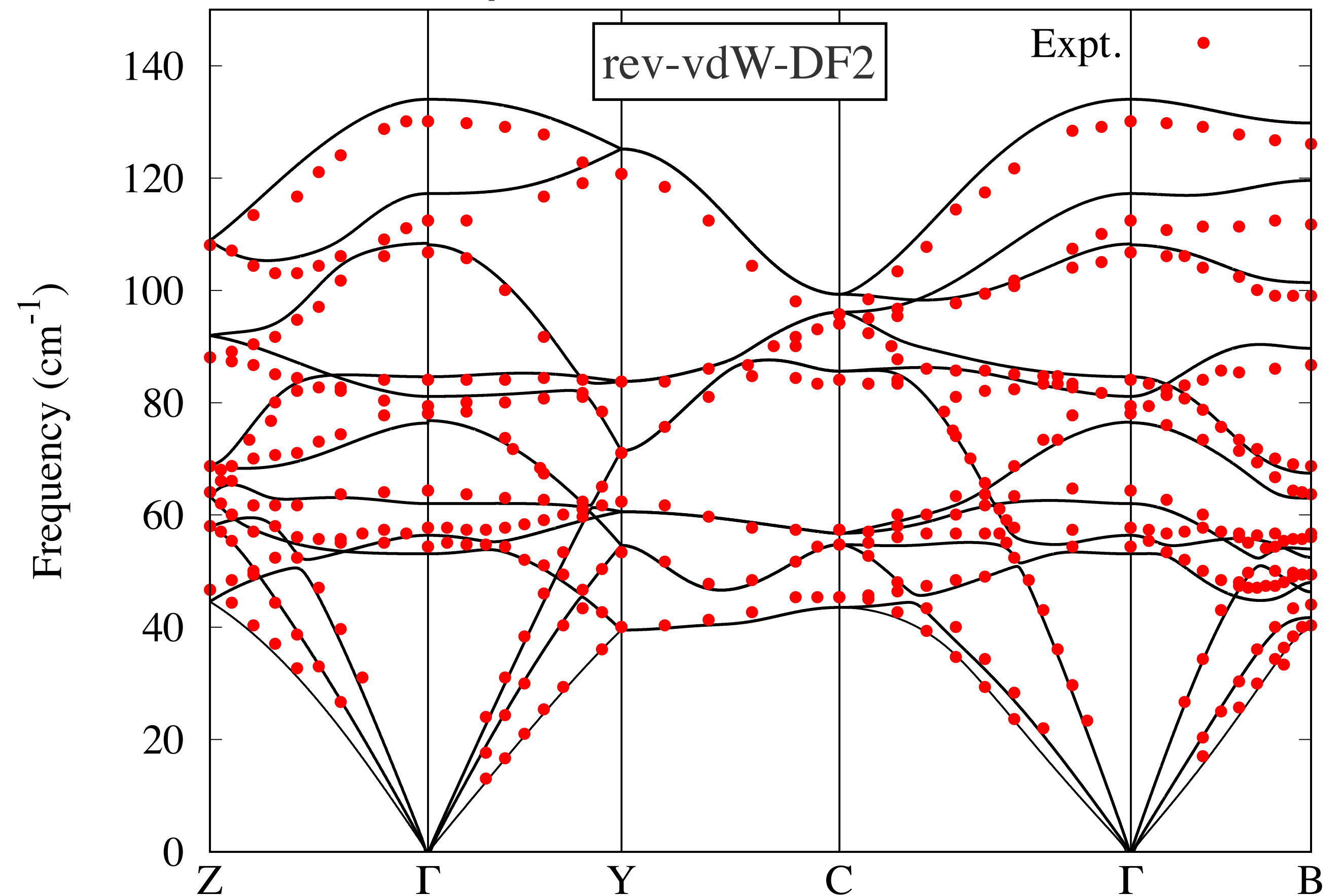
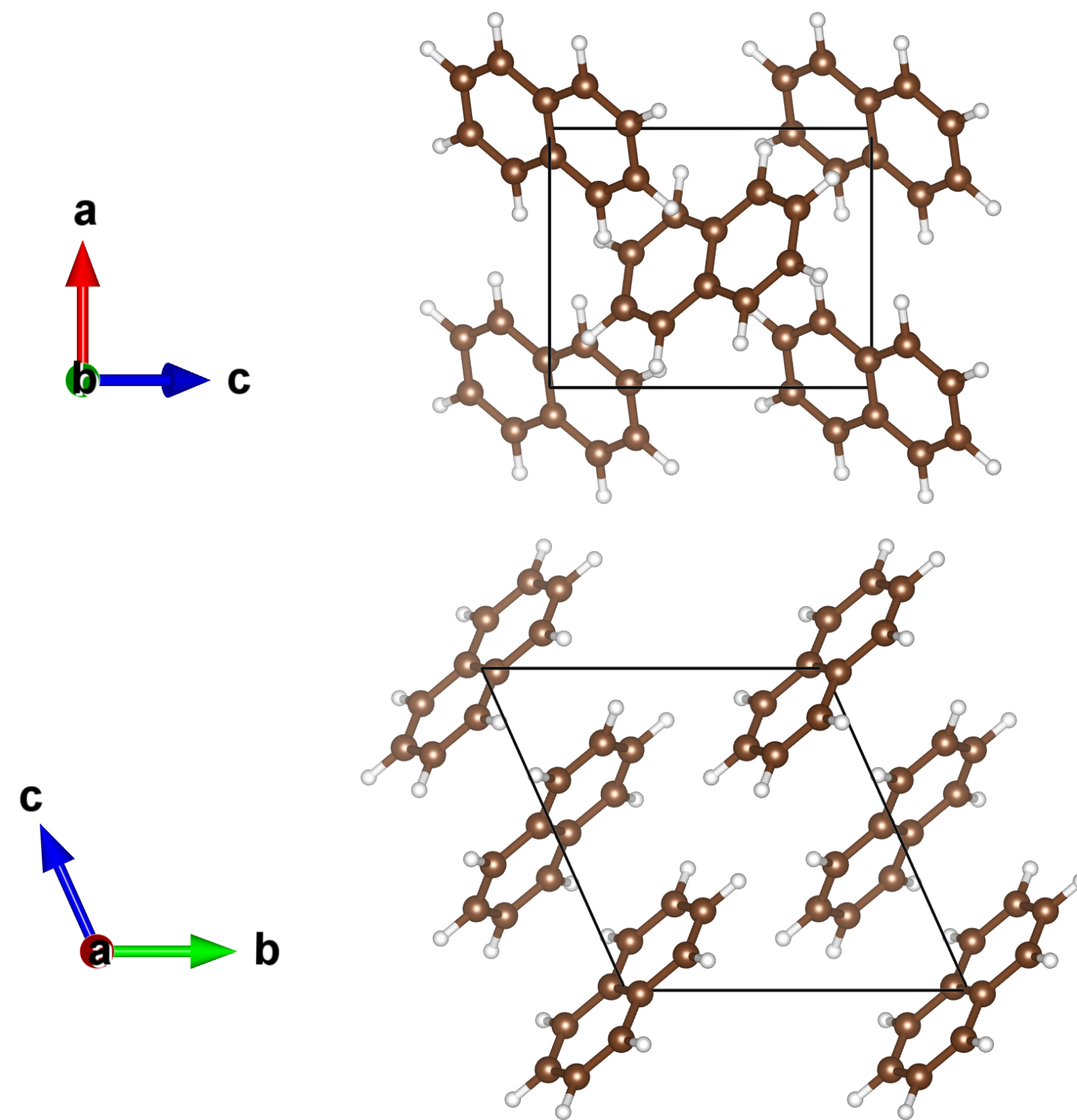
Treatment of incommensurate perturbations (arbitrary $\mathbf{q}$) allowed,
in contrast to the finite difference method

Lattice dynamics

Phonon band structure of d₈-naphthalene

Lattice dynamics

Phonon band structure of d₈-naphthalene

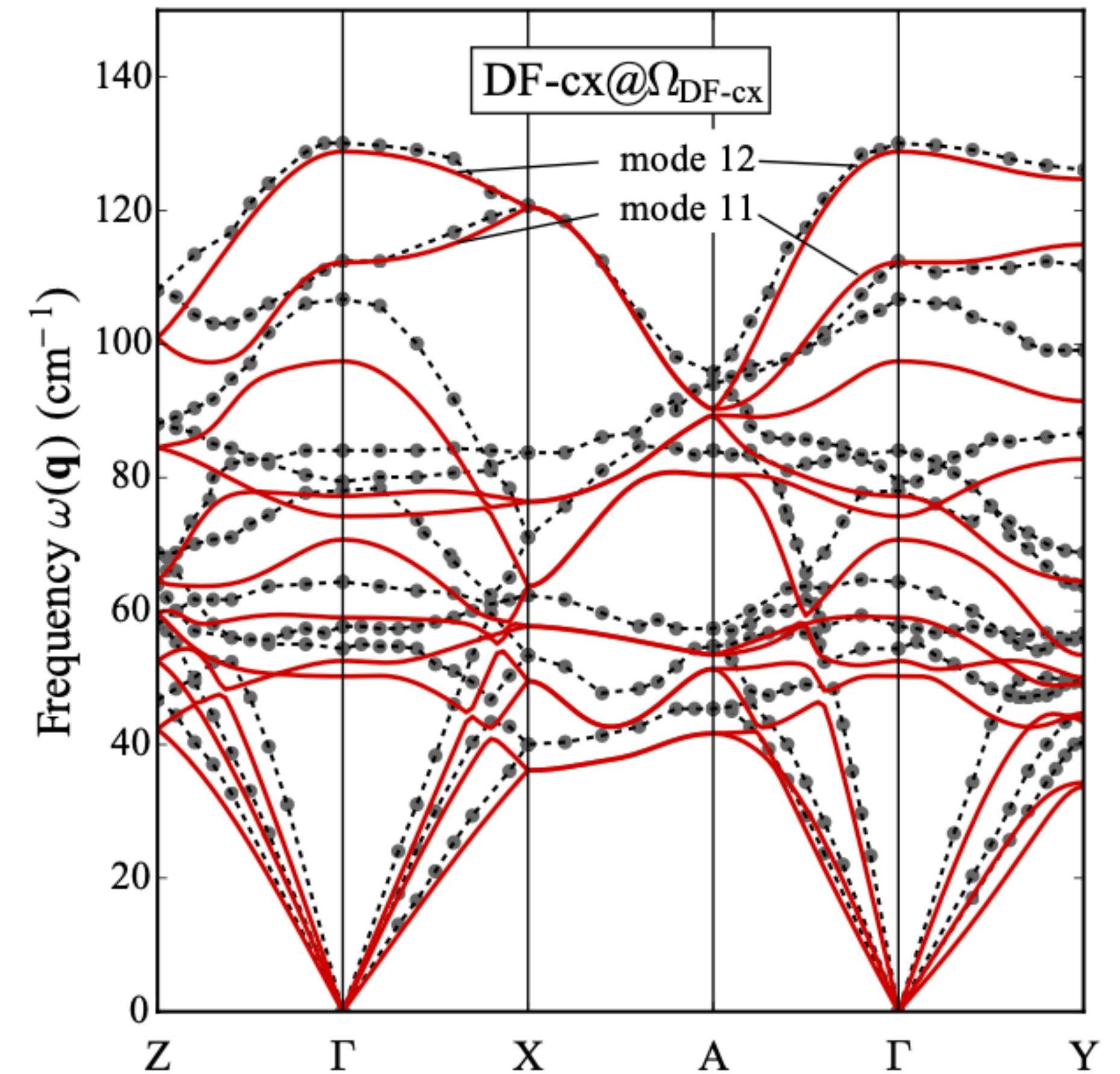
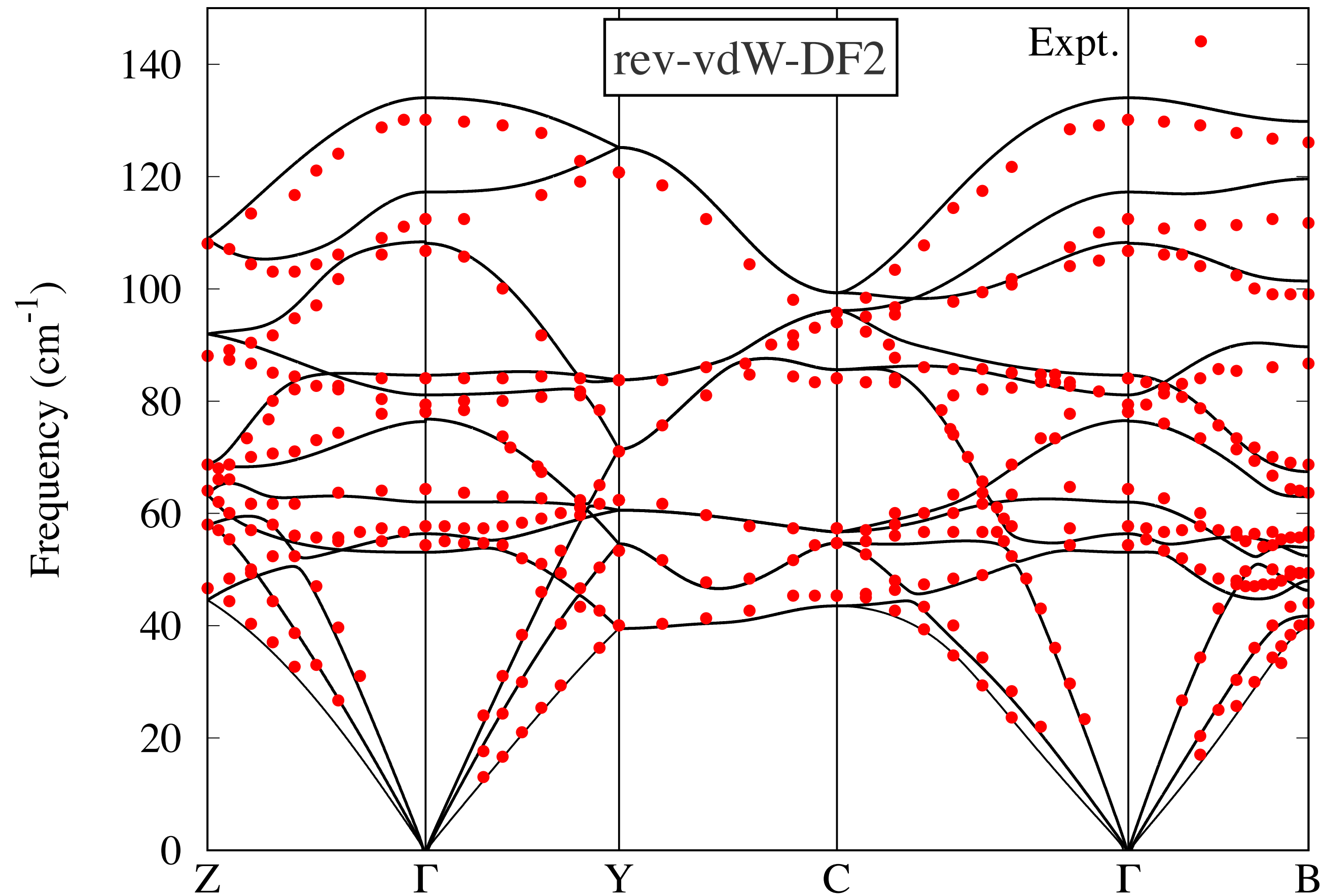


rev-vdW-DF2: IH (unpublished).

Expt. Natkaniec *et al.*, *J. Phys. C: Solid State Phys.* **13**, 4265 (1980).

Lattice dynamics

Phonon band structure of d₈-naphthalene

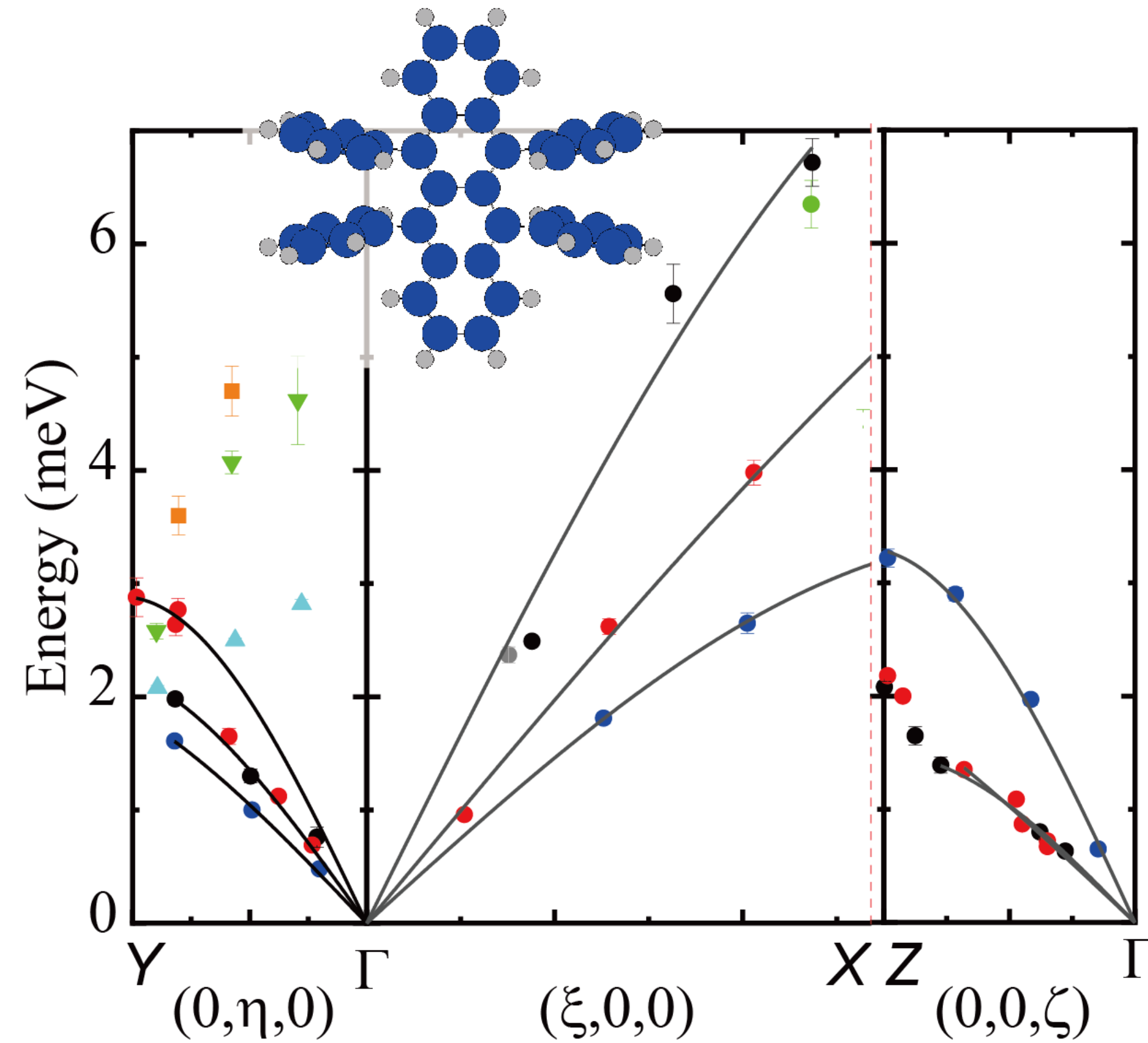


rev-vdW-DF2: IH (unpublished).

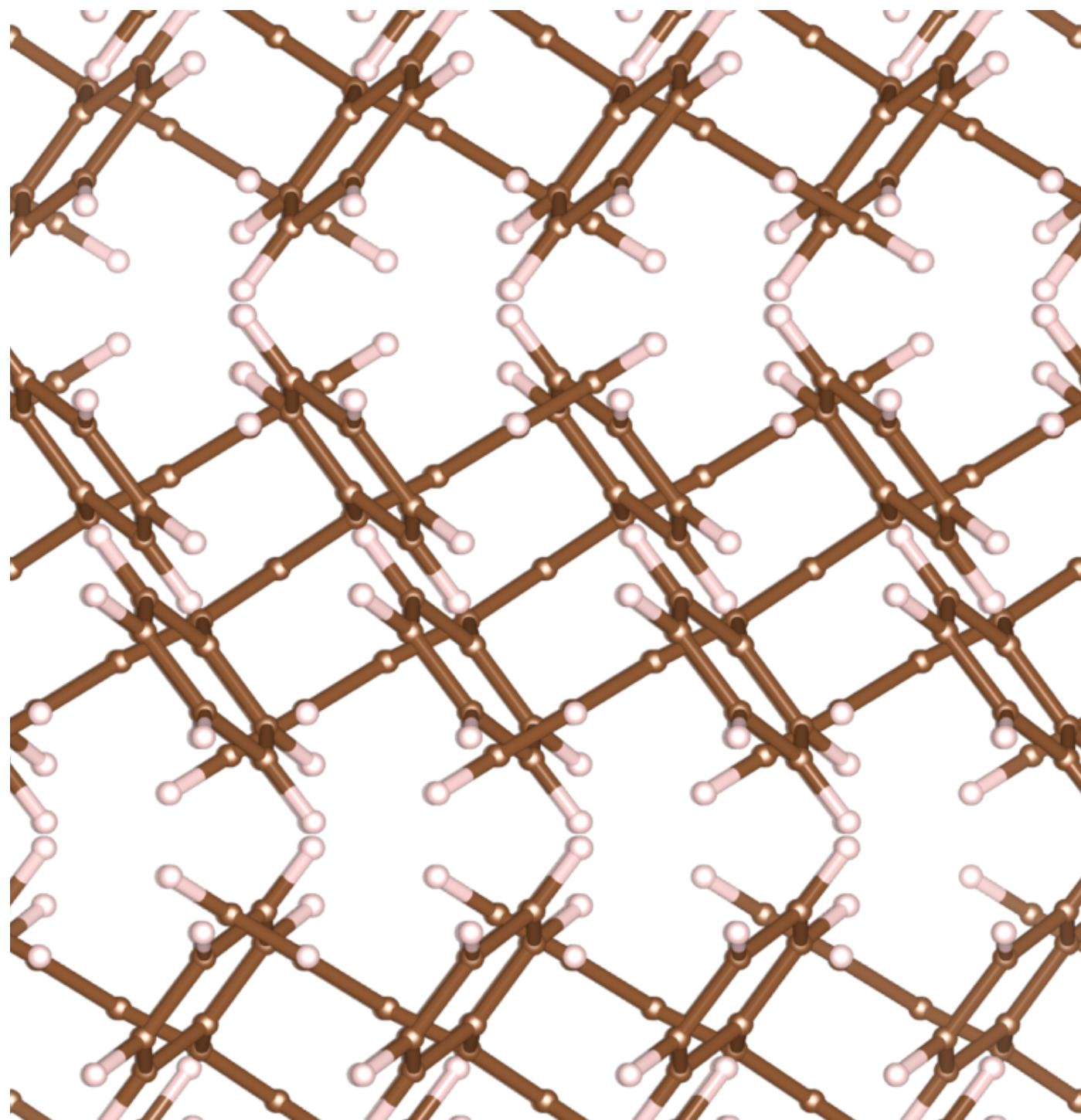
vdW-DF-cx (DF-cx): Brown-Altvater *et al.*, *Phys. Rev. B* **93**, 195206 (2016).

Expt. Natkaniec *et al.*, *J. Phys. C: Solid State Phys.* **13**, 4265 (1980).

Phonon dispersion of the organic semiconductor rubrene



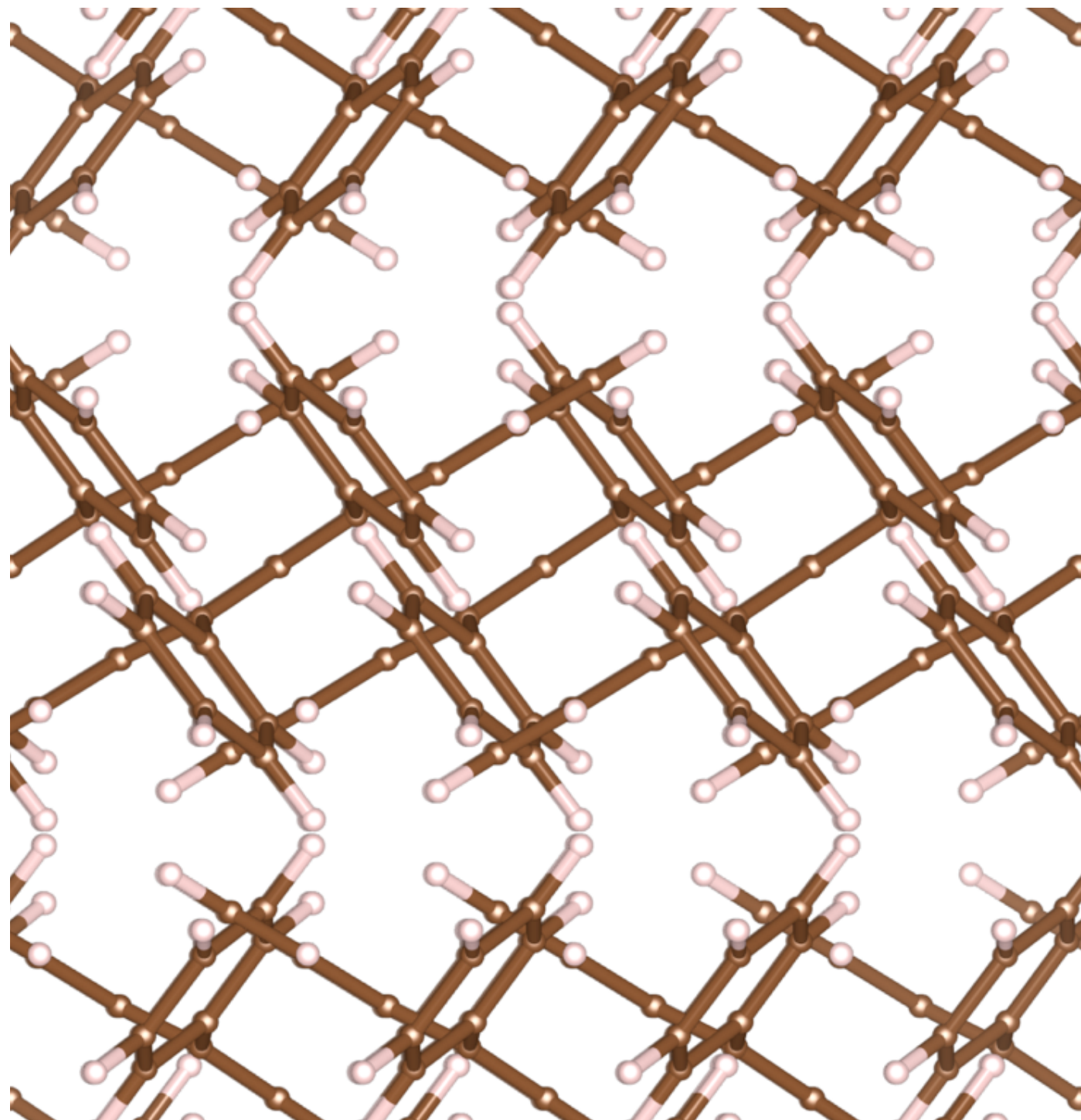
Crystal structure of rubrene



	a (Å)	b (Å)	c (Å)	V_0 (Å ³)	E_b (eV/mol)
PBE	27.989	7.804	16.585	1811.1	0.234
vdW-DF2	26.969	7.167	14.523	1403.5	2.319
rev-vdW-DF2	26.828	7.134	14.04	1354.3	2.18
Expt. (100K)	26.789	7.17	14.211	1364.8	

Expt. : O. D. Jurchescu, A. Meetsma, T. T. M. Palstra, *Acta Cryst. B* **62**, 330 (2006).

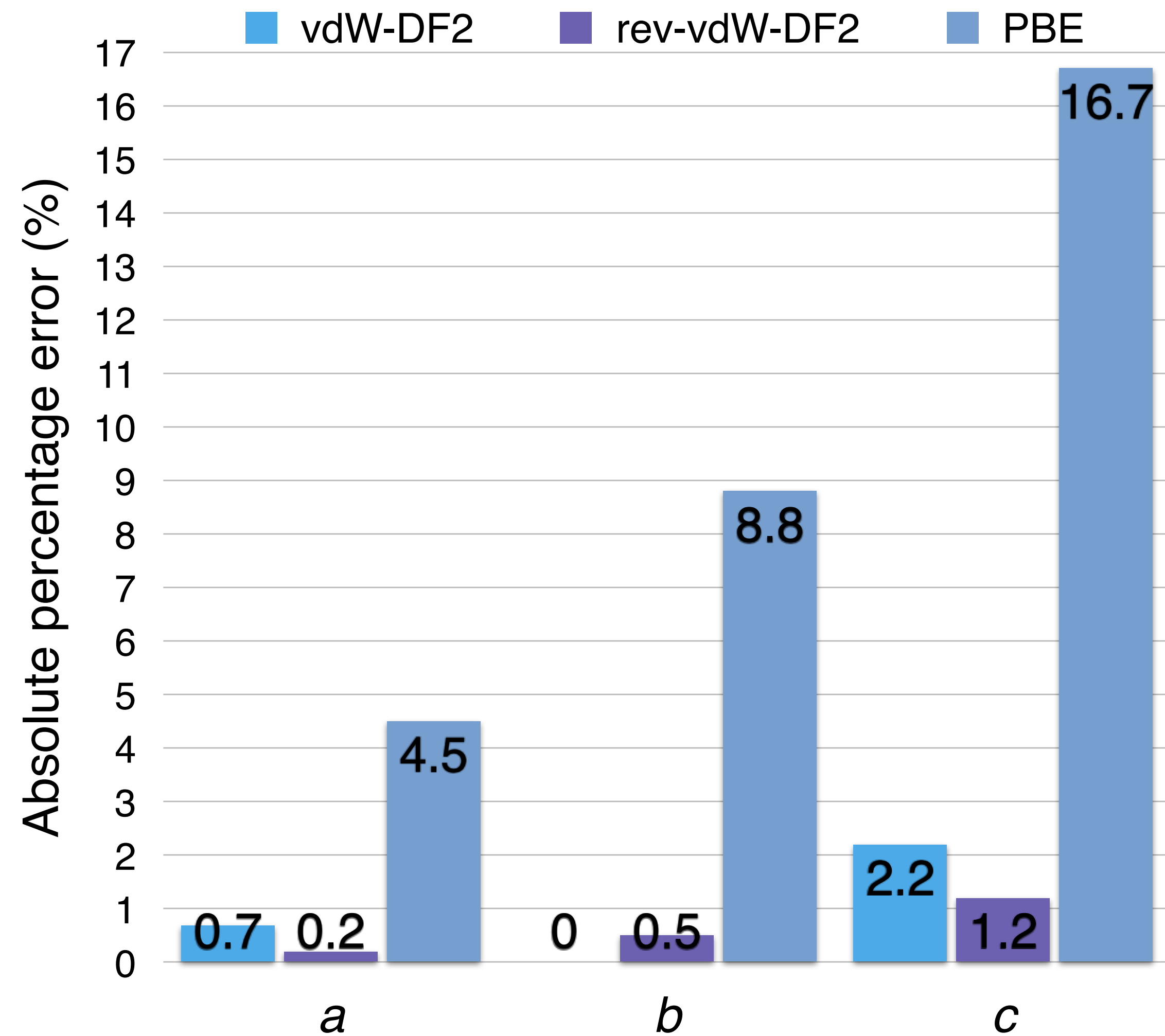
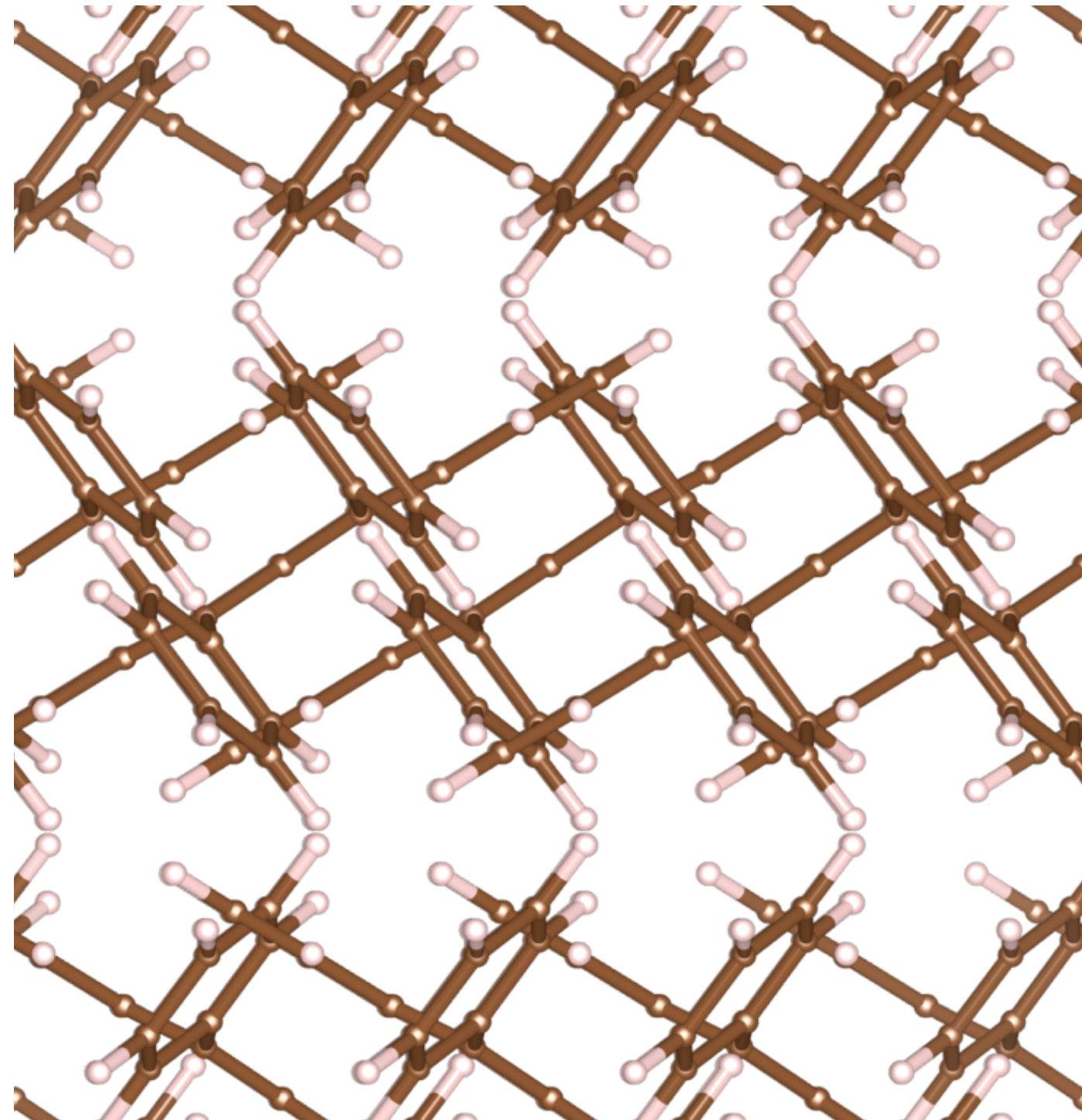
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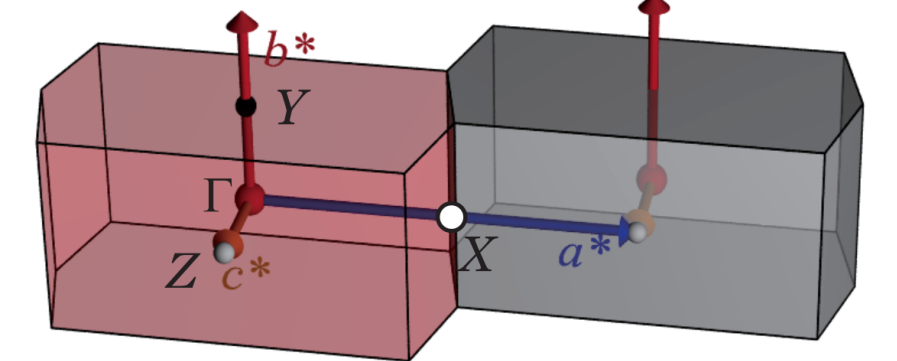
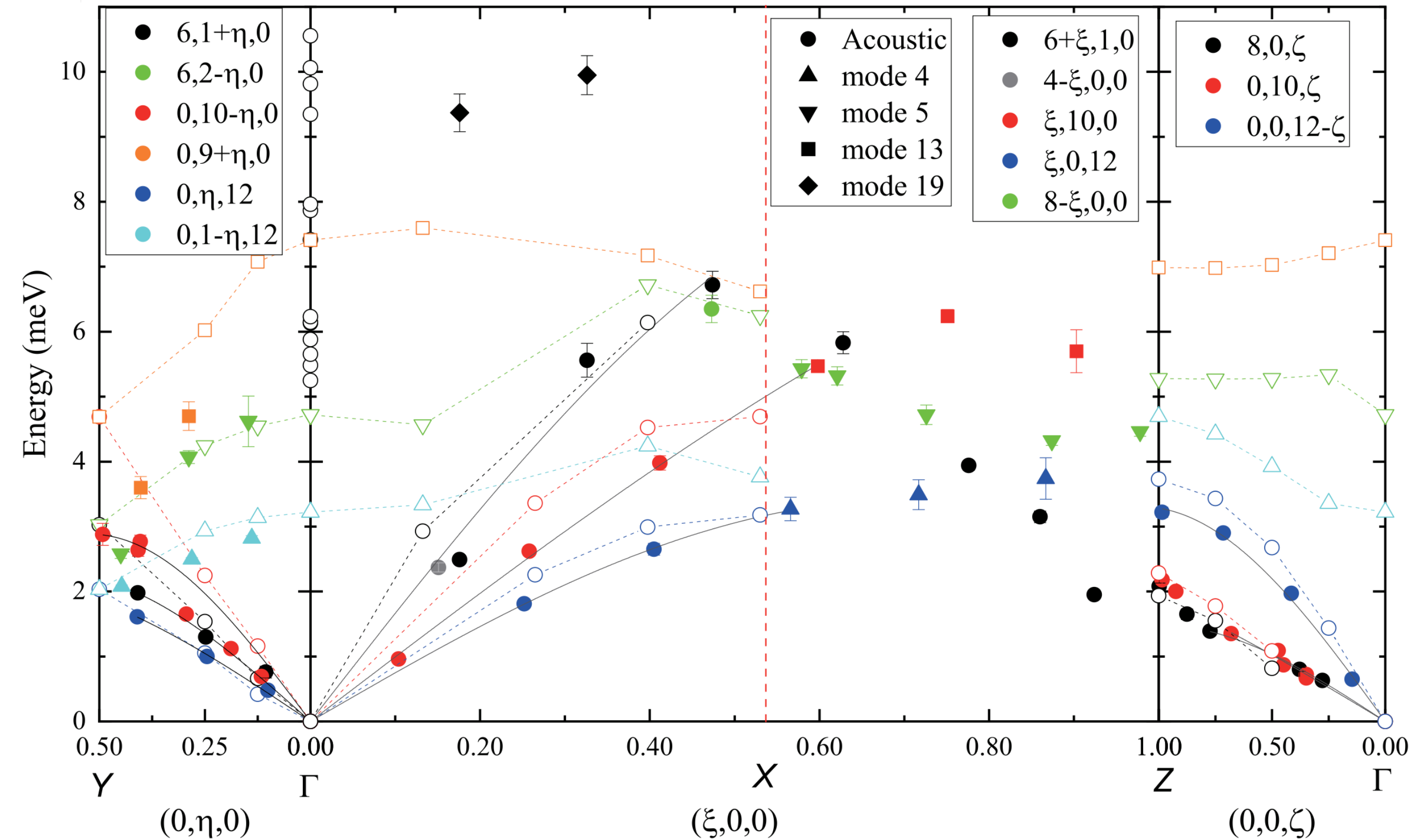
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vdW-DF2	26.969	7.167	14.523	1403.5	2.319
rev-vdW-DF2	26.828	7.134	14.04	1354.3	2.18
Expt. (100K)	26.789	7.17	14.211	1364.8	

Expt. : O. D. Jurchescu, A. Meetsma, T. T. M. Palstra, *Acta Cryst. B* **62**, 330 (2006).

Crystal structure of rubrene



Phonon band structure of rubrene



Filled circle: Expt.
Open circle: Calc.

Calculated phonon frequencies overestimated —temperature (unharmonic) effect?

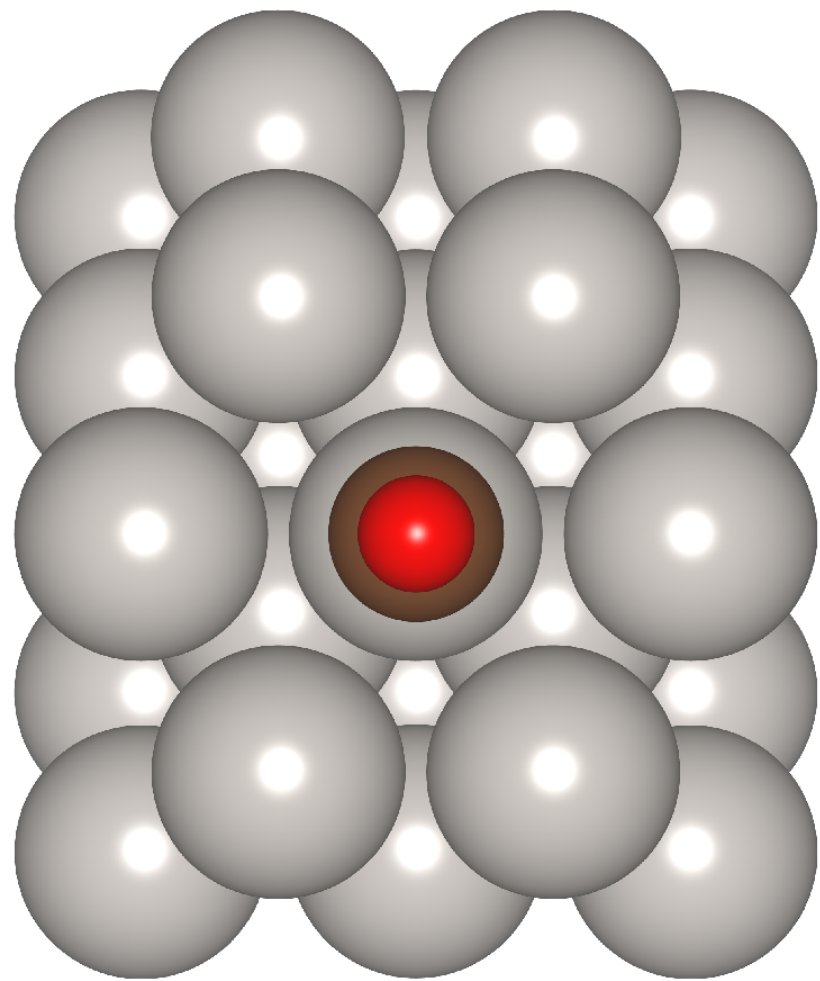
Outline

- Background: Dispersion forces in density functional theory
- Theory and application of van der Waals density functional (vdW-DF)
 - Benchmark for molecules, materials, and interfaces
 - Case study
 - Layered materials
 - Organic semiconductors
 - **Recent development**
- Summary

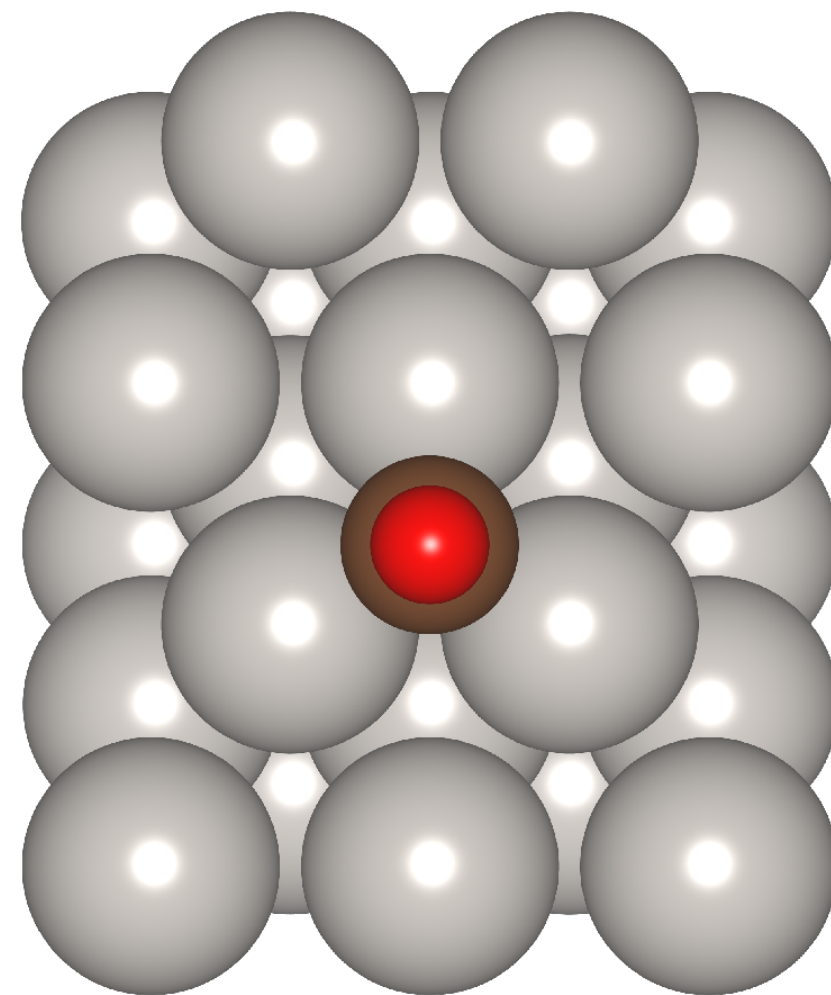
CO puzzle: CO adsorption on Pt(111)

$$\Delta E_{\text{site}} = E_{\text{ads}}^{\text{top}} - E_{\text{ads}}^{\text{fcc}}$$

Top



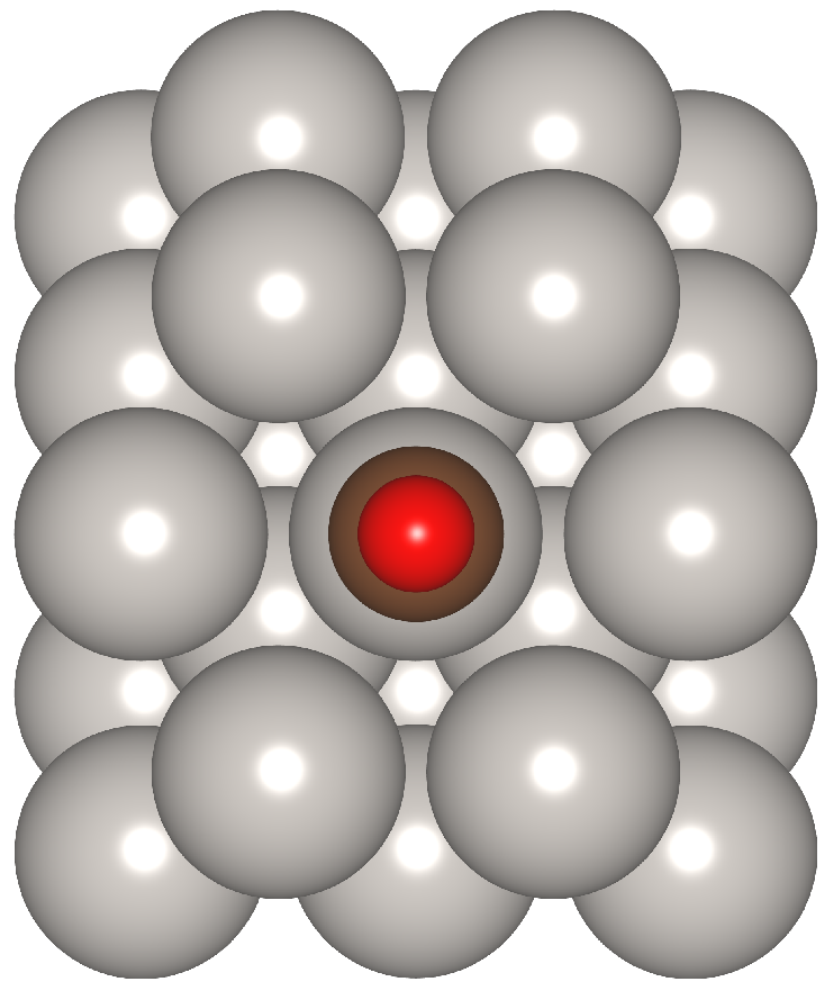
Fcc-hollow



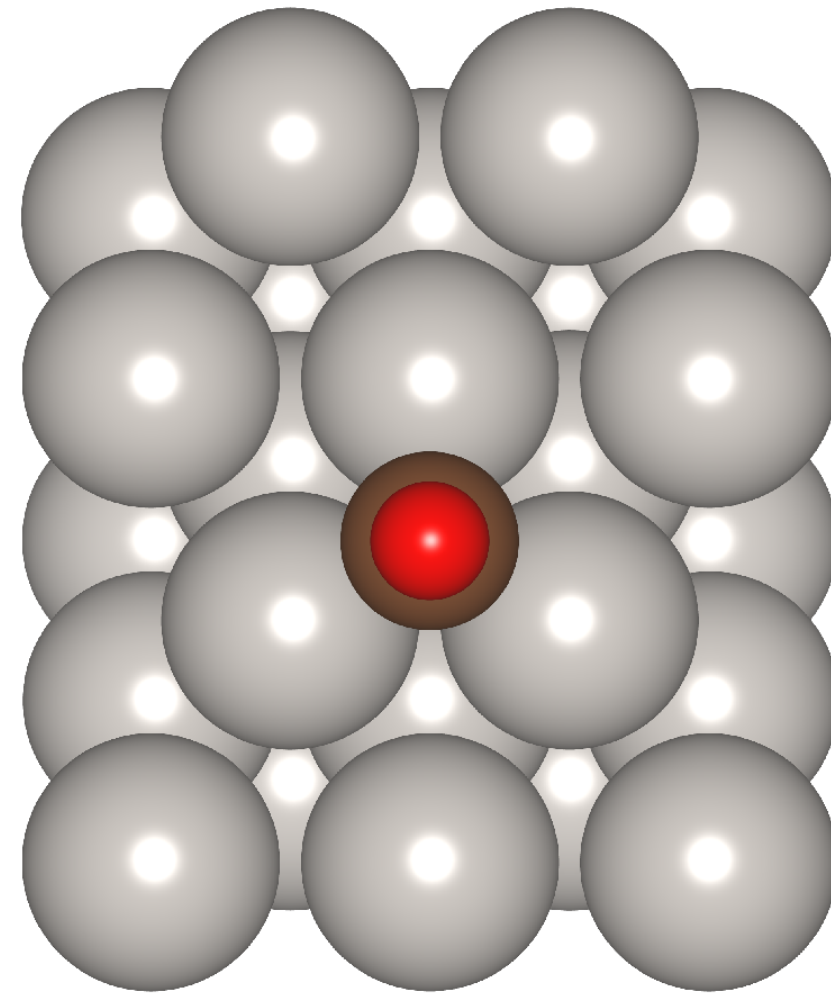
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Top



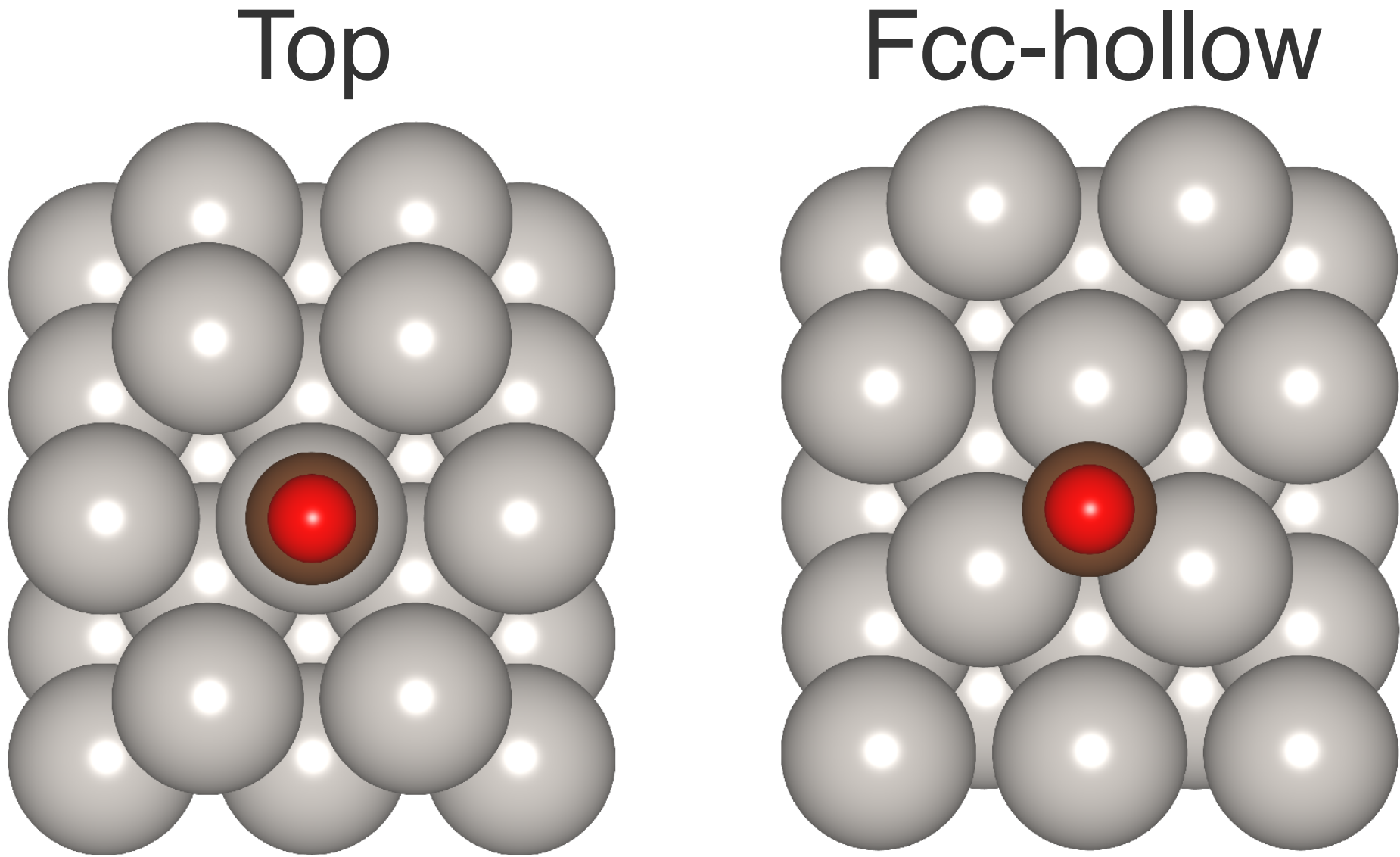
Fcc-hollow



Semilocal DFT (PBE, RPBE, ...) predicts wrong adsorption site (fcc)

CO puzzle: CO adsorption on Pt(111)

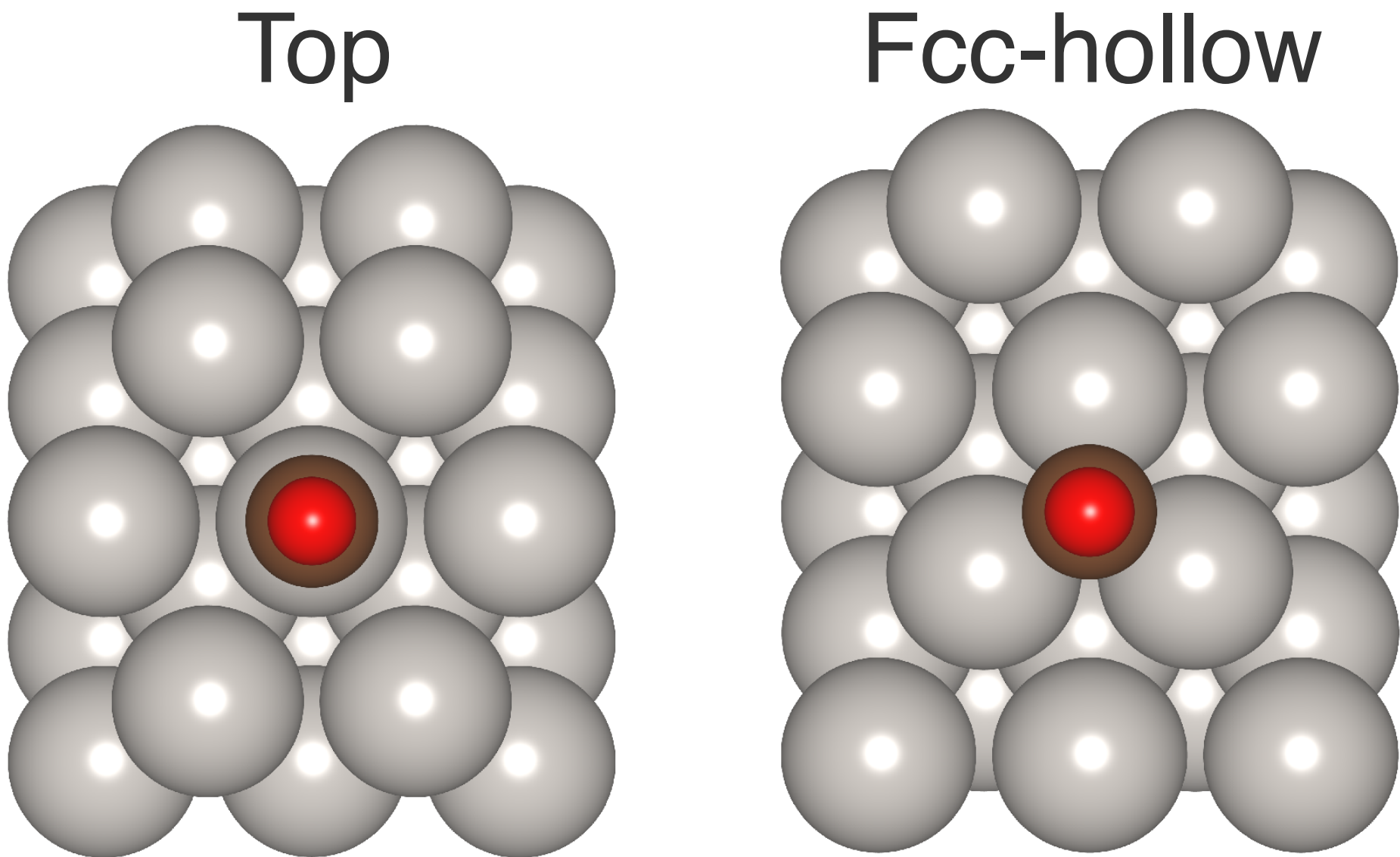
$$\Delta E_{\text{site}} = E_{\text{ads}}^{\text{top}} - E_{\text{ads}}^{\text{fcc}}$$



	vdW-DF2-B86R	Expt
Top	-1.683	-1.32
Fcc-hollow	-1.78	-

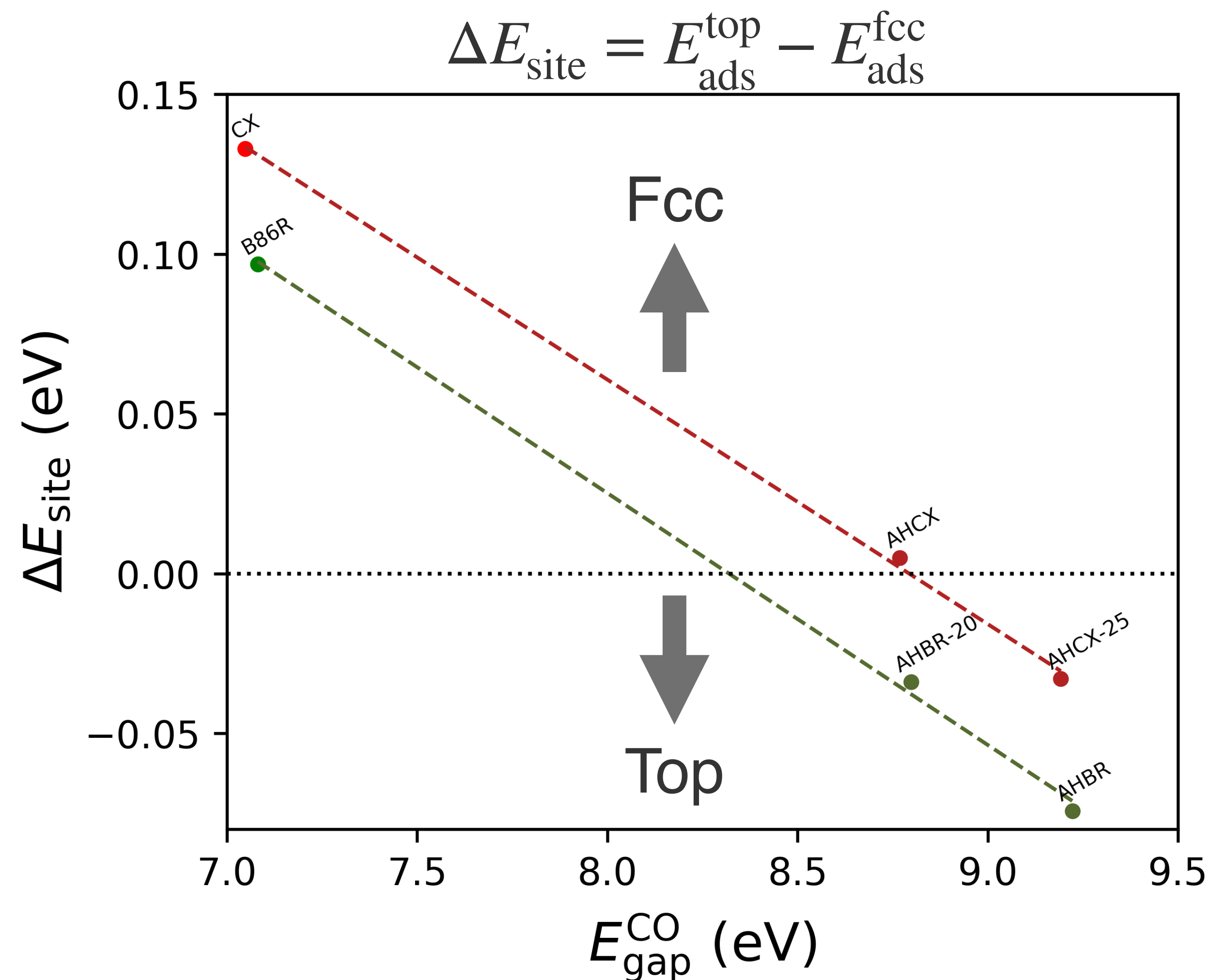
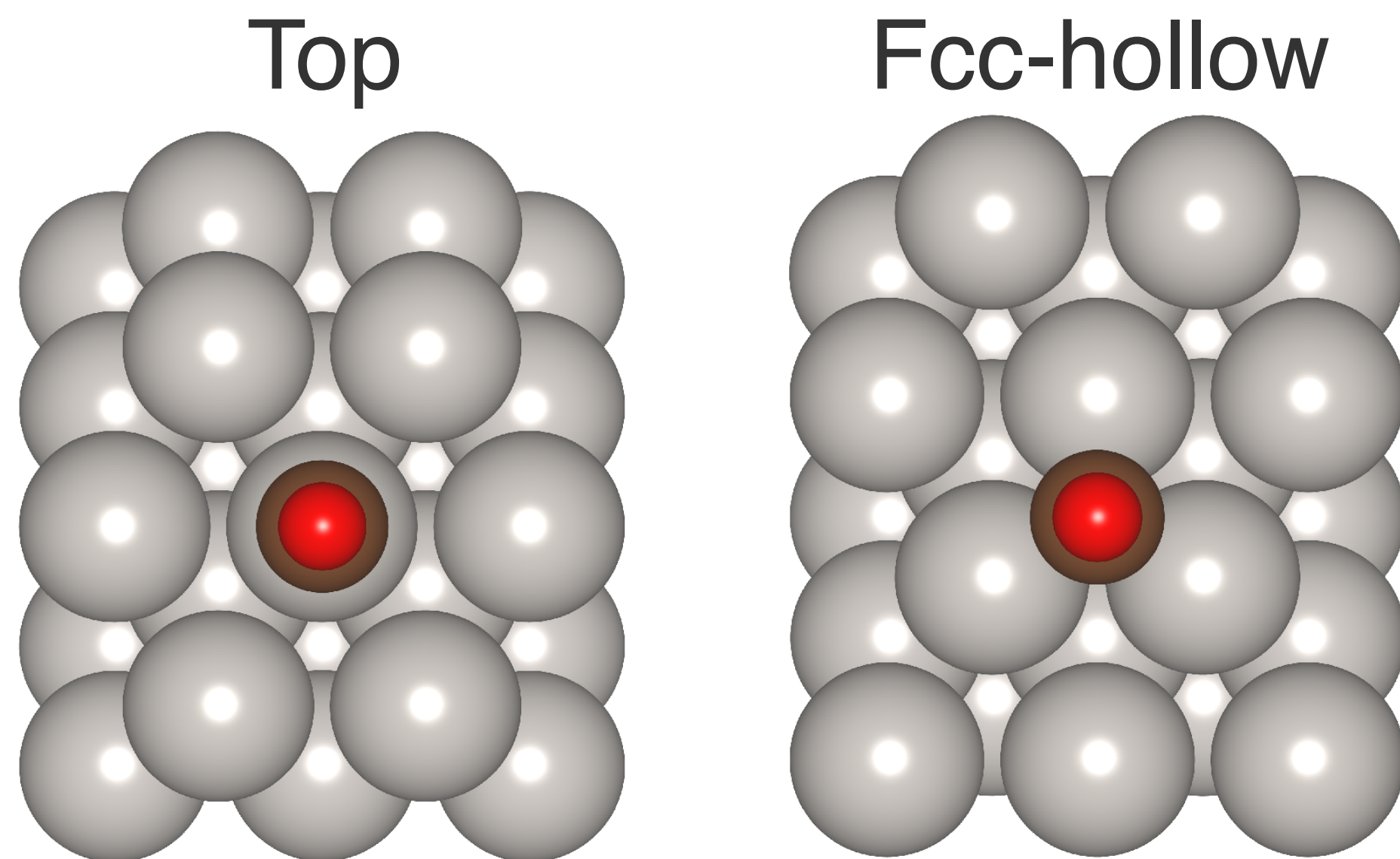
CO puzzle: CO adsorption on Pt(111)

$$\Delta E_{\text{site}} = E_{\text{ads}}^{\text{top}} - E_{\text{ads}}^{\text{fcc}}$$



	vdW-DF2-B86R	vdW-DF2-AHBR (hybrid vdW-DF)	Expt
Top	-1.683	-1.824	-1.32
Fcc-hollow	-1.78	-1.752	-

CO puzzle: CO adsorption on Pt(111)



vdW-DF2-B86R predicts the wrong adsorption site (fcc),
but (range-separate) hybrid vdW-DF2-AHBR corrects it (top)

Summary

- rev-vdW-DF2 / vdW-DF2-B86R
 - Accurate for energetics, geometries, and dynamics of variety systems
 - Accurate as PBE / PBEsol, but with the dispersion force included
 - Less accurate for band / HOMO-LUMO gap and related problem
- Further improvement is being underway, e.g., hybrid vdW-DF2-AHBR, but (periodic) quantum chemistry and/or many-body methods are highly desirable

References

- vdW-DF
 - Dion *et al.* *Phys. Rev. Lett.* **92**, 246401 (2004).
 - Lee *et al.*, *Phys. Rev. B* **82**, 081101 (2010).
 - Berland *et al.*, *Rep. Prog. Phys.* **78**, 066501 (2015).
 - Hamada, Activity Report 2021, Supercomputer Center, ISSP, Univ. of Tokyo.
https://mdcl.issp.u-tokyo.ac.jp/scc/wp/wp-content/uploads/2022/08/2021_0301_invited1.pdf
- rev-vdW-DF2
 - Hamada, *Phys. Rev. B* **89**, 121103 (2014).
 - Callsen and Hamada, *Phys. Rev. B* **91**, 195103 (2015).

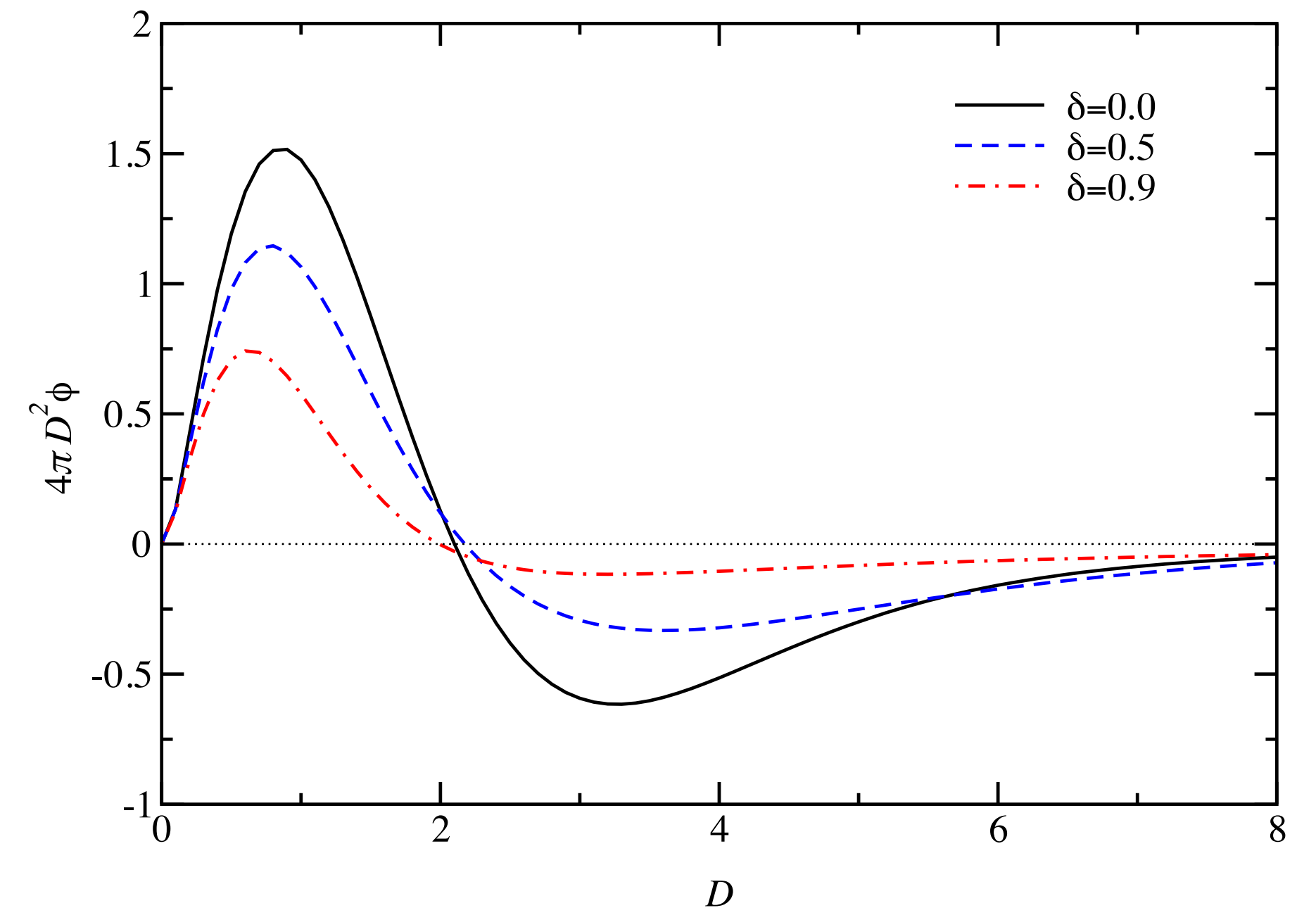
Backup

Efficient implementation of vdW-DF

Nonlocal correlation

$$E_c^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$



Román-Pérez and Soler, *Phys. Rev. Lett.* **103**, 096102 (2009).

Wu and Gygi, *J. Chem. Phys.* **136**, 224107 (2012).

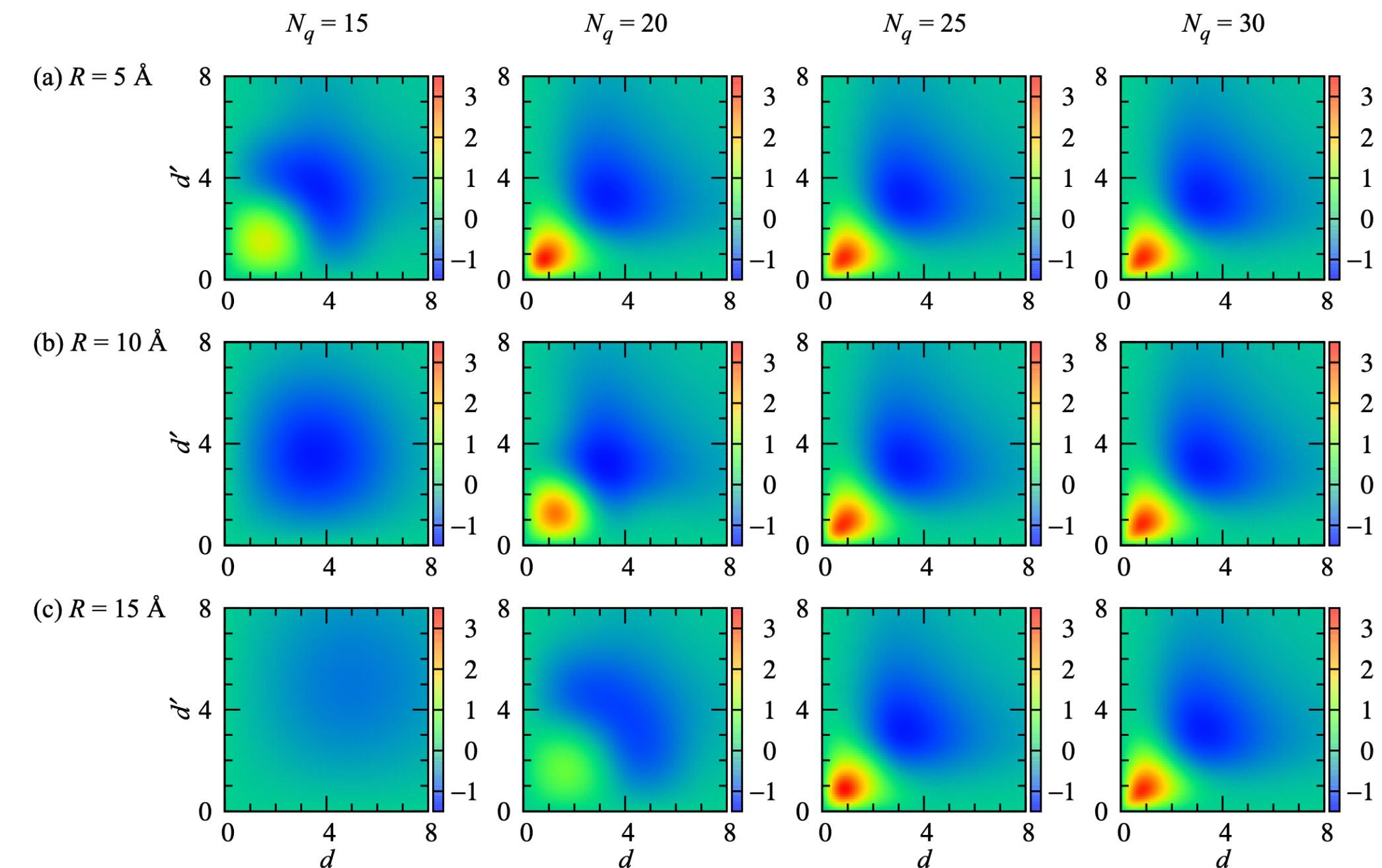
Hamamoto *et al.*, *Phys. Rev. B* **93**, 245440 (2016).

Efficient implementation of vdW-DF

Nonlocal correlation

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$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$



$$d = q(\mathbf{r}) |\mathbf{r} - \mathbf{r}'|, d' = q(\mathbf{r}') |\mathbf{r} - \mathbf{r}'|$$

Román-Pérez and Soler, *Phys. Rev. Lett.* **103**, 096102 (2009).

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Efficient implementation of vdW-DF

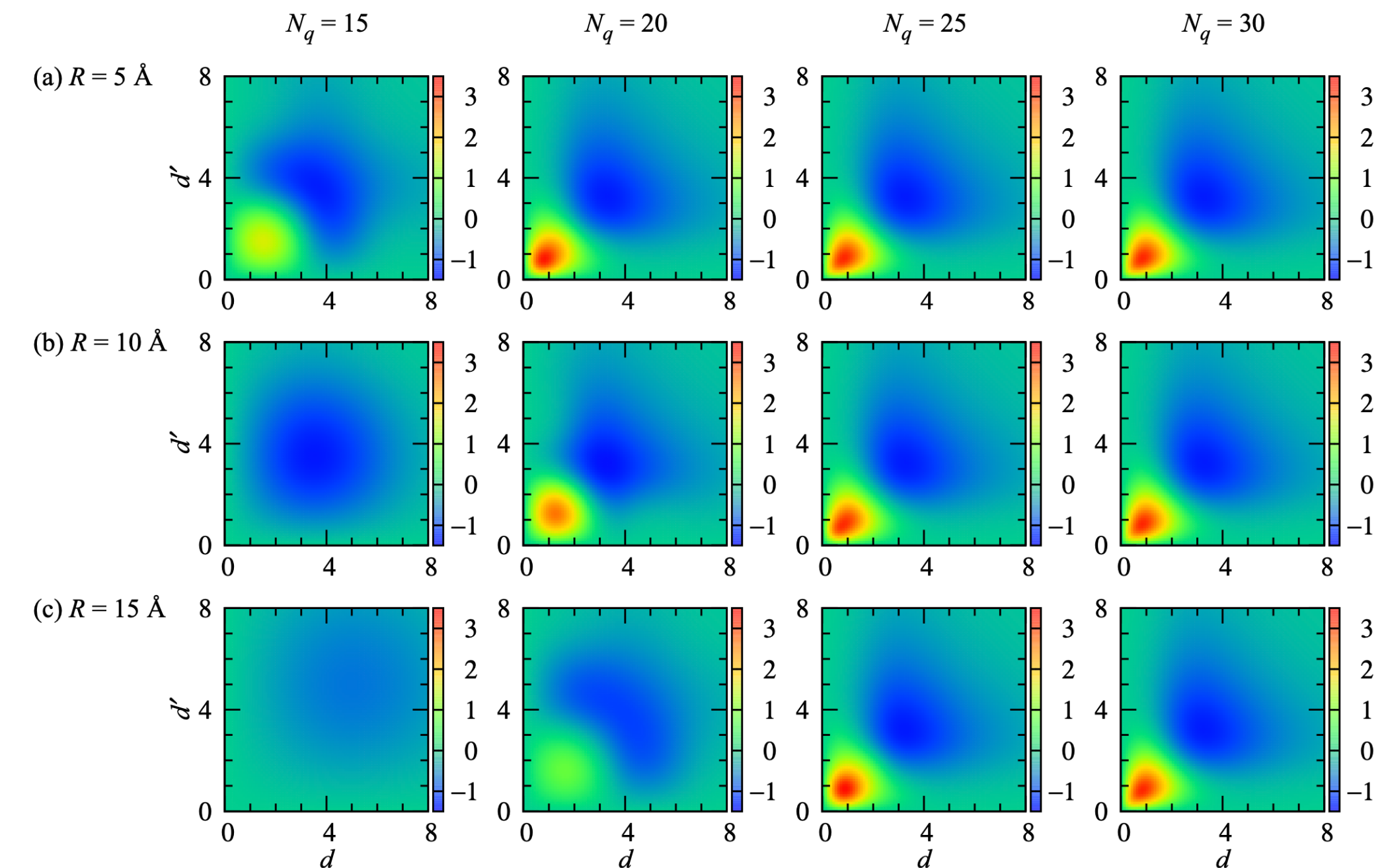
Nonlocal correlation

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$$q_i = q_0(n(\mathbf{r}_i), |\nabla n(\mathbf{r}_i)|)$$

$$\phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|)$$

$$\approx \sum_{\alpha, \beta} p_{\alpha}(q_1) \phi_{\alpha\beta}(|r_1 - r_2|) p_{\beta}(q_2)$$



$$d = q(\mathbf{r}) |\mathbf{r} - \mathbf{r}'|, d' = q(\mathbf{r}') |\mathbf{r} - \mathbf{r}'|$$

Román-Pérez and Soler, *Phys. Rev. Lett.* **103**, 096102 (2009).

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Efficient implementation of vdW-DF

Nonlocal correlation

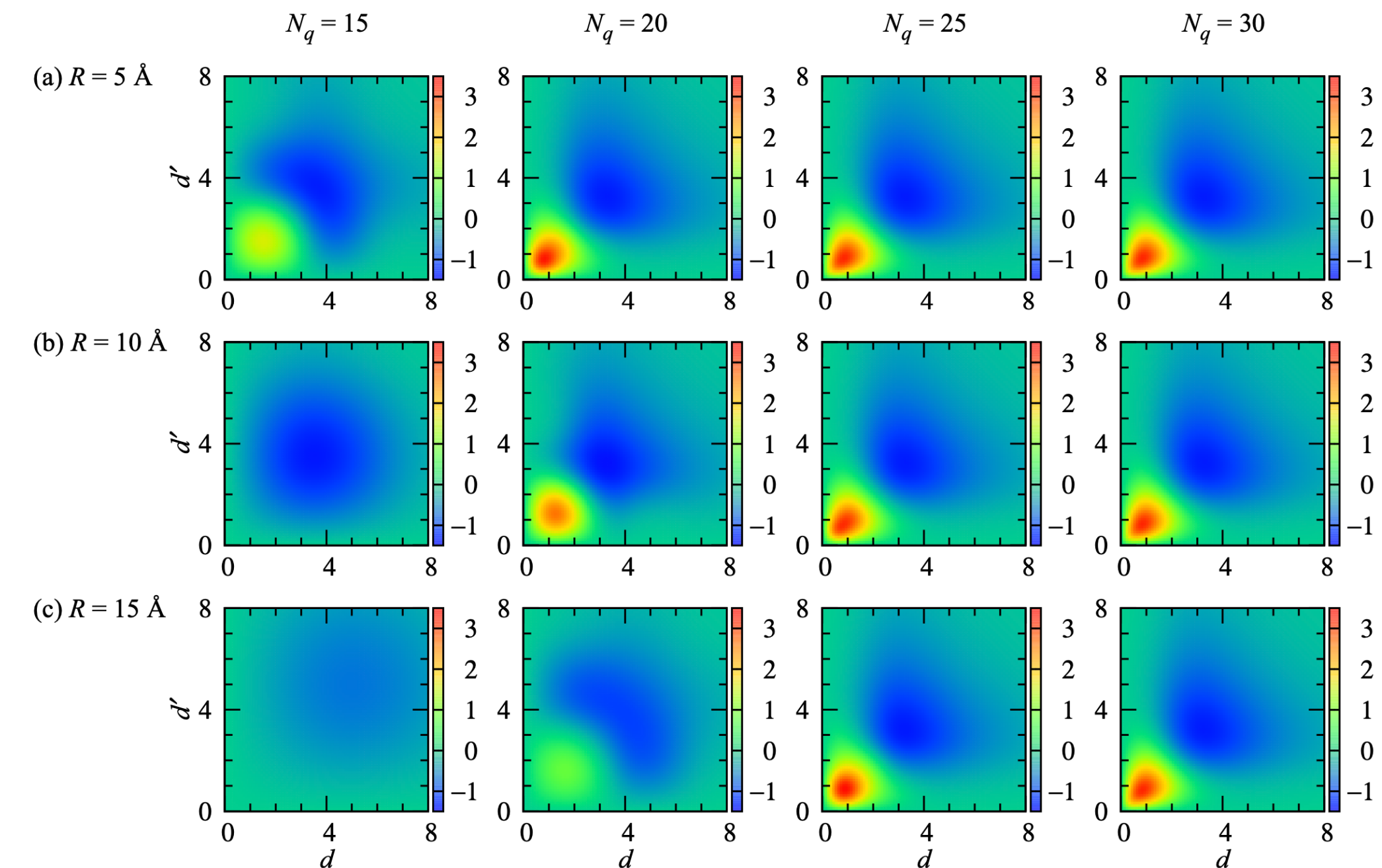
$$E_c^{\text{nl}} = \frac{1}{2} \iint d\mathbf{r}_1 d\mathbf{r}_2 n(\mathbf{r}_1) \phi(q_1 |r_1 - r_2|, q_2 |r_1 - r_2|) n(\mathbf{r}_2)$$

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$$\approx \sum_{\alpha, \beta} p_{\alpha}(q_1) \phi_{\alpha\beta}(|r_1 - r_2|) p_{\beta}(q_2)$$

$$E_c^{\text{nl}} = \frac{1}{2} \sum_{\alpha\beta} \int d\mathbf{k} \theta_{\alpha}^*(\mathbf{k}) \phi_{\alpha\beta}(k) \theta_{\beta}(\mathbf{k})$$



$$d = q(\mathbf{r}) |\mathbf{r} - \mathbf{r}'|, d' = q(\mathbf{r}') |\mathbf{r} - \mathbf{r}'|$$

Román-Pérez and Soler, *Phys. Rev. Lett.* **103**, 096102 (2009).

Wu and Gygi, *J. Chem. Phys.* **136**, 224107 (2012).

Hamamoto *et al.*, *Phys. Rev. B* **93**, 245440 (2016).

Graphite

	In-plane lattice constant (Å)	Interlayer distance (Å)	Binding energy (meV)
PBE	2.47	4.38	1
optB88-vdW	2.46	3.35	52
vdW-DF2	2.48	3.52	70
vdW-DF2-C09	2.48	3.27	56
rev-vdW-DF2	2.47	3.32	60
Expt.	2.4589	3.34	50±5